

# An Introduction to the Theory and Practice of LC Method Translations and LC Method Transfers

Gemma Lo



## Method Translations: What Are You Trying to Achieve?

- ◆ **Faster separations with the same performance?**
  - ◆ eg HPLC → UHPLC methods for increased productivity / sample throughput with similar efficiency, selectivity and resolution?
- ◆ **Converting UHPLC → HPLC methods?**
  - ◆ eg for offshoring or third party labs or manufacturing?
- ◆ **Porous particle → solid core particle method change?**
  - ◆ Take advantage of solid core efficiency, speed or low backpressure?
- ◆ **Faster method development?**
  - ◆ eg reduce overall cycle time?
- ◆ **Higher resolution / peak capacity?**
  - ◆ eg for related substances or complex samples?
- ◆ **Same methods...transferred between instruments?**

## Isocratic Translations

## Isocratic Method Translations: General Principles

- **Maintain a constant length to particle size ratio, L/d<sub>p</sub> (for the same phase type and phase vendor)**
- **Will give ~ similar performance i.e. efficiency (selectivity, resolution)\***
- **Thus, 300 x 3.9 mm, 10 µm ≈ 150 x 4.6 mm, 5 µm = 30,000**

	Column Length (mm)					
	50	75	100	150	250	300
Particle Size (µm)	1.7	29,412	44,118	58,824		
	1.8	27,778	41,667	55,556		
	1.9	26,316	39,474	52,632		
	2	25,000	37,500	50,000	75,000	
	2.5	20,000	30,000	40,000	60,000	100,000
	2.6	19,231	28,846	38,462	57,692	96,154
	2.7	18,519	27,778	37,037	55,556	92,593
	3	16,667	25,000	33,333	50,000	83,333
	5	10,000	15,000	20,000	30,000	50,000
	10	5,000	7,500	10,000	15,000	25,000
						30,000

 Columns meeting L/d<sub>p</sub> of 30,000

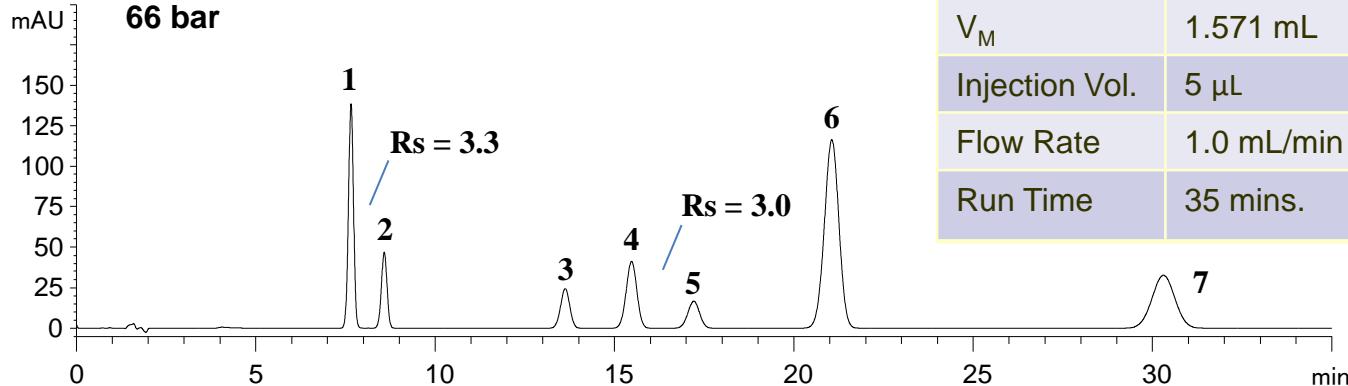
## Isocratic Method Translations: General Principles

- Once the new column format has been selected, isocratic translations are **fairly straightforward**.
  - Step 1: Scale injection volume ( $V_i$ ) to new column dead volume ( $V_M$ )
 
$$V_{i2} = \frac{V_{i1} \times V_{M2}}{V_{M1}}$$
  - Step 2: Scale flow rate (F)
    - If no change in  $d_p$ 

$$F_2 = \frac{F_1 \times d_{c2}^2}{d_{c1}^2}$$
    - If  $d_p$  changes can use
 
$$F_2 = \frac{F_1 \times d_{c2}^2 \times d_{p1}}{d_{c1}^2 \times d_{p2}}$$
  - Step 3: determine new run time
 
$$t_2 = \frac{t_1 F_1 V_{M2}}{F_2 V_{M1}}$$
  - Backpressure of new method can be estimated
 
$$P_2 = \frac{P_1 \times F_2 \times L_2 \times d_{c1}^2 \times d_{p1}^2}{F_1 \times L_1 \times d_{c2}^2 \times d_{p2}^2}$$

# Isocratic Method Translations: HPLC to UHPLC

**Methoxybenzene Structural Isomers**  
**ACE 5 C18-PFP 150 x 4.6 mm**  
**66 bar**



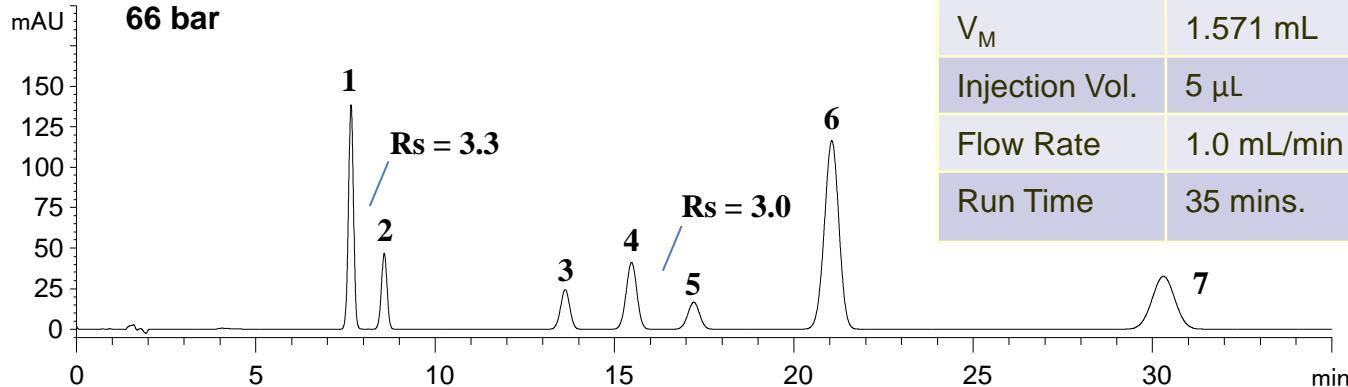
	HPLC Method	UHPLC Method
Column	150 x 4.6 mm, 5 $\mu$ m	50 x 3.0 mm, 1.7 $\mu$ m
$V_M$	1.571 mL	0.223 mL
Injection Vol.	5 $\mu$ L	
Flow Rate	1.0 mL/min	
Run Time	35 mins.	

- **ACE 5 C18-PFP 150 x 4.6 mm**  $V_M = 1.571 \text{ mL}$
- $L/d_P = 30,000$
- **ACE 1.7 C18-PFP 50 x 3.0 mm**  $V_M = 0.223 \text{ mL}$
- $L/d_P = 29,412$
- **$V_M$  experimentally determined**
- **Or use** 
$$V_M = \pi \left( \frac{d_c}{2} \right)^2 L \varepsilon$$
- **$\varepsilon$  = column porosity = ~0.63 for fully porous, ~0.55 for solid core**



## Isocratic Method Translations: HPLC to UHPLC

Methoxybenzene Structural Isomers  
ACE 5 C18-PFP 150 x 4.6 mm  
66 bar



	HPLC Method	UHPLC Method
Column	150 x 4.6 mm, 5µm	50 x 3.0 mm, 1.7µm
V <sub>M</sub>	1.571 mL	0.223 mL
Injection Vol.	5 µL	0.7 µL
Flow Rate	1.0 mL/min	<b>1.25 mL/min</b>
Run Time	35 mins.	

➤ Step 1: Scale **injection volume**

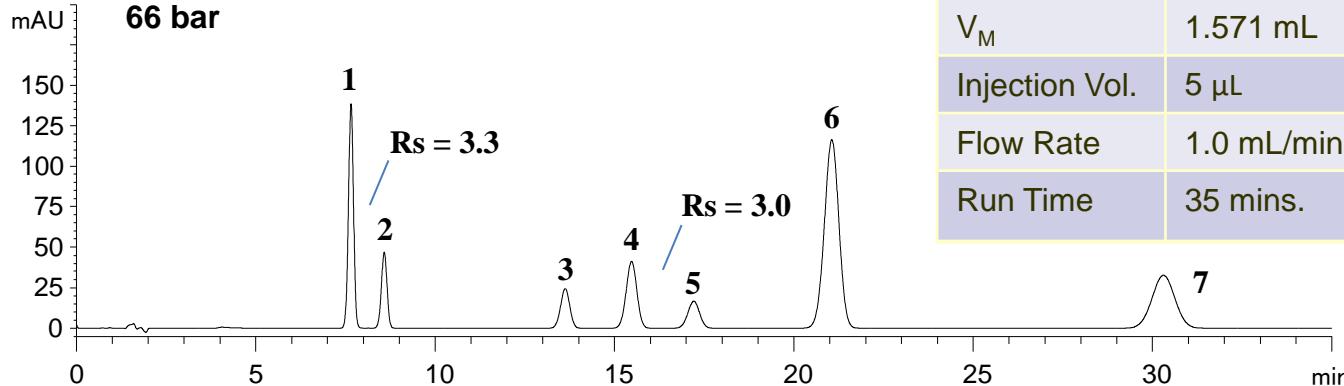
$$V_{i2} = \frac{V_{i1} \times V_{M2}}{V_{M1}} = \frac{5 \times 0.223}{1.571} = 0.7\mu L$$

➤ Step 2: Scale **flow rate**

$$F_2 = \frac{F_1 \times d_{c2}^2 \times d_{p1}}{d_{c1}^2 \times d_{p2}} = \frac{1.00 \times 3.0^2 \times 5}{4.6^2 \times 1.7} = 1.25 \text{ mL/min}$$

## Isocratic Method Translations: HPLC to UHPLC

**Methoxybenzene Structural Isomers**  
**ACE 5 C18-PFP 150 x 4.6 mm**  
**66 bar**



	HPLC Method	UHPLC Method
Column	150 x 4.6 mm, 5µm	50 x 3.0 mm, 1.7µm
V <sub>M</sub>	1.571 mL	0.223 mL
Injection Vol.	5 µL	0.7 µL
Flow Rate	1.0 mL/min	1.25 mL/min
Run Time	35 mins.	4 mins

➤ **Step 3: Calculate UHPLC run time**

$$t_2 = \frac{t_1 F_1 V_{M2}}{F_2 V_{M1}} = \frac{35 \times 1.00 \times 0.223}{1.25 \times 1.571} = 4 \text{ minutes}$$

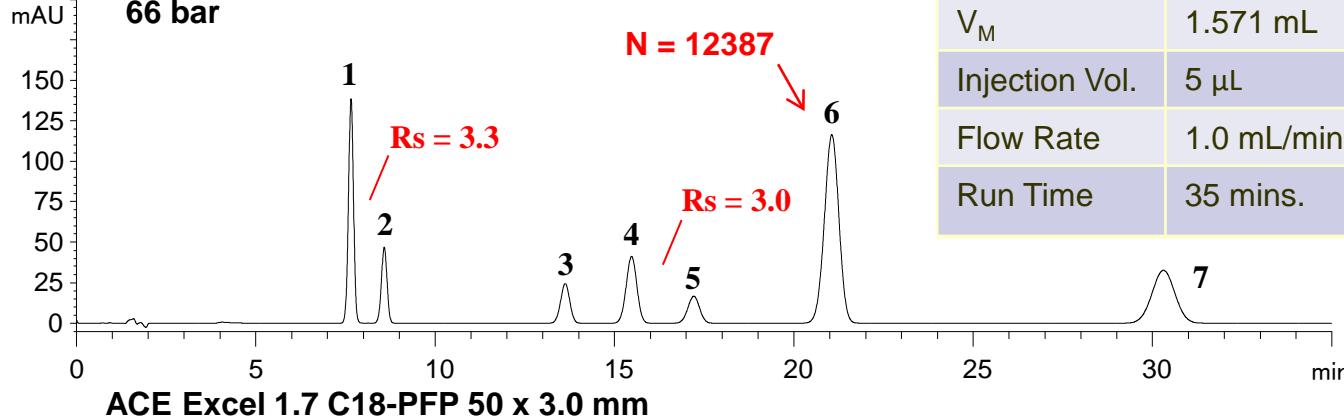
➤ **We can now estimate the backpressure of the translated method**

$$P_2 = \frac{P_1 \times F_2 \times L_2 \times d_{c1}^2 \times d_{p1}^2}{F_1 \times L_1 \times d_{c2}^2 \times d_{p2}^2} = \frac{66 \times 1.25 \times 50 \times 4.6^2 \times 5^2}{1.00 \times 150 \times 3^2 \times 1.7^2} = 559 \text{ bar}$$

# Isocratic Method Translations: HPLC to UHPLC

**Methoxybenzene Structural Isomers**  
**ACE 5 C18-PFP 150 x 4.6 mm**

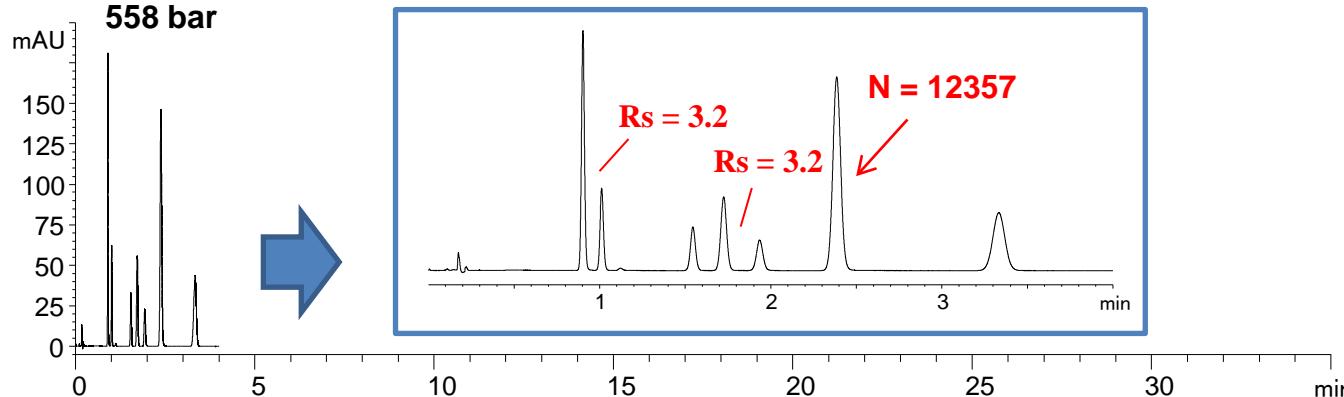
**66 bar**



	HPLC Method	UHPLC Method
Column	150 x 4.6 mm, 5 $\mu$ m	50 x 3.0 mm, 1.7 $\mu$ m
$V_M$	1.571 mL	0.223 mL
Injection Vol.	5 $\mu$ L	0.7 $\mu$ L
Flow Rate	1.0 mL/min	1.25 mL/min
Run Time	35 mins.	4 mins

**ACE Excel 1.7 C18-PFP 50 x 3.0 mm**

**558 bar**

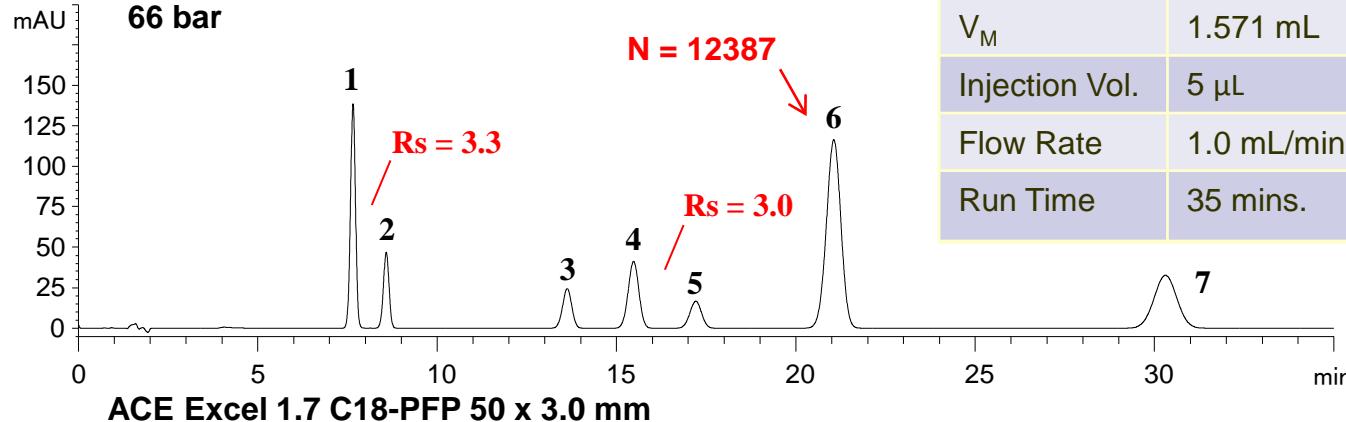


	Original method	Translated method	Difference
<b>Run time</b>	35 minutes	4 minutes	<b>-88.6%</b>
<b>Solvent consumption</b>	35.0 mL	5.0 mL	<b>-85.7%</b>



## Isocratic Method Translations: HPLC to UHPLC

Methoxybenzene Structural Isomers  
ACE 5 C18-PFP 150 x 4.6 mm  
66 bar



	HPLC Method	UHPLC Method
Column	150 x 4.6 mm, 5 $\mu$ m	50 x 3.0 mm, 1.7 $\mu$ m
$V_M$	1.571 mL	0.223 mL
Injection Vol.	5 $\mu$ L	0.7 $\mu$ L
Flow Rate	1.0 mL/min	1.25 mL/min
Run Time	35 mins.	4 mins

- Note: the same translation principles are also applicable to UHPLC to HPLC method translations

## ACE LC Translator

- ♦ All of the translation / transfer discussed in this talk have been conveniently combined in the ACE LC Translator

V1.3

### ACE LC Translator

This tool has been developed to aid the practicing chromatographer to efficiently translate and transfer LC methods between different format columns and different LC systems. Also included are a set of useful tools for the calculation of everyday chromatographic and method parameters.

The various tools can be accessed via the links below.

**Tools**

[Method Translation](#) [Method Transfer](#) [Dwell Volume Calculator](#)

[Column Porosity Calculator](#)  
[Extra Column Volume Calculator](#)  
[Column Equilibration Calculator](#)  
[Buffer Calculator](#)  
[Mobile Phase Consumption Calculator](#)

**Free Literature**

[ACE Application Notes](#) [ACE Product Portfolio](#)  
[Contact Us / Feedback](#)

[\\*Register for Future Updates\\* - register to receive future updates to the ACE LC Translator](#)

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[Disclaimer](#)

## ACE LC Translator

- ♦ All of the translation / transfer discussed in this talk have been conveniently combined in the **ACE LC Translator**
- ♦ Download free at:  
[mac-mod.com](http://mac-mod.com)

V1.3

### Method Translation

Methods are frequently translated from one column format to another and among different brands and models of instruments (different system volumes, dwell volumes etc). In order to maintain chromatographic and method performance through the translation process, a number of method parameter changes are required, such as flow rate, injection volume and  $t_c$ . Separate tools for isocratic and gradient methods are included on this page. For details on how to determine system dwell volumes, please refer to the Dwell Volume tab.

<b>Isocratic</b>	
<b>Column Information</b> <b>Current</b> Column Length (L) <input type="text" value="150"/> Column i.d. (d <sub>i</sub> ) <input type="text" value="4.6"/> Particle Diameter (d <sub>p</sub> ) <input type="text" value="5.0"/> L/d <sub>p</sub> <input type="text" value="3000"/> Column Porosity <input type="text" value="0.63"/> Column Volume (V <sub>H</sub> ) <input type="text" value="1.570"/>  <b>Method</b> <b>Current</b> Injection Volume <input type="text" value="10.0"/> Flow Rate <input type="text" value="1.00"/>  Run Time <input type="text" value="25.0"/> Recorded Backpressure <input type="text" value="60"/> Solvent Use <input type="text" value="25"/>	<b>Translated</b> Column Length (L) <input type="text" value="50"/> Column i.d. (d <sub>i</sub> ) <input type="text" value="3.0"/> Particle Diameter (d <sub>p</sub> ) <input type="text" value="17"/> L/d <sub>p</sub> <input type="text" value="2942"/> Column Porosity <input type="text" value="0.63"/> Column Volume (V <sub>H</sub> ) <input type="text" value="0.223"/>  <b>Translated</b> Injection Volume <input type="text" value="1.4"/> Flow Rate (scaled to linear velocity) <input type="text" value="0.43"/> Flow Rate (scaled to particle size) <input type="text" value="1.25"/> Input Flow Rate <input type="text" value="1.25"/>  New Run Time <input type="text" value="2.8"/> Estimated Backpressure <input type="text" value="508"/> Estimated Solvent Use Difference <input type="text" value="-86 %"/>

[Main Menu](#)

<b>Gradient</b>	
<b>Column Information</b> <b>Current</b> Column Length (L) <input type="text" value="mm"/> Column i.d. (d <sub>i</sub> ) <input type="text" value="mm"/> Particle Diameter (d <sub>p</sub> ) <input type="text" value="μm"/> L/d <sub>p</sub> <input type="text" value=""/>	<b>Translated</b> Column Length (L) <input type="text" value="mm"/> Column i.d. (d <sub>i</sub> ) <input type="text" value="mm"/> Particle Diameter (d <sub>p</sub> ) <input type="text" value="μm"/> L/d <sub>p</sub> <input type="text" value=""/>
<b>Method</b> <b>Current</b> Injection Volume <input type="text" value="μL"/> Flow Rate <input type="text" value="mL/min"/>  LC Name <input type="text" value=""/> Dwell Volume (V <sub>D</sub> ) <input type="text" value="mL"/>	<b>Translated</b> Injection Volume <input type="text" value="μL"/> Flow Rate (scaled to linear velocity) <input type="text" value="mL/min"/> Flow Rate (scaled to particle size) <input type="text" value="mL/min"/> Input Flow Rate <input type="text" value="mL/min"/>  LC Name <input type="text" value=""/> Dwell Volume (V <sub>D</sub> ) <input type="text" value="mL"/>

# Method Translation Tool Example - Isocratic

**V1.3**

## Method Translation

Methods are frequently translated from one column format to another and among different brands and models of instruments (different system volumes, dwell volumes etc). In order to maintain chromatographic and method performance through the translation process, a number of method parameter changes are required, such as flow rate, injection volume and  $t_g$ . Separate tools for isocratic and gradient methods are included on this page. For details on how to determine system dwell volumes, please refer to the Dwell Volume tab.

**Isocratic**

Column Information		Translated	
<b>Current</b>		<b>Translated</b>	
Column Length (L)	150:mm	Column Length (L)	50:mm
Column i.d. ( $d_c$ )	4.6:mm	Column i.d. ( $d_c$ )	3.0:mm
Particle Diameter ( $d_p$ )	5.0: $\mu$ m	Particle Diameter ( $d_p$ )	1.7: $\mu$ m
$L/d_p$	30000	$L/d_p$	29412
Column Porosity	0.63:What's This?	Column Porosity	0.63:What's This?
Column Volume ( $V_M$ )	1.570 mL	Column Volume ( $V_M$ )	0.223 mL
<b>Method</b>		<b>Translated</b>	
<b>Current</b>		<b>Translated</b>	
Injection Volume	10.0: $\mu$ L	Injection Volume	1.4 $\mu$ L
Flow Rate	1.00:mL/min	Flow Rate (scaled to linear velocity)	0.43 mL/min
Run Time	25.0:mins	Flow Rate (scaled to particle size)	1.25 mL/min
Recorded Backpressure	60:bar	Input Flow Rate	1.25 mL/min
Solvent Use	25 mL	New Run Time	2.8 mins
		Estimated Backpressure	508 bar
		Estimated Solvent Use Difference	-86 %

**Main Menu**

**Gradient**

**Fill in the grey input boxes to translate the method**

**Displays new run time and solvent saving**

**Optional input for flow rate – great for allowing greater analyst choice**

**Please complete all input fields\*\***



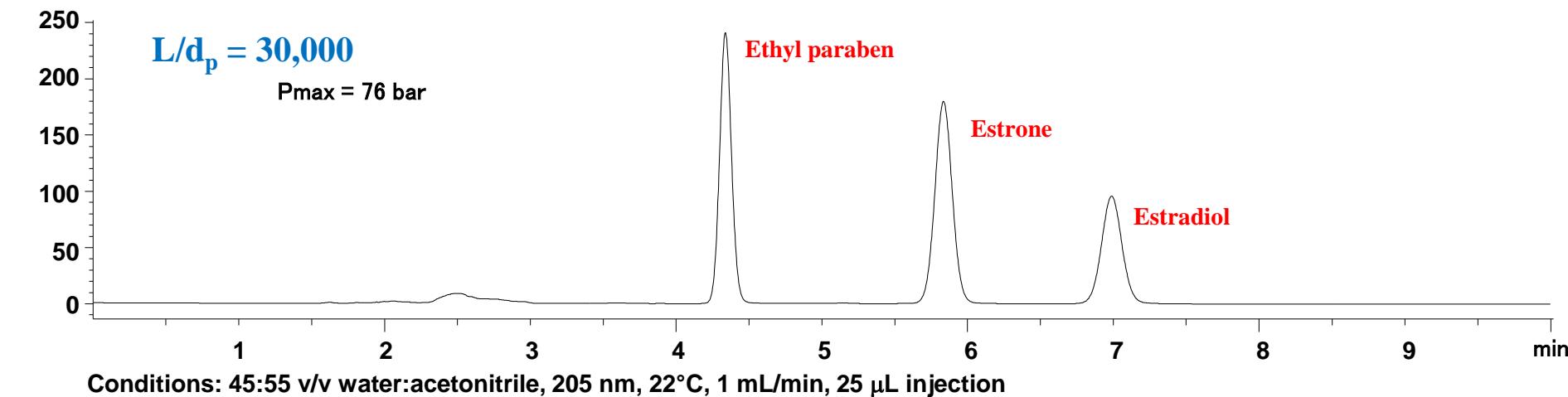
## Isocratic Method Translations: Pharmacopeial Methods

- Revised guidance on allowable parameter changes recently updated for **USP <621> chapter\***
  - Significant update to the general chapter – reading recommended
  - Isocratic monograph LC methods can be **-25% to +50% of the original L/d<sub>p</sub>** according to the USP guidance
  - Also, L/d<sub>p</sub> ratios outside this range can be **-25% to +50% of the original peak efficiency, N** ← large flexibility
- **Very few changes / flexibility for gradient monograph methods**
- When translating to smaller columns dimensions (and therefore smaller column volume), **system dispersion effects are important**



## Method Translations: Isocratic Estradiol Porous → Porous (I)

## Original Monograph Separation: ACE 10µm, L1 (C18), 300 x 3.9 mm



1. Find suitable ‘new’ column: 300x3.9mm, 10µm = L/d<sub>p</sub> of 30,000 = 150x4.6, 5µm

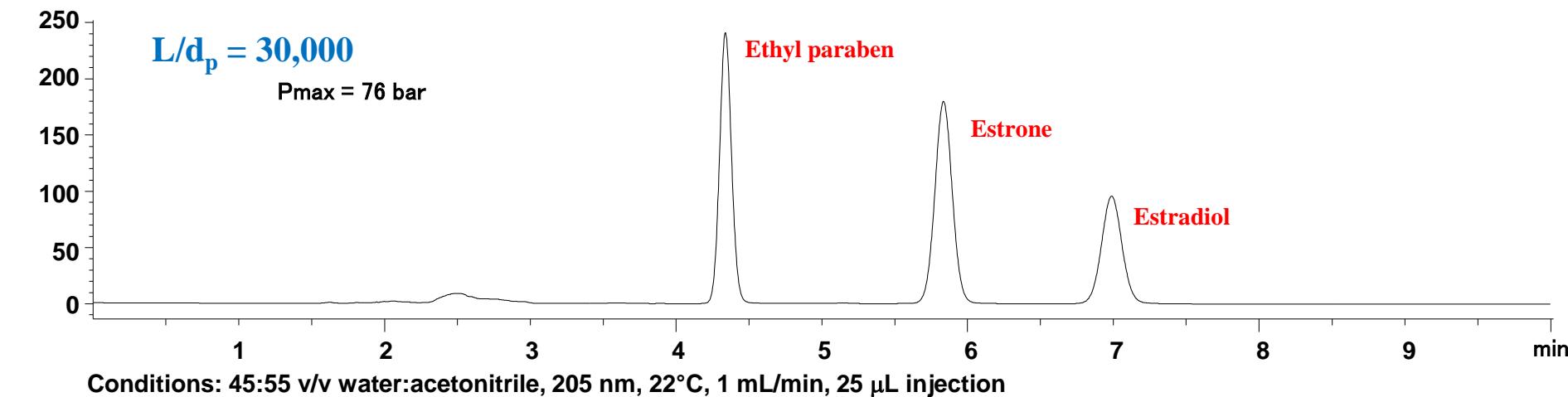
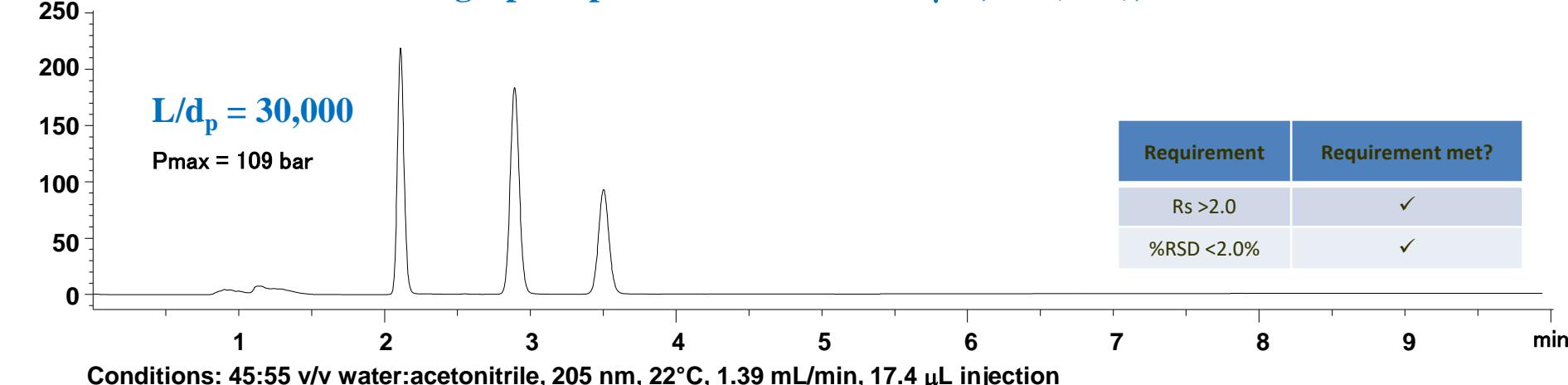
2. Geometrically scale flow rate:  $F_2 = F_1 \times \frac{d_{c_2}^2}{d_{c_1}^2} = 1 \times 4.6^2 / 3.9^2 = \underline{1.39 \text{ mL/min}}$

3. Volumetrically scale injection:  $Inj_2 = Inj_1 \times \frac{V_{m_2}}{V_{m_1}} = 25 \times 1.620 / 2.329 = \underline{17.4 \mu\text{L}}$

$$V_M \approx \pi \left(\frac{d}{2}\right)^2 L \varepsilon$$



## Method Translations: Isocratic Estradiol Porous → Porous (II)

Original Monograph Separation: ACE 10 $\mu$ m, L1 (C18), 300 x 3.9 mmFirst Translated Monograph Separation: ACE Excel 5 $\mu$ m, L1 (C18), 150 x 4.6 mm



## Method Translations: Isocratic Methods

- Isocratic monograph LC methods can be **-25% to +50% of the original L/d<sub>p</sub>** according to the USP guidance\*

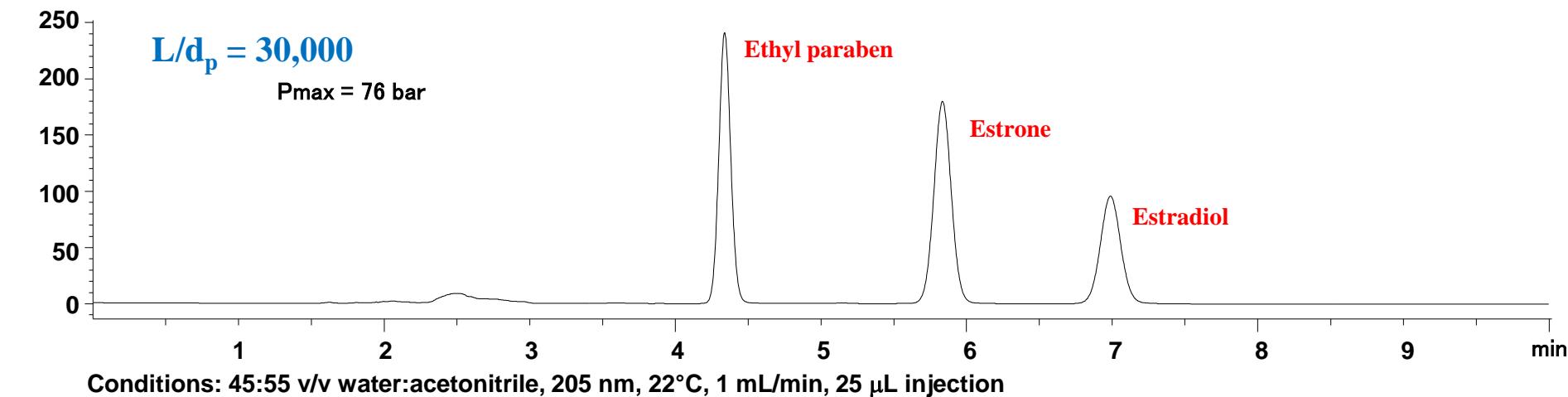
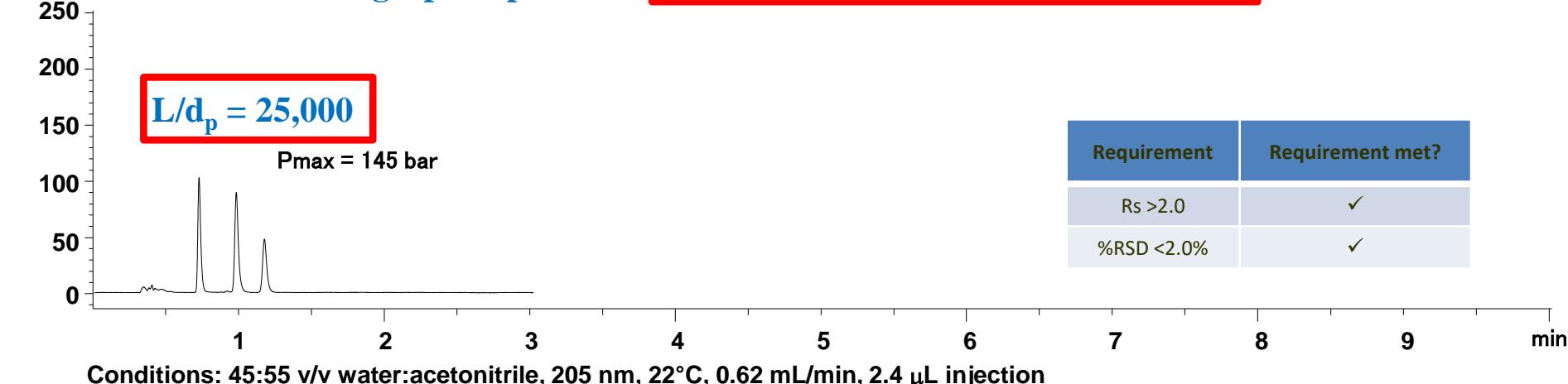
Column Length (mm)	300
Particle Size (μm)	10
Lower L/d <sub>p</sub> (-25%)	22,500
L/d <sub>p</sub>	30,000
Upper L/d <sub>p</sub> (+50%)	45,000

	Column Length (mm)					
	50	75	100	150	250	300
1.7	29,412	44,118	58,824			
1.8	27,778	41,667	55,556			
1.9	26,316	39,474	52,632			
2	25,000	37,500	50,000	75,000		
2.5	20,000	30,000	40,000	60,000	100,000	
2.6	19,231	28,846	38,462	57,692	96,154	
2.7	18,519	27,778	37,037	55,556	92,593	
3	16,667	25,000	33,333	50,000	83,333	
5	10,000	15,000	20,000	30,000	50,000	
10	5,000	7,500	10,000	15,000	25,000	30,000

Columns meeting L/d<sub>p</sub> of 22,400 to 45,000



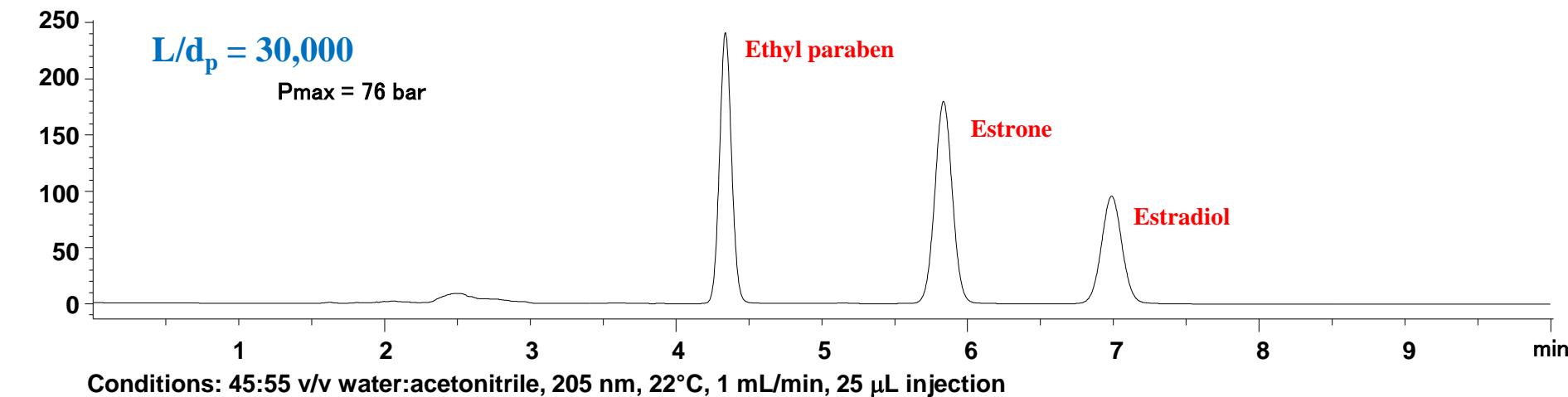
## Method Translations: Isocratic Estradiol Porous → Porous (III)

Original Monograph Separation: ACE 10 $\mu$ m, L1 (C18), 300 x 3.9 mmTranslated Monograph Separation: ACE Excel 2 $\mu$ m, L1 (C18), 50 x 3.0 mm

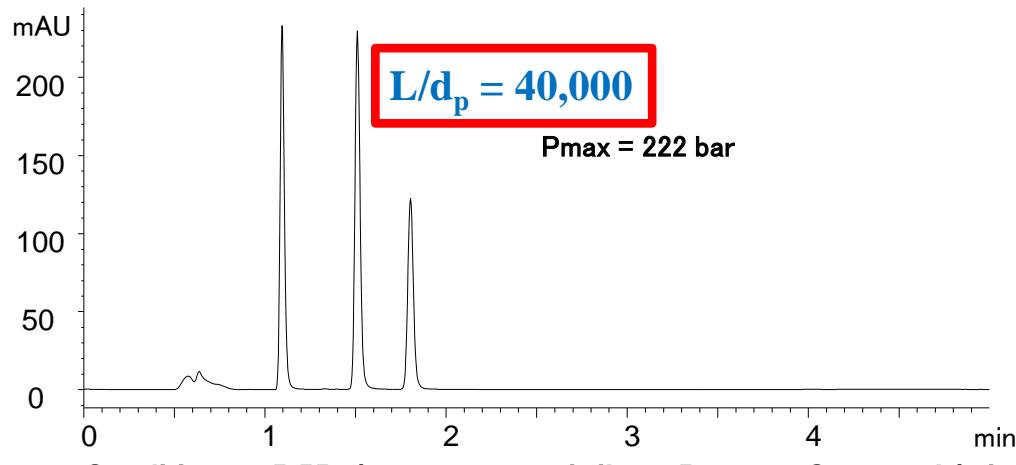


## Method Translations: Isocratic Estradiol Porous → Solid Core

## Original Monograph Separation: ACE 10μm, L1 (C18), 300 x 3.9 mm



## Translated Monograph Separation: ACE UltraCore 2.5μm, L1 (SuperC18), 100 x 4.6 mm

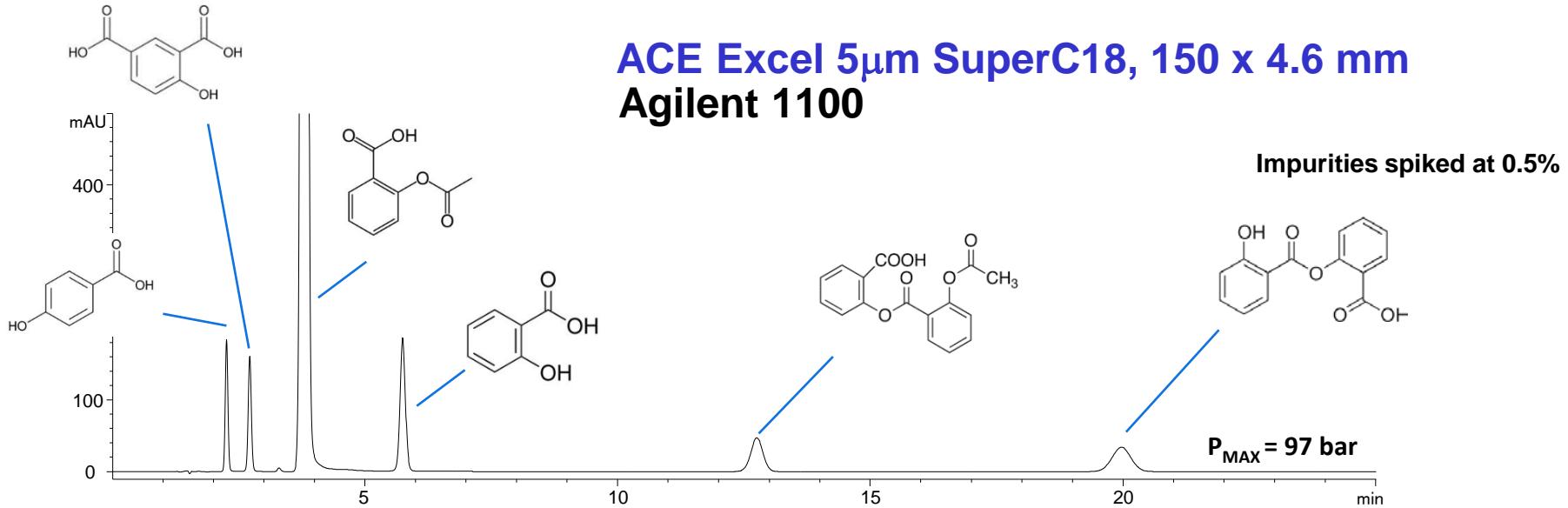


Requirement	Requirement met?
$Rs > 2.0$	✓
%RSD < 2.0%	✓

➤ Look out for negative effects of extra-column band broadening!

$$V_M \approx \pi \left(\frac{d}{2}\right)^2 L \varepsilon$$

## Isocratic Method Translations: Aspirin Porous → Solid Core (I)


Translation to ACE UltraCore 5 $\mu$ m SuperC18, 150 x 4.6 mm:

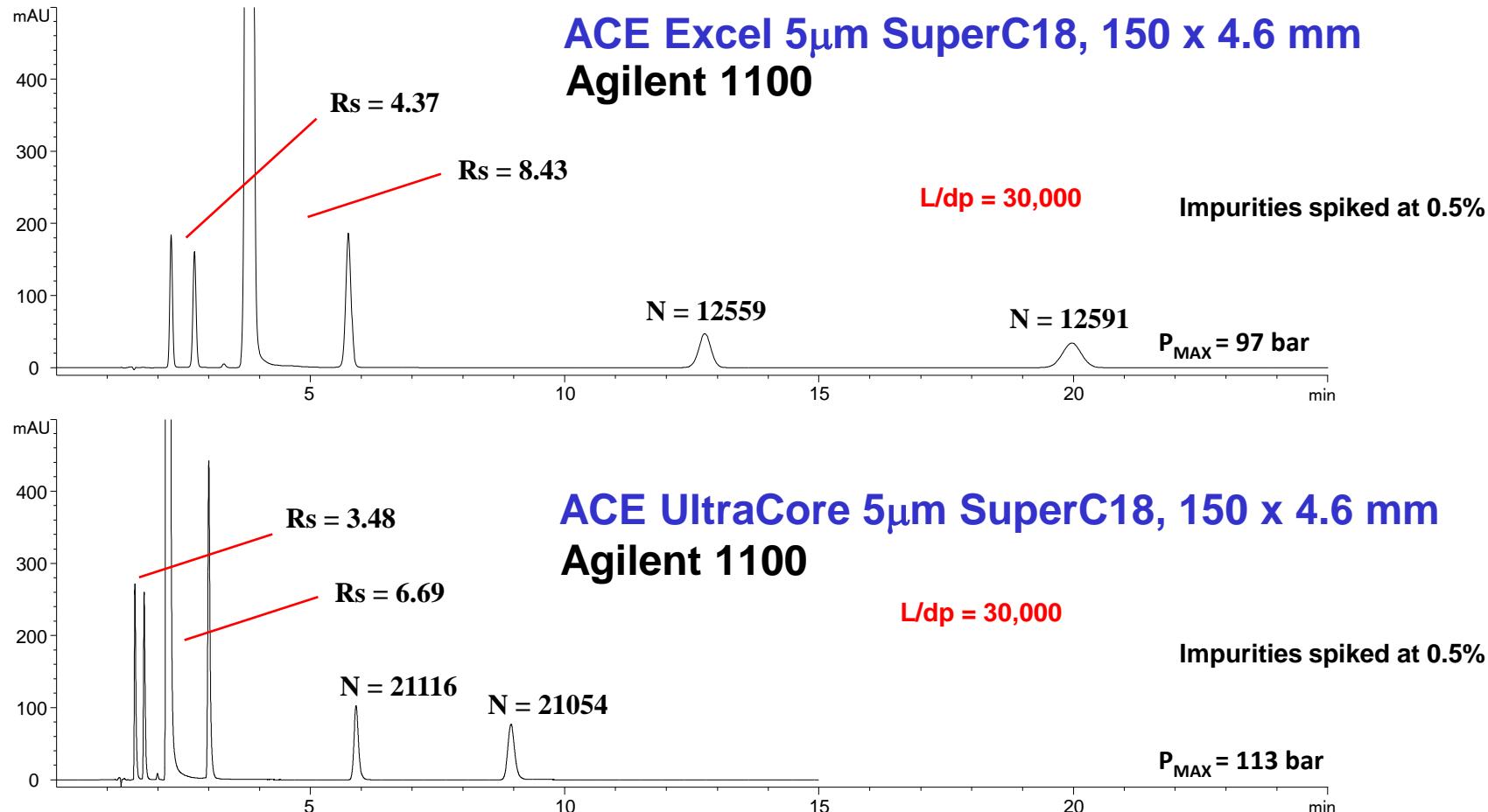
1. Column dimensions and particle size similar so same flow rate used

2. Determine column volumes and scale injection volume     $V_M \approx \pi \left(\frac{d}{2}\right)^2 L \varepsilon$

$$Inj_2 = Inj_1 \times \left( \frac{V_{m2}}{V_{m1}} \right)$$

Conditions: 60:35:5:0.2 v/v/v/v water:acetonitrile:methanol:85% phosphoric acid, 237 nm, 25°C, 1 mL/min, 5  $\mu$ L injection

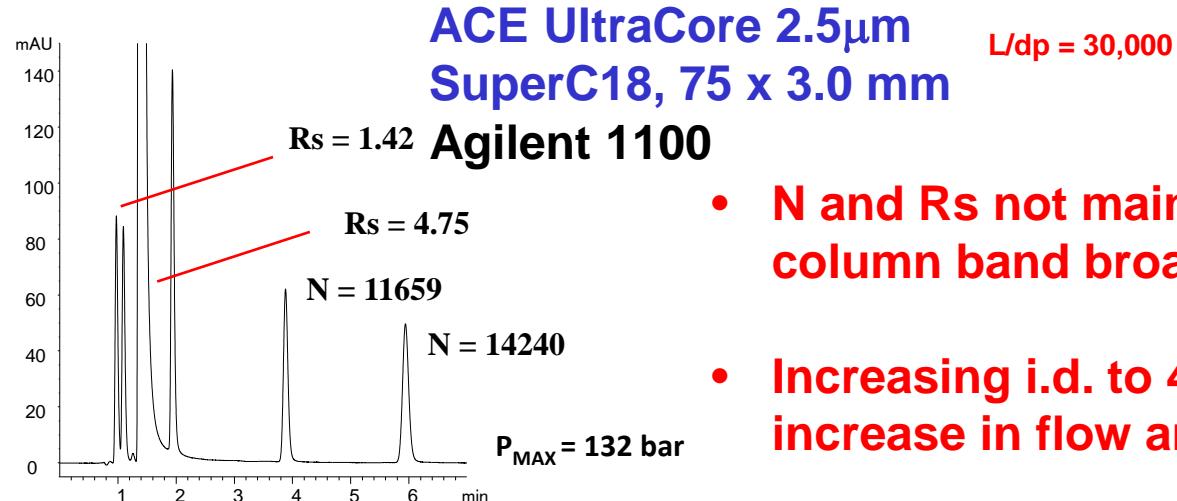
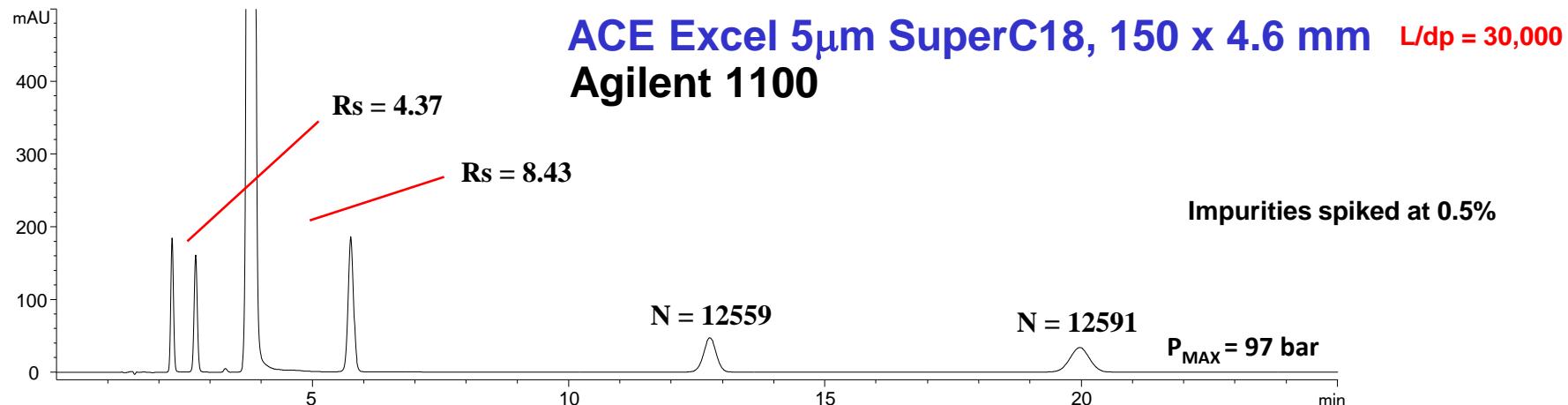
# Isocratic Method Translations: Aspirin Porous → Solid Core (II)



- Reduced hydrophobicity of solid core particles leads to ‘faster’ analysis
- Can we go faster...? (within allowable range)

Conditions: (Top): 60:35:5:0.2 v/v/v/v water:acetonitrile:methanol:85% phosphoric acid, 237 nm (2.5 Hz), 25°C, 1 mL/min, 5  $\mu$ L injection  
 (Bottom): 60:35:5:0.2 v/v water:acetonitrile:methanol:85% phosphoric acid, 237 nm (20 Hz), 25°C, 1 mL/min, 3.9  $\mu$ L injection

# Isocratic Method Translations: Aspirin Porous → Solid Core (III)



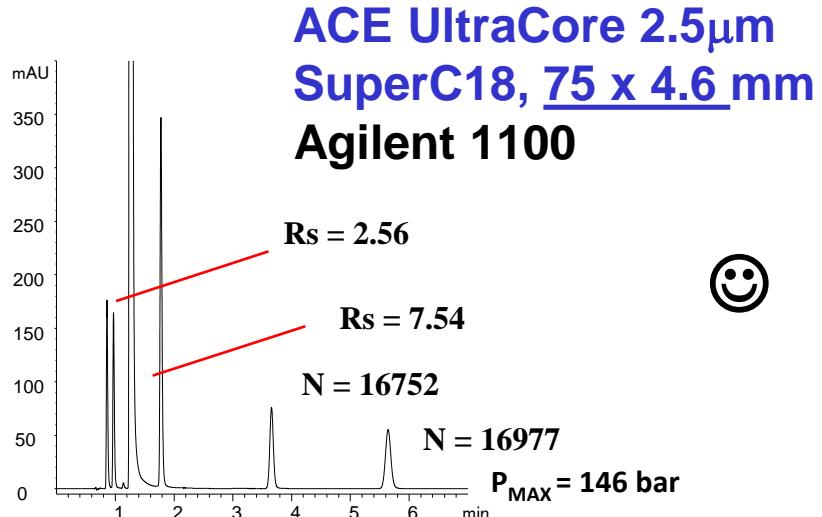
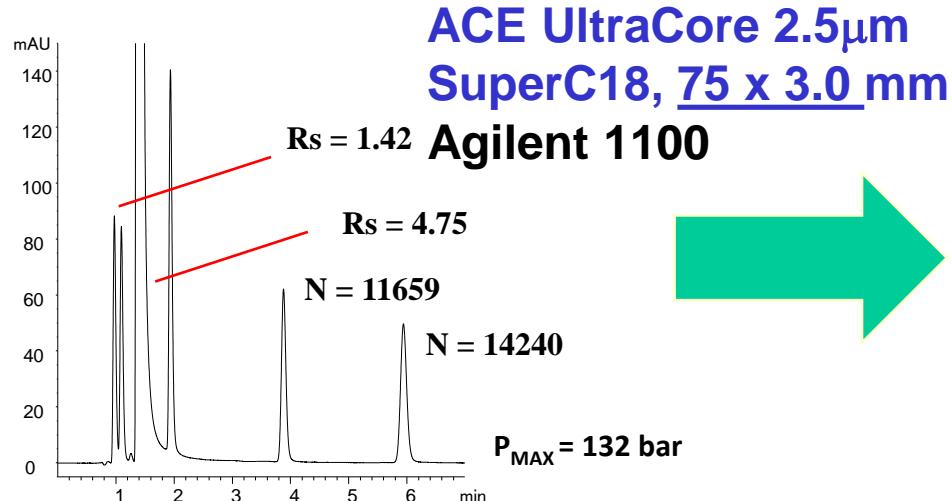
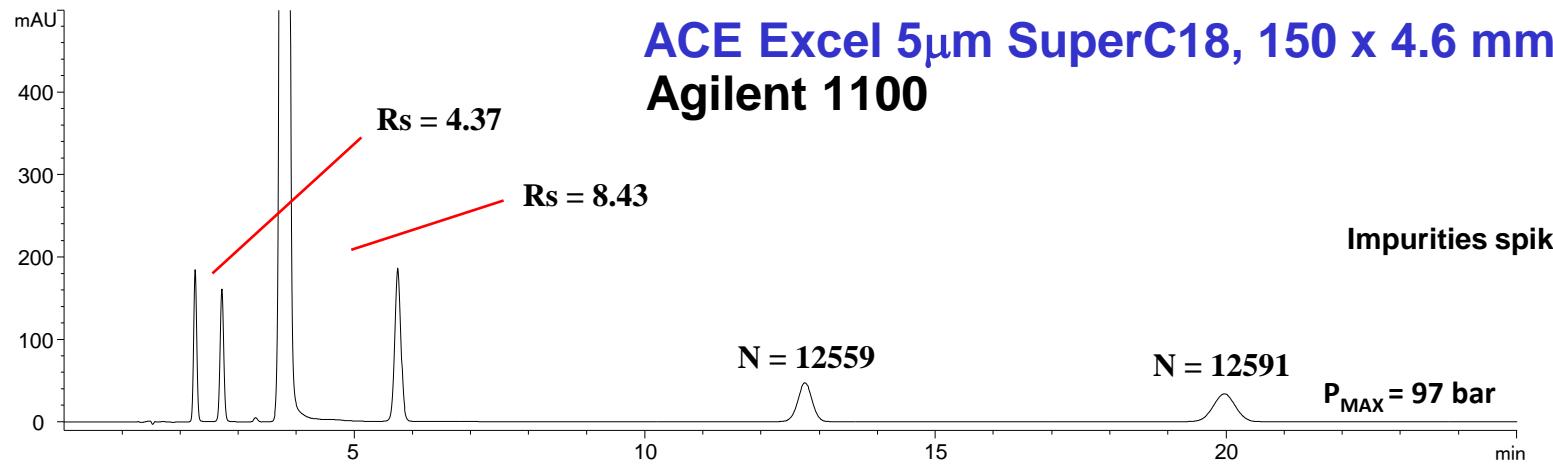
- **N and Rs not maintained. Impact of extra column band broadening of LC system** 😞
- **Increasing i.d. to 4.6 mm (with associated increase in flow and inj. vol.) should help.** ?

Conditions

(Top): 60:35:5:0.2 v/v/v/v water:acetonitrile:methanol:85% phosphoric acid, 237 nm, 25°C, 1 mL/min, 5  $\mu$ L injection  
 (Bottom): 60:35:5:0.2 v/v/v/v water:acetonitrile:methanol:85% phosphoric acid, 237 nm (20 Hz), 25°C, 0.43 mL/min, 0.9  $\mu$ L injection



## Isocratic Method Translations: Aspirin Porous → Solid Core (IV)



Conditions

ACE UltraCore 2.5 Super C18 75 x 4.6 mm:

60:35:5:0.2 v/v/v/v water:acetonitrile:methanol:85% phosphoric acid, 237 nm (20 Hz), 25°C, 1.0 mL/min, 2  $\mu$ L injection

## Gradient Translations



## Translations of Gradient Methods – The Basics

### 1. Calculate column volumes

- Better to experimentally determine porosity for accuracy

$$V_M \approx \pi \left(\frac{d}{2}\right)^2 L \varepsilon$$

### 2. Translate injection volume

- To give similar response

$$Inj_2 = Inj_1 \times \left(\frac{V_{m2}}{V_{m1}}\right)$$

Experimentally determined

### 3. Translate flow rate

- Constant linear velocity

$$F_2 = F_1 \times \frac{d_{c_2}^2}{d_{c_1}^2}$$

- Scaled to new  $d_p$

$$F_2 = \frac{F_1 \times d_{c2}^2 \times d_{p1}}{d_{c1}^2 \times d_{p2}}$$

### 4. Translate gradient time

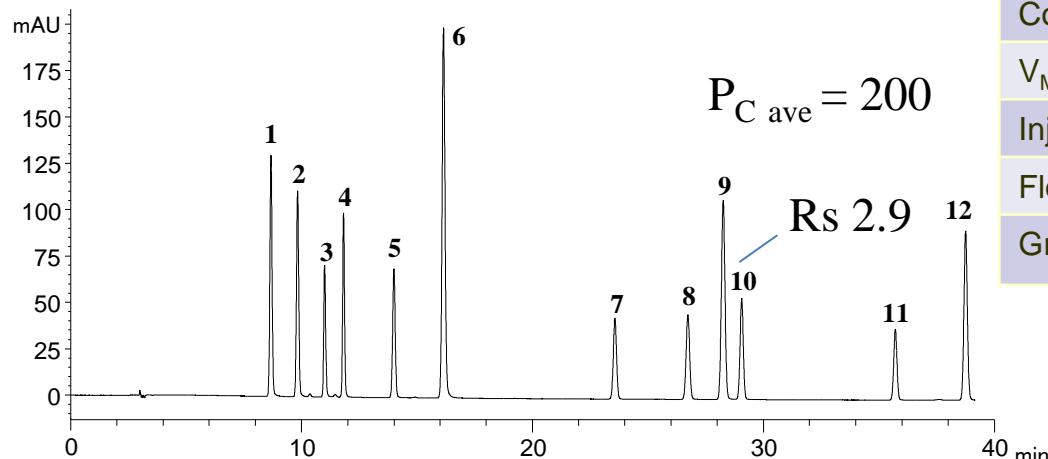
- To maintain constant  $k^*$

$$t_{G2} = \frac{t_{G1} V_{M2} F_1}{V_{M1} F_2}$$

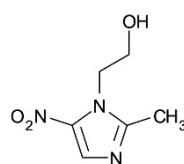
### 5. Consider $V_D/V_M$ ratio

# Gradient Method Translations: Porous → Porous

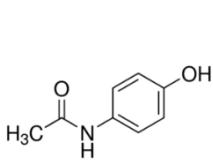
ACE Excel 5 C18-Amide 250 x 4.6 mm, 1 mL/min



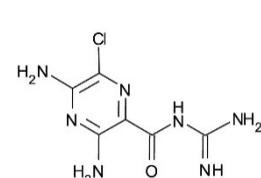
	Original Method
Column	250 x 4.6 mm, 5 $\mu\text{m}$
$V_M$	2.668 mL
Injection Vol.	10 $\mu\text{L}$
Flow Rate	1.0 mL/min
Gradient time	45 min.



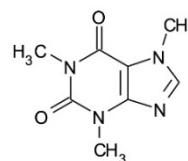
1. Metronidazole



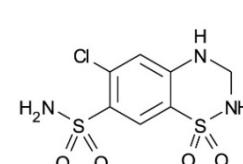
2. 4-acetamidophenol



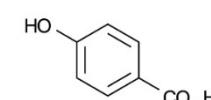
3. amiloride



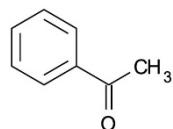
4. caffeine



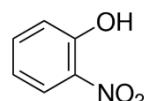
5. Hydrochlorothiazide



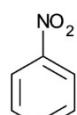
6. 4-Hydroxybenzoic acid



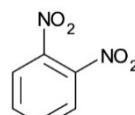
7. Acetophenone



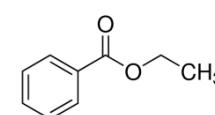
8. 2-nitrophenol



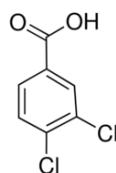
9. Nitrobenzene



10. 1,2-Dinitrobenzene



11. ethylbenzoate



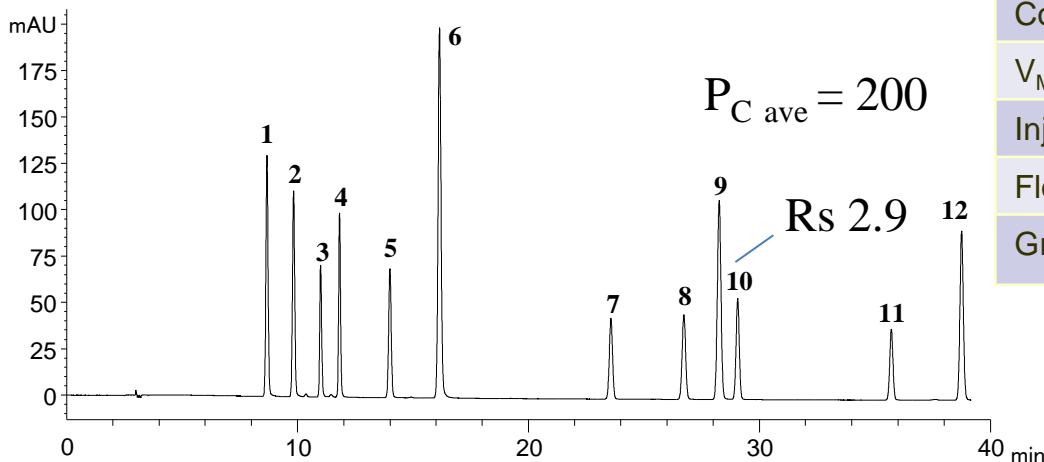
12. 3,4-dichlorobenzoic acid

Gradient analysis, A: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 (aq), B: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 in MeCN:water 65:35 v/v, 5-100% B in 45.0 mins, hold 100% B for 5.0 mins, 40°C, 1.0 mL/min, 254 nm.

1. metronidazole, 2. 4-acetamidophenol, 3. amiloride, 4. caffeine, 5. hydrochlorothiazide, 6. 4-hydroxybenzoic acid, 7. acetophenone, 8. 2-nitrophenol, 9. nitrobenzene, 10. 1,2-dinitrobenzene, 11. ethylbenzoate, 12. 3,4-dichlorobenzoic acid.

# Gradient Method Translations: Porous → Porous

**ACE Excel 5 C18-Amide 250 x 4.6 mm, 1 mL/min**



	Original Method
Column	250 x 4.6 mm, 5 $\mu\text{m}$
$V_M$	2.668 mL
Injection Vol.	10 $\mu\text{L}$
Flow Rate	1.0 mL/min
Gradient time	45 min.

➤ **ACE 5 C18-Amide 250 x 4.6 mm**

$V_M = 2.668 \text{ mL}$

➤  $L/d_P = 50,000$

Gradient analysis, A: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 (aq), B: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 in MeCN:water 65:35 v/v,

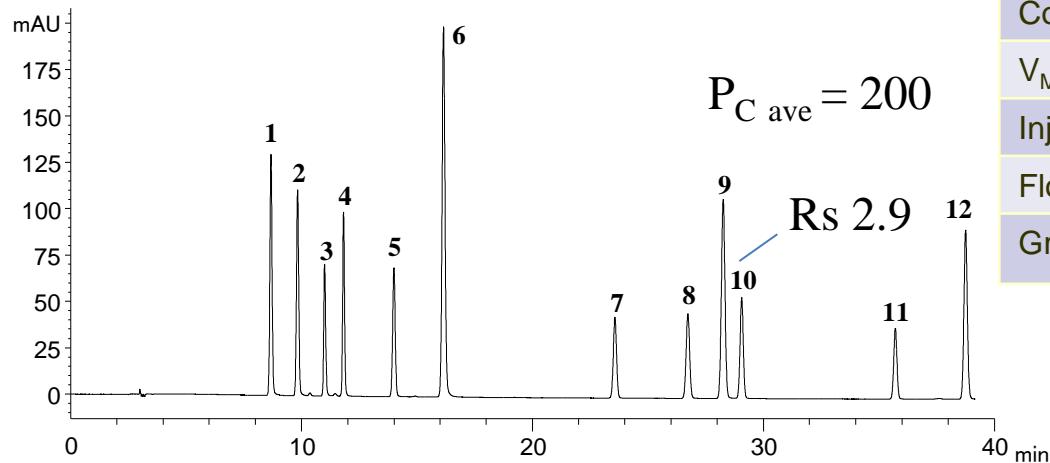
5-100% B in 45.0 mins, hold 100% B for 5.0 mins, 40°C, 1.0 mL/min, 254 nm.

1. metronidazole, 2. 4-acetamidophenol, 3. amiloride, 4. caffeine, 5. hydrochlorothiazide, 6. 4-hydroxybenzoic acid, 7.

acetophenone, 8. 2-nitrophenol, 9. nitrobenzene, 10. 1,2-dinitrobenzene, 11. ethylbenzoate, 12. 3,4-dichlorobenzoic acid.

# Gradient Method Translations: Porous → Porous

**ACE Excel 5 C18-Amide 250 x 4.6 mm, 1 mL/min**



	Original Method	Translated Method
Column	250 x 4.6 mm, 5 µm	<b>150 x 4.6 mm, 3 µm</b>
$V_M$	2.668 mL	<b>1.571 mL</b>
Injection Vol.	10 µL	
Flow Rate	1.0 mL/min	
Gradient time	45 min.	

➤ **ACE 5 C18-Amide 250 x 4.6 mm**

$$V_M = 2.668 \text{ mL}$$

➤  $L/d_P = 50,000$

➤ **ACE 3 C18-Amide 150 x 4.6 mm**

$$V_M = 1.571 \text{ mL}$$

➤  $L/d_P = 50,000$

➤ **1.  $V_M$  experimentally determined**

➤ **Or use**

$$V_M = \pi \left( \frac{d_c}{2} \right)^2 L \varepsilon$$

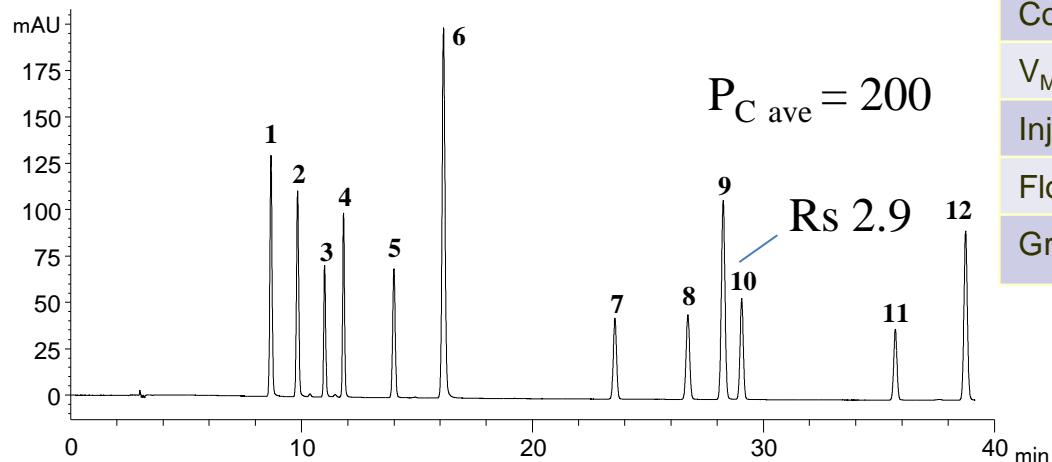
Gradient analysis, A: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 (aq), B: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 in MeCN:water 65:35 v/v, 5-100% B in 45.0 mins, hold 100% B for 5.0 mins, 40°C, 1.0 mL/min, 254 nm.

1. metronidazole, 2. 4-acetamidophenol, 3. amiloride, 4. caffeine, 5. hydrochlorothiazide, 6. 4-hydroxybenzoic acid, 7. acetophenone, 8. 2-nitrophenol, 9. nitrobenzene, 10. 1,2-dinitrobenzene, 11. ethylbenzoate, 12. 3,4-dichlorobenzoic acid.

➤  $\varepsilon = \text{column porosity} = \sim 0.63 \text{ for fully porous, } \sim 0.55 \text{ for solid core}$

# Gradient Method Translations: Porous → Porous

**ACE Excel 5 C18-Amide 250 x 4.6 mm, 1 mL/min**



	Original Method	Translated Method
Column	250 x 4.6 mm, 5 µm	150 x 4.6 mm, 3 µm
$V_M$	2.668 mL	1.571 mL
Injection Vol.	10 µL	<b>5.9 µL</b>
Flow Rate	1.0 mL/min	
Gradient time	45 min.	

## 2. Translate injection volume

$$V_{i2} = \frac{V_{i1} \times V_{M2}}{V_{M1}} = \frac{10 \times 1.571}{2.668} = 5.9 \mu L$$

Gradient analysis, A: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 (aq), B: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 in MeCN:water 65:35 v/v,

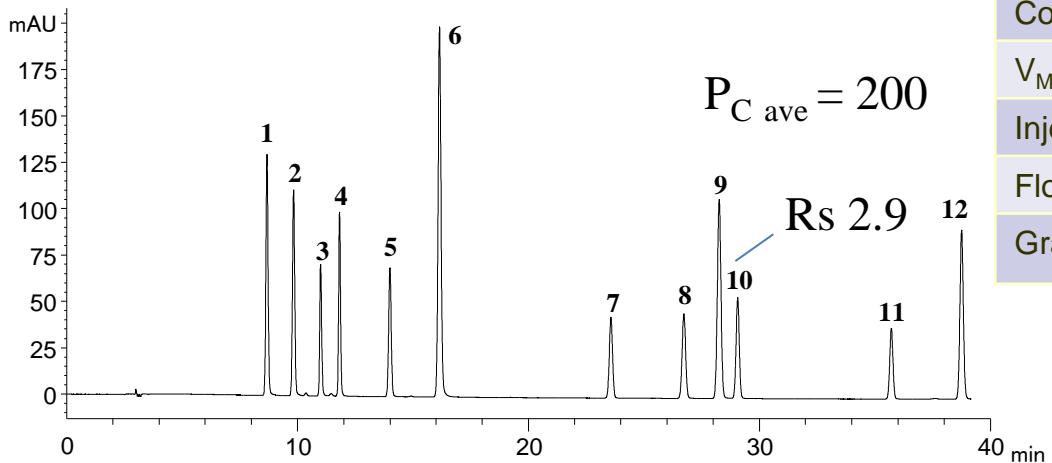
5-100% B in 45.0 mins, hold 100% B for 5.0 mins, 40°C, 1.0 mL/min, 254 nm.

1. metronidazole, 2. 4-acetamidophenol, 3. amiloride, 4. caffeine, 5. hydrochlorothiazide, 6. 4-hydroxybenzoic acid, 7.

acetophenone, 8. 2-nitrophenol, 9. nitrobenzene, 10. 1,2-dinitrobenzene, 11. ethylbenzoate, 12. 3,4-dichlorobenzoic acid.

# Gradient Method Translations: Porous → Porous

**ACE Excel 5 C18-Amide 250 x 4.6 mm, 1 mL/min**



	Original Method	Translated Method
Column	250 x 4.6 mm, 5 µm	150 x 4.6 mm, 3 µm
$V_M$	2.668 mL	1.571 mL
Injection Vol.	10 µL	5.9 µL
Flow Rate	1.0 mL/min	<b>1.67 mL/min</b>
Gradient time	45 min.	

## 2. Translate injection volume

$$V_{i2} = \frac{V_{i1} \times V_{M2}}{V_{M1}} = \frac{10 \times 1.571}{2.668} = 5.9 \mu\text{L}$$

## 3. Translate flow rate

$$F_2 = \frac{F_1 \times d_{c2}^2 \times d_{p1}}{d_{c1}^2 \times d_{p2}} = \frac{1.00 \times 4.6^2 \times 5}{4.6^2 \times 3} = 1.67 \text{ mL/min}$$

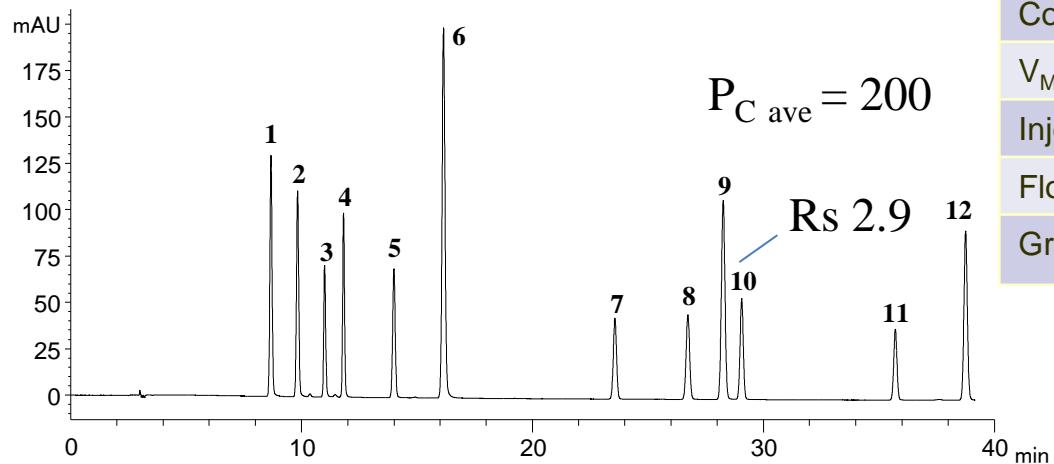
Gradient analysis, A: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 (aq), B: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 in MeCN:water 65:35 v/v,

5-100% B in 45.0 mins, hold 100% B for 5.0 mins, 40°C, 1.0 mL/min, 254 nm.

1. metronidazole, 2. 4-acetamidophenol, 3. amiloride, 4. caffeine, 5. hydrochlorothiazide, 6. 4-hydroxybenzoic acid, 7. acetophenone, 8. 2-nitrophenol, 9. nitrobenzene, 10. 1,2-dinitrobenzene, 11. ethylbenzoate, 12. 3,4-dichlorobenzoic acid.

# Gradient Method Translations: Porous → Porous

**ACE Excel 5 C18-Amide 250 x 4.6 mm, 1 mL/min**



	Original Method	Translated Method
Column	250 x 4.6 mm, 5 µm	150 x 4.6 mm, 3 µm
$V_M$	2.668 mL	1.571 mL
Injection Vol.	10 µL	5.9 µL
Flow Rate	1.0 mL/min	1.67 mL/min
Gradient time	45 min.	<b>15.87 min.</b>

### 3. Translate gradient time

$$t_{G2} = \frac{t_{G1} V_{M2} F_1}{V_{M1} F_2} = \frac{45 \times 1.571 \times 1.0}{2.668 \times 1.67} = 15.87 \text{ min.}$$

Gradient analysis, A: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 (aq), B: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 in MeCN:water 65:35 v/v,

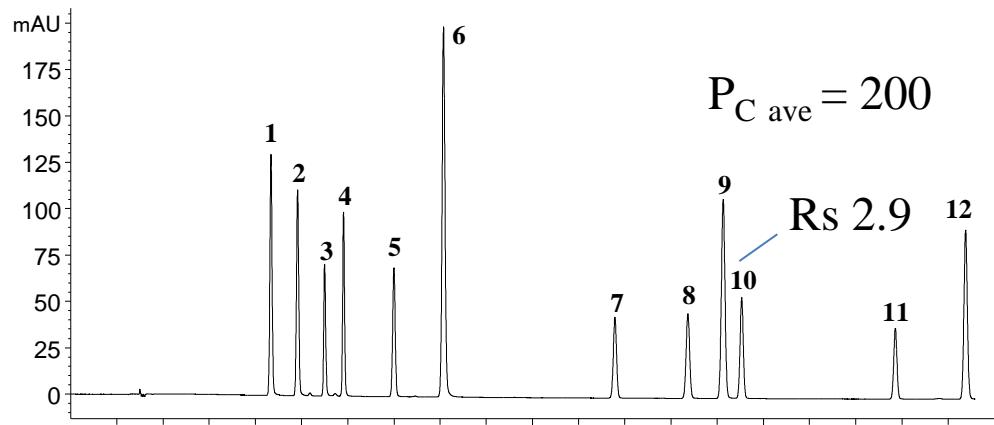
5-100% B in 45.0 mins, hold 100% B for 5.0 mins, 40°C, 1.0 mL/min, 254 nm.

1. metronidazole, 2. 4-acetamidophenol, 3. amiloride, 4. caffeine, 5. hydrochlorothiazide, 6. 4-hydroxybenzoic acid, 7.

acetophenone, 8. 2-nitrophenol, 9. nitrobenzene, 10. 1,2-dinitrobenzene, 11. ethylbenzoate, 12. 3,4-dichlorobenzoic acid.

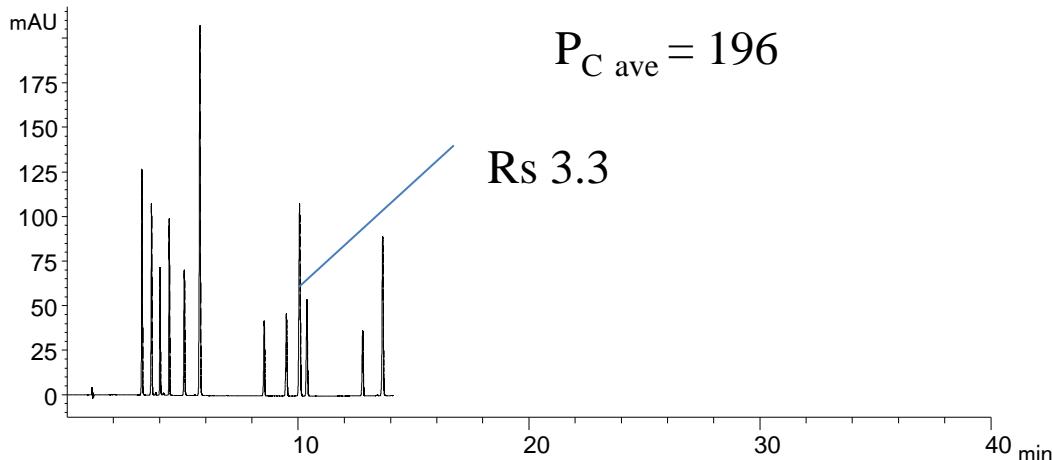
# Gradient Method Translations: Porous → Porous

**ACE Excel 5 C18-Amide 250 x 4.6 mm, 1 mL/min**



	Original Method	Translated Method
Column	250 x 4.6 mm, 5 µm	150 x 4.6 mm, 3 µm
$V_M$	2.668 mL	1.571 mL
Injection Vol.	10 µL	5.9 µL
Flow Rate	1.0 mL/min	1.67 mL/min
Gradient time	45 min.	15.87 min.

**ACE Excel 3 C18-Amide 150 x 4.6 mm, 1.67 mL/min**



**Total run time:** -64.7%

**Solvent consumption:** -41%

Gradient analysis, A: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 (aq), B: 20 mM KH<sub>2</sub>PO<sub>4</sub> pH 2.7 in MeCN:water 65:35 v/v, 5-100% B in 45.0 mins, hold 100% B for 5.0 mins, 40°C, 1.0 mL/min, 254 nm.

1. metronidazole, 2. 4-acetamidophenol, 3. amiloride, 4. caffeine, 5. hydrochlorothiazide, 6. 4-hydroxybenzoic acid, 7. acetophenone, 8. 2-nitrophenol, 9. nitrobenzene, 10. 1,2-dinitrobenzene, 11. ethylbenzoate, 12. 3,4-dichlorobenzoic acid.

## Step 1. Calculate Column Volumes

- Due to column volume differences with porous and solid core columns, errors can creep into gradient translation work

		Porous (0.63) column volumes						
		Column Length						
		20mm	30mm	50mm	75mm	100mm	150mm	250mm
Col. i.d.	1mm	0.010	0.015	0.025	0.037	0.049	0.074	0.124
	2.1mm	0.044	0.065	0.109	0.164	0.218	0.327	0.546
	3.0mm	0.089	0.134	0.223	0.334	0.445	0.668	1.113
	4.6mm	0.209	0.314	0.523	0.785	1.047	1.570	2.617

		Solid core (0.55) column volumes						
		Column Length						
		20mm	30mm	50mm	75mm	100mm	150mm	250mm
Col. i.d.	1mm	0.009	0.013	0.022	0.032	0.043	0.065	0.108
	2.1mm	0.038	0.057	0.095	0.143	0.190	0.286	0.476
	3.0mm	0.078	0.117	0.194	0.292	0.389	0.583	0.972
	4.6mm	0.183	0.274	0.457	0.686	0.914	1.371	2.285

$$V_M \approx \pi \left(\frac{d}{2}\right)^2 L \varepsilon$$

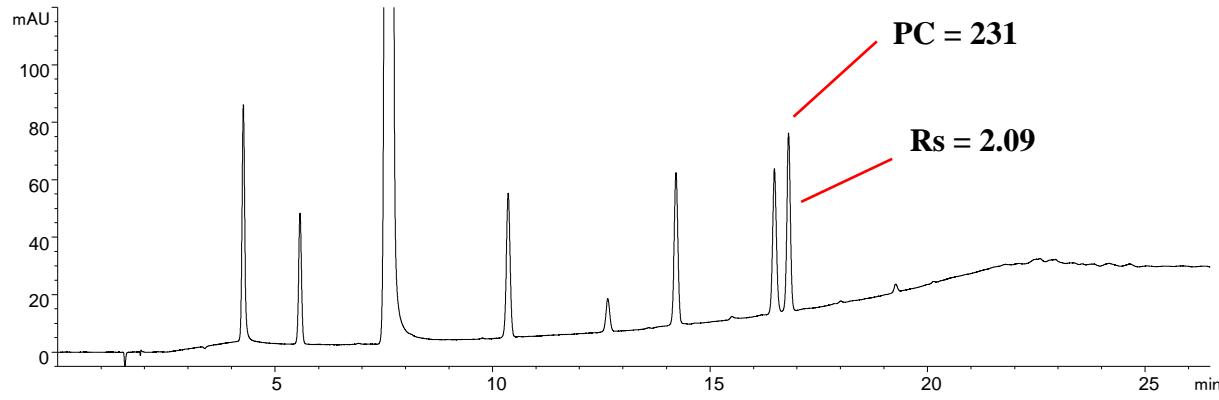
**Porosity values:**  
porous = 0.63  
solid core = 0.55

- Solid core particle columns have ~13% LESS column volume than porous particle columns
- Better to experimentally determine porosity for vendor column



## Gradient Method Translations: Acetaminophen Porous → Solid Core

## ACE Excel 5μm SuperC18 150 x 4.6 mm



$t_G$	20 min
Flow:	1 mL/min
Inj. Vol.	5 μL
$P_{MAX}$	138 bar

$$\text{Peak Capacity} = 1 + \frac{t_G}{W_{0.5}}$$

Translation to ACE UltraCore 5μm SuperC18 150 x 4.6 mm:

1. Calculate column volumes...solid core porosity differences
2. Translate gradient time  $\frac{t_{G1}F_1}{V_{M1}} = \frac{t_{G2}F_2}{V_{M2}}$   $V_M \approx \pi \left(\frac{d}{2}\right)^2 L \varepsilon$
3. No flow rate change needed  $F_2 = F_1 \times \frac{d_{c_2}^2}{d_{c_1}^2}$
4. Translate injection volume  $Inj_2 = Inj_1 \times \left(\frac{V_{m2}}{V_{m1}}\right)$
5. Calculate whether an injection hold or pre-injection is needed

150x4.6mm, 5 μm, gradient analysis, A= 20mM ammonium acetate pH 6.0 (aq), B= 20mM ammonium acetate pH 6.0 in MeCN:water 9:1 v/v, 5-95% B in 20.0 mins, hold 95% B for 5.0 mins, 30°C, 1.0 mL/min, 230 nm.

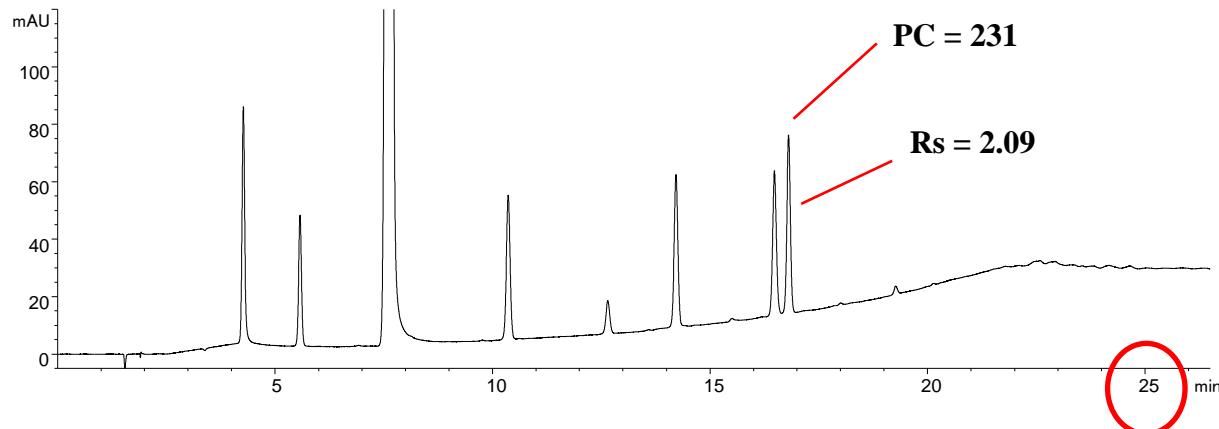
1. 4-aminophenol 2. Hydroquinone 3. acetaminophen 4. 4-acetamidophenol 5. phenol 6. 4-nitrophenol 7. 2-nitrophenol 8. 4-chloroacetanilide

$$\Delta = \left( \frac{V_D}{V_M} \right)_{original} - \left( \frac{V_D}{V_M} \right)_{new}$$



## Gradient Method Translations: Acetaminophen Porous → Solid Core

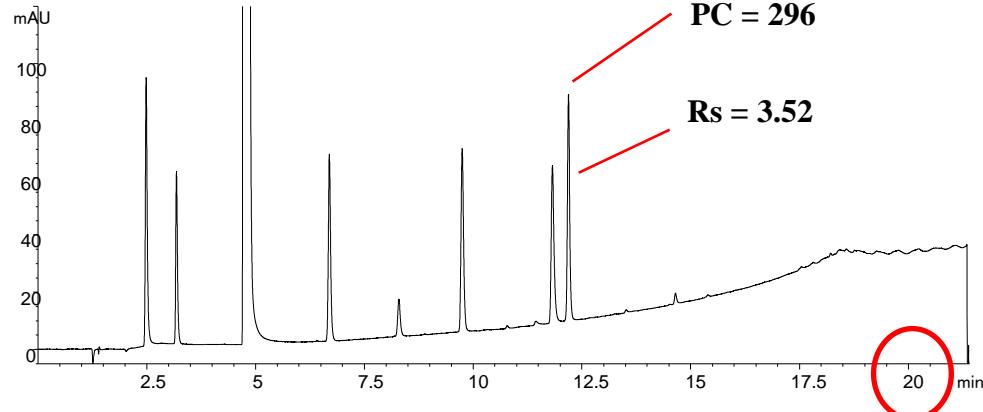
## ACE Excel 5μm SuperC18 150 x 4.6 mm



t <sub>G</sub>	20 min
Flow:	1 mL/min
Inj. Vol.	5 μL
P <sub>MAX</sub>	138 bar

$$\text{Peak Capacity} = 1 + \frac{t_G}{W_{0.5}}$$

## ACE UltraCore 5μm SuperC18 150 x 4.6 mm



t <sub>G</sub>	16.4 min
Flow:	1 mL/min
Inj. Vol.	3.9 μL
P <sub>MAX</sub>	159 bar

- Corrected  
- V<sub>m</sub> (porosity)  
- t<sub>G</sub> adjusted for constant k<sup>\*</sup>  
- Injection volume adjusted

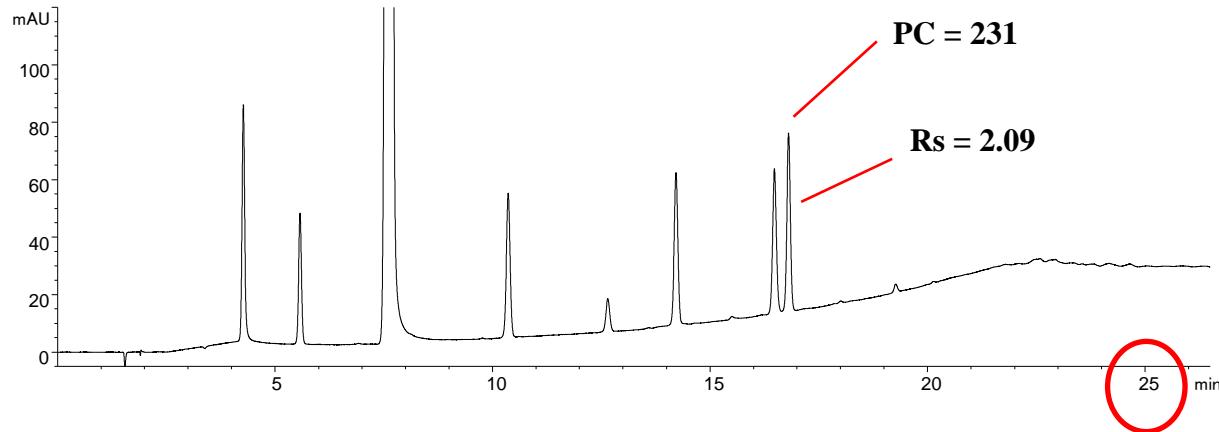
Run time reduced whilst peak capacity and resolution are increased

$$\frac{t_{G1}F_1}{V_{M1}} = \frac{t_{G2}F_2}{V_{M2}}$$



## Gradient Method Translations: Acetaminophen Porous → Solid Core

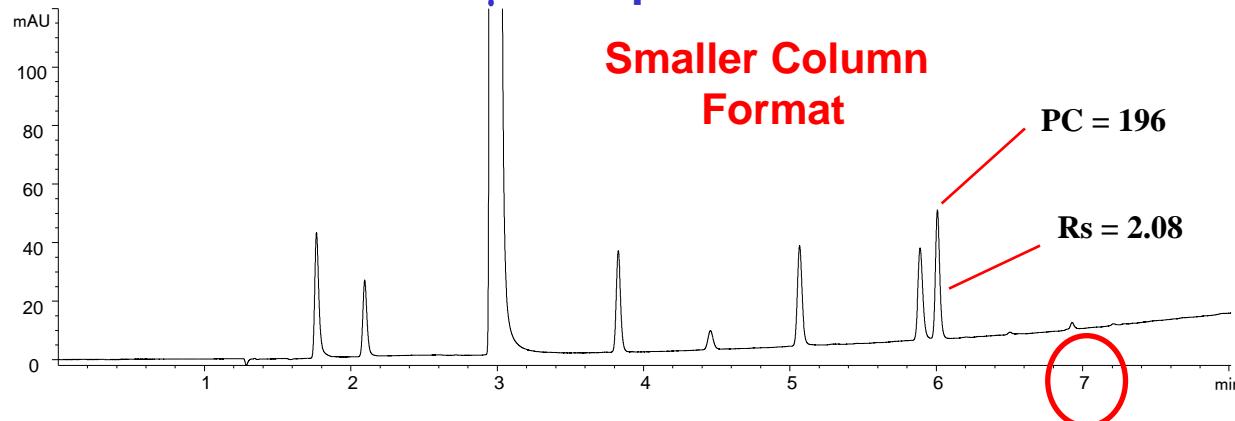
## ACE Excel 5μm SuperC18 150 x 4.6 mm



$t_G$  20 min  
Flow: 1 mL/min  
Inj. Vol. 5 μL  
 $P_{MAX}$  138 bar

$$\text{Peak Capacity} = 1 + \frac{t_G}{W_{0.5}}$$

## ACE UltraCore 2.5μm SuperC18 50 x 3.0 mm



$t_G$  5.92 min  
Flow: 0.43 mL/min  
Inj. Vol. 0.63 μL  
 $P_{MAX}$  162 bar

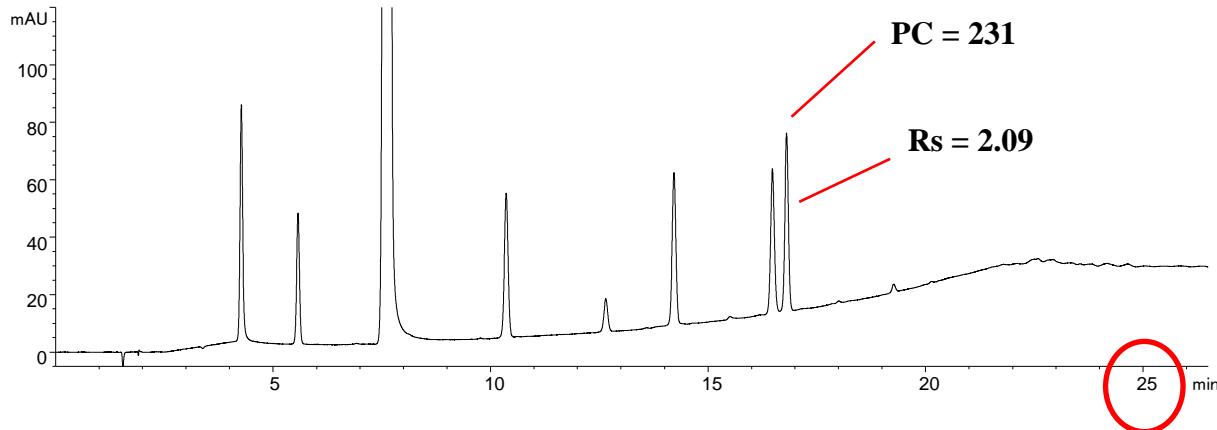
- Corrected
- $V_m$  (porosity)
  - Constant  $k^*$  ( $t_G$  & F)
  - Injection volume ( $V_m$ )

Significant reduction in run time whilst maintaining resolution

$$\frac{t_{G1}F_1}{V_{M1}} = \frac{t_{G2}F_2}{V_{M2}}$$

# Gradient Method Translations: Acetaminophen Porous → Solid Core

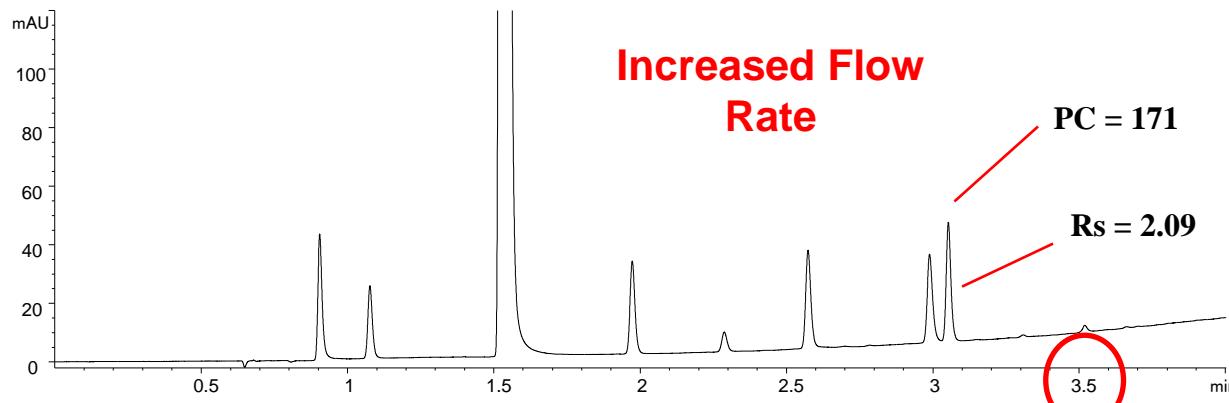
## ACE Excel 5µm SuperC18 150 x 4.6 mm



$t_G$	20 min
Flow:	1 mL/min
Inj. Vol.	5 µL
$P_{MAX}$	138 bar

$$\text{Peak Capacity} = 1 + \frac{t_G}{W_{0.5}}$$

## ACE UltraCore 2.5µm SuperC18 50 x 3.0 mm



$t_G$	2.99 min
Flow:	0.85 mL/min
Inj. Vol.	0.63 µL
$P_{MAX}$	315 bar

Corrected  
- Constant  $k^*$  ( $t_G$  & F)

**Significant reduction in run time whilst still maintaining resolution**

$$\frac{t_{G1}F_1}{V_{M1}} = \frac{t_{G2}F_2}{V_{M2}}$$



## Gradient Method Translations: Correcting for $V_D/V_M$

- It is **important** when translating to **smaller i.d.** (therefore **small  $V_M$** ) UHPLC columns to consider the  $V_D/V_M$  ratio.

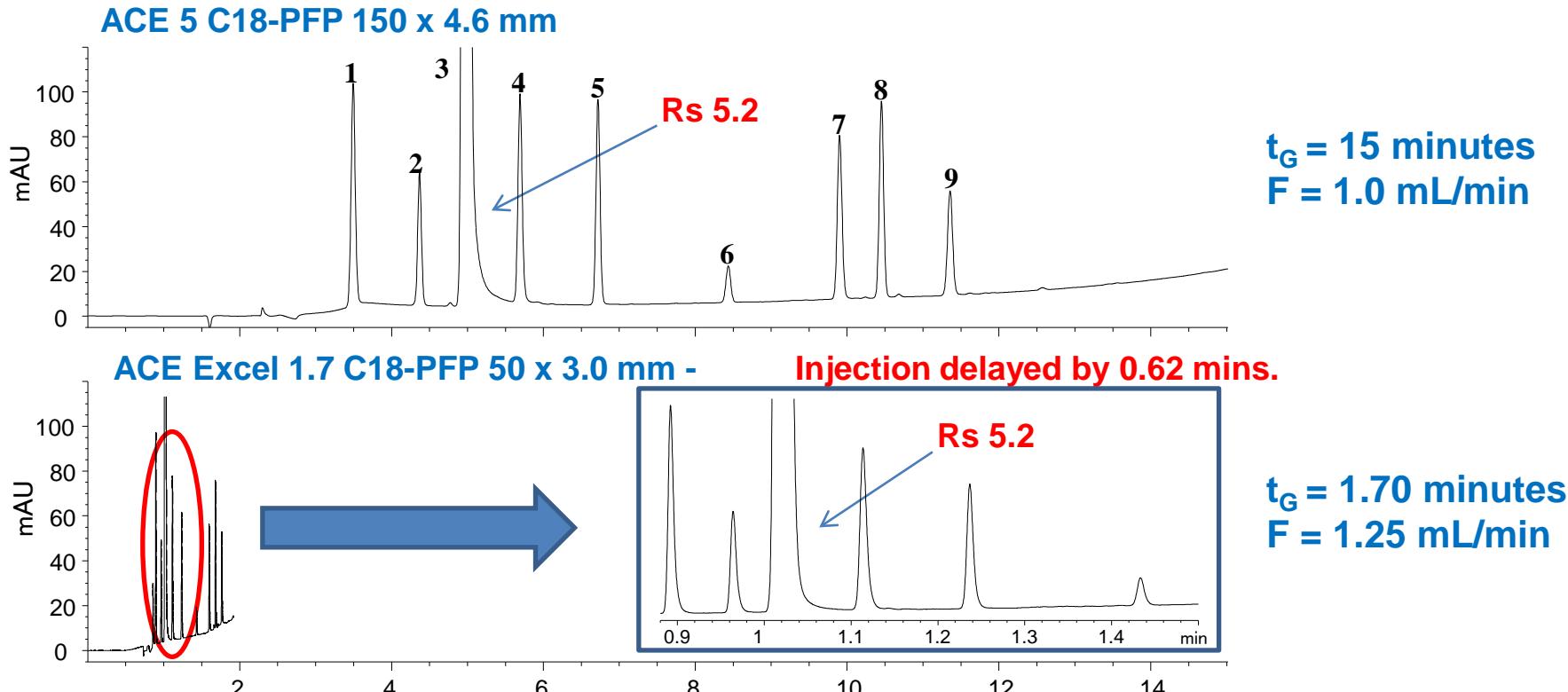
$$\Delta = \left( \frac{V_D}{V_M} \right)_{original} - \left( \frac{V_D}{V_M} \right)_{new} \text{ must approach zero}$$

- For **maximum accuracy** of gradient translations, it may be necessary to correct for the **change in ratio of  $V_D/V_M$**  between the **original** and **translated** method.
- The correction can be made using

$$x = \left[ \left( \frac{V_{D1}}{V_{M1}} \right) - \left( \frac{V_{D2}}{V_{M2}} \right) \right] \times \frac{V_{M2}}{F_2}$$

- **Negative value:** injection must be delayed x minutes after gradient starts.
- **Positive value:** a **pre-gradient isocratic hold** of x minutes should be added to gradient program.

## Gradient Method Translations: Why correct for $V_D/V_M$ ?

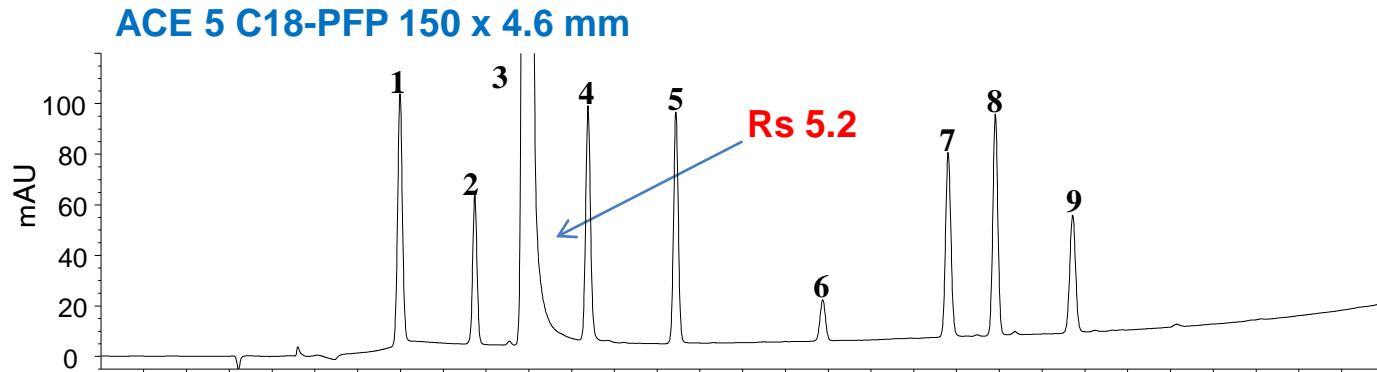


- Calculations tell us to **delay injection until 0.62 minutes** after the gradient starts to correct for  $V_D/V_M$

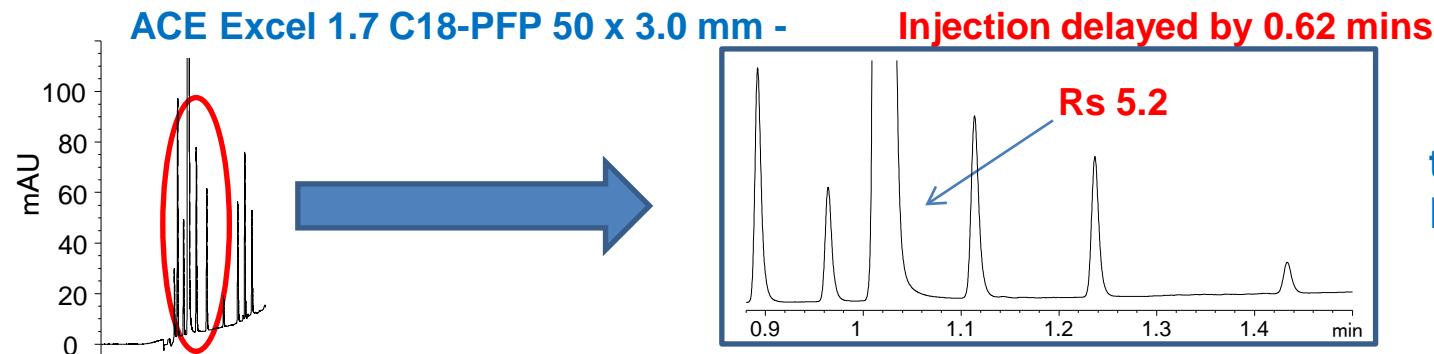
$$x = \left[ \left( \frac{V_{D1}}{V_{M1}} \right) - \left( \frac{V_{D2}}{V_{M2}} \right) \right] \times \frac{V_{M2}}{F_2} = \left[ \left( \frac{1.098}{1.570} \right) - \left( \frac{0.926}{0.223} \right) \right] \times \frac{0.223}{1.25} = -0.62 \text{ minutes}$$

Gradient analysis, A= 20mM ammonium acetate pH 6.0 (aq), B= 20 mM ammonium acetate pH 6.0 in MeCN:water 80:20 v/v, 5-95% B in 15.0 mins, hold 95% B for 2.0 mins, 40°C, 1.0 mL/min, 230 nm.

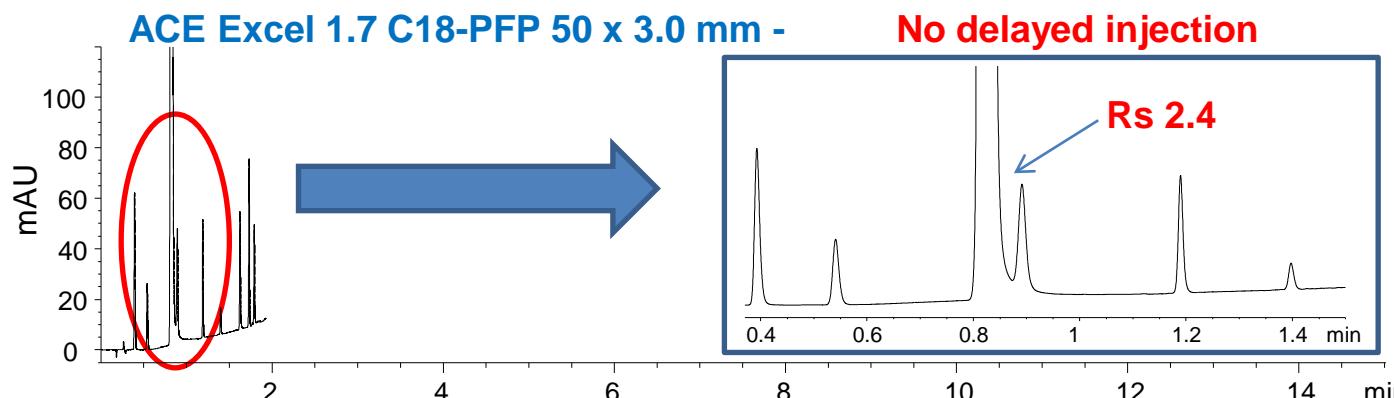
1. 4-aminophenol, 2. hydroquinone, 3. 4-acetamidophenol (paracetamol), 4. 2-aminophenol, 5. 2-acetamidophenol, 6. phenol, 7. 4-nitrophenol, 8. 4-chloroacetanilide, 9. 2-nitrophenol. Impurities were spiked at 0.5% w/w.

Gradient Method Translations: Why correct for  $V_D/V_M$ ?

$t_G = 15$  minutes  
 $F = 1.0$  mL/min



$t_G = 1.70$  minutes  
 $F = 1.25$  mL/min



$t_G = 1.70$  minutes  
 $F = 1.25$  mL/min

# Method Translator Tool Example - Gradient

**Input data into  
the grey boxes**

**Input dwell  
volume of LC  
systems**

<b>Gradient</b>																																																																									
<b>Column Information</b> <table border="1"> <thead> <tr> <th colspan="2"><b>Current</b></th> <th colspan="2"><b>Translated</b></th> </tr> </thead> <tbody> <tr> <td>Column Length (L)</td> <td>150:mm</td> <td>Column Length (L)</td> <td>50:mm</td> </tr> <tr> <td>Column i.d. (<math>d_c</math>)</td> <td>4.6:mm</td> <td>Column i.d. (<math>d_c</math>)</td> <td>3.0:mm</td> </tr> <tr> <td>Particle Diameter (<math>d_p</math>)</td> <td>5.0:<math>\mu\text{m}</math></td> <td>Particle Diameter (<math>d_p</math>)</td> <td>1.7:<math>\mu\text{m}</math></td> </tr> <tr> <td>L/<math>d_p</math></td> <td>30000</td> <td>L/<math>d_p</math></td> <td>29412</td> </tr> <tr> <td>Column Porosity</td> <td>0.63:What's This?</td> <td>Column Porosity</td> <td>0.63:What's This?</td> </tr> <tr> <td>Column Volume (<math>V_M</math>)</td> <td>1.570 mL</td> <td>Column Volume (<math>V_M</math>)</td> <td>0.223 mL</td> </tr> </tbody> </table>		<b>Current</b>		<b>Translated</b>		Column Length (L)	150:mm	Column Length (L)	50:mm	Column i.d. ( $d_c$ )	4.6:mm	Column i.d. ( $d_c$ )	3.0:mm	Particle Diameter ( $d_p$ )	5.0: $\mu\text{m}$	Particle Diameter ( $d_p$ )	1.7: $\mu\text{m}$	L/ $d_p$	30000	L/ $d_p$	29412	Column Porosity	0.63:What's This?	Column Porosity	0.63:What's This?	Column Volume ( $V_M$ )	1.570 mL	Column Volume ( $V_M$ )	0.223 mL																																												
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**New Gradient**

This section tells you if a delayed injection of X mins is required

## Transferring Between Instruments



## Instrument to Instrument Same Method Transfers

### ◆ Isocratic

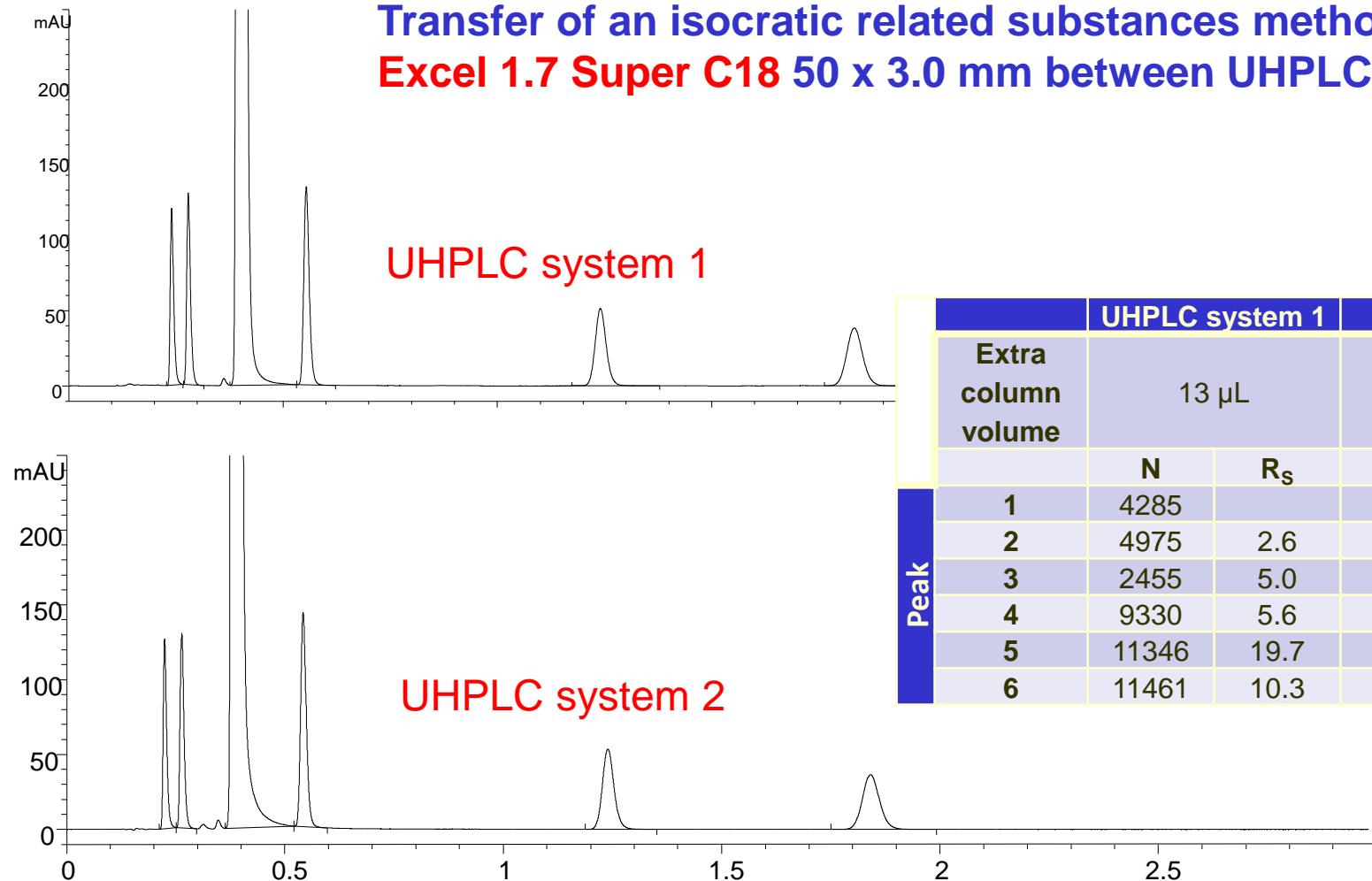
- If column format remains identical, no changes necessary.
- (If column format changes, translation required.)

### ◆ Gradient

- If column format remains identical, need to correct for influence of differing system dwell volumes only. Flow rate, injection volume and gradient times remain unchanged.
- (If column format changes, translation plus differing system dwell effects need to be calculated.)

# Instrument to instrument method transfer: Isocratic

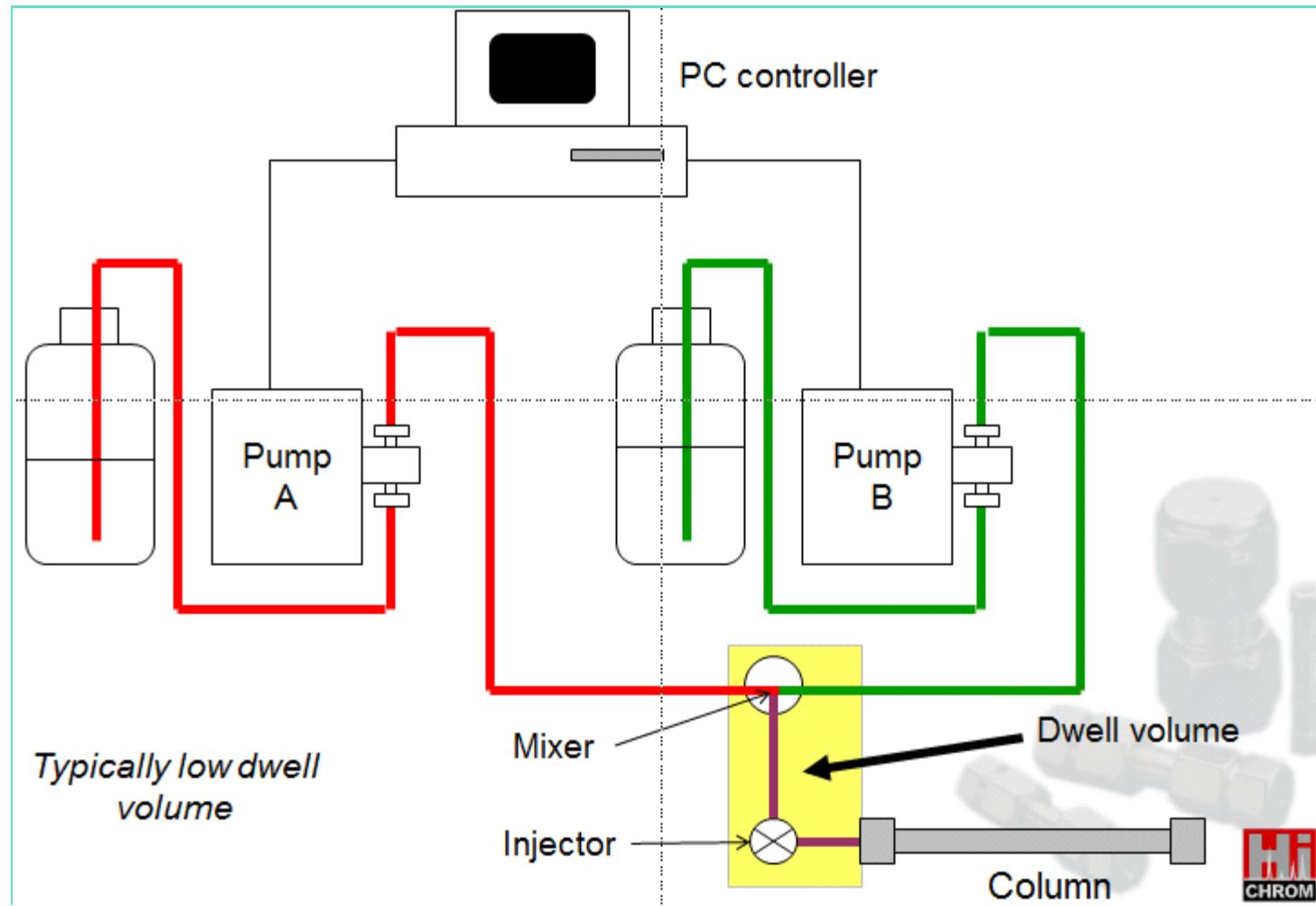
**Transfer of an isocratic related substances method on ACE Excel 1.7 Super C18 50 x 3.0 mm between UHPLC systems.**



Isocratic analysis 0.2% H<sub>3</sub>PO<sub>4</sub> in MeCN:MeOH:water 35:5:60 v/v/v, 40°C, 1.25 mL/min, 0.7 µL, Injection vol., 254 nm.  
 1. 4-hydroxybenzoic acid, 2. 4-hydroxisophthalic acid, 3. acetylsalicylic acid (aspirin), 4. salicylic acid, 5. acetylsalicylsalicylic acid, 6. salsalate. Impurities were spiked at 0.5% w/w.

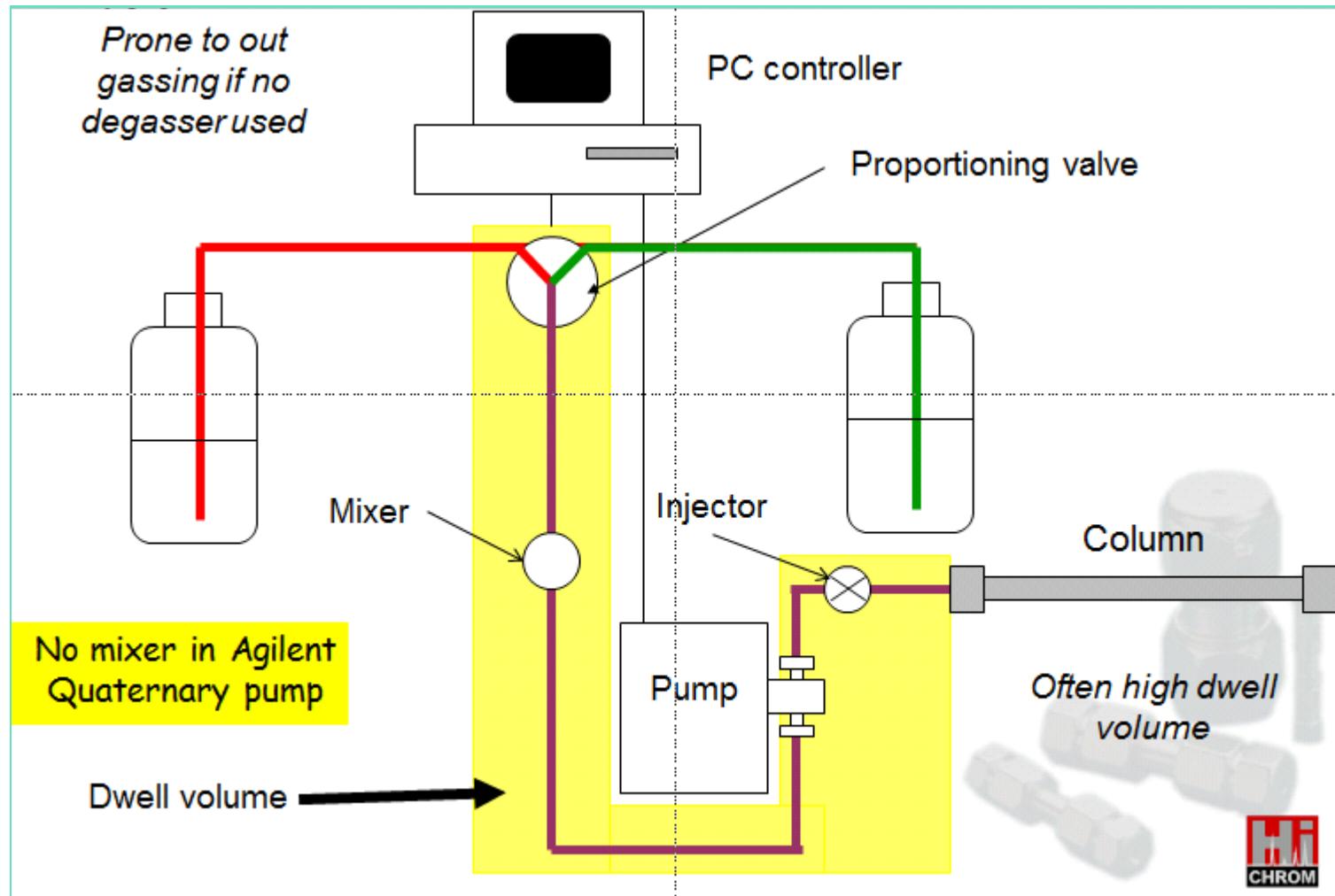


## Binary Pump Configuration: High Pressure Mixing

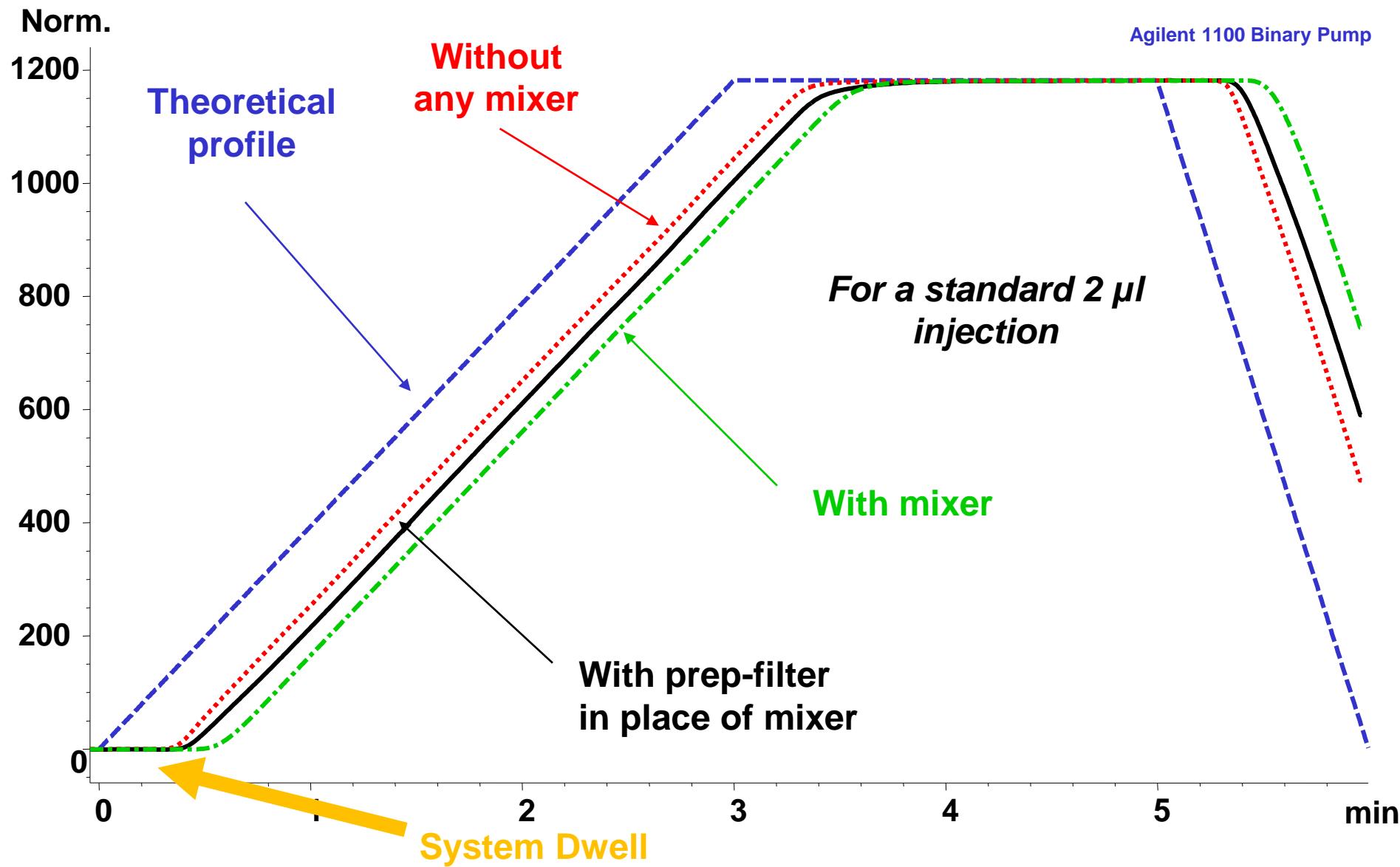




## Quaternary Pump Configuration: Low Pressure Mixing



## Gradient Cycles: What Really Happens With Dwell Volume





# Instrument Data Can Vary Depending Upon Configuration

Instrument	Typical Dwell Volume ( $\mu\text{L}$ )
Agilent HP1100 Binary	180-900
Agilent HP1100 Quaternary	800-1100
Agilent 1200 RRLC	~300
Dionex P680A Quaternary	<400
Thermoquest P4000 Quat	<600
Waters Alliance 2695 Quat	600
Waters Varian 9012 Ternary	1000
Waters Acquity UPLC	~100

	ACQUITY H-Class	1290B (Standard)	1290B (Optimised)	1290Q	Chromaster UltraRs	1200RR	1100 LC125	1200MD	1100 LC94	1100 LC98
ECV ( $\mu\text{L}$ )	21.4	13.6	8.3	21.3	18.3	18.7	23.9	29.8	62.0	102.5
Bandwidth	9.96	9.82	7.40	12.78	10.94	11.45	15.48	14.28	25.62	42.87
Dwell volume ( $\mu\text{L}$ )	372	198	198	950	262	740	1082	1110	1104	1100

**See ACE Knowledge Note #0001 for how to measure system dwell volume**



## Instrument to Instrument Same Method Transfers

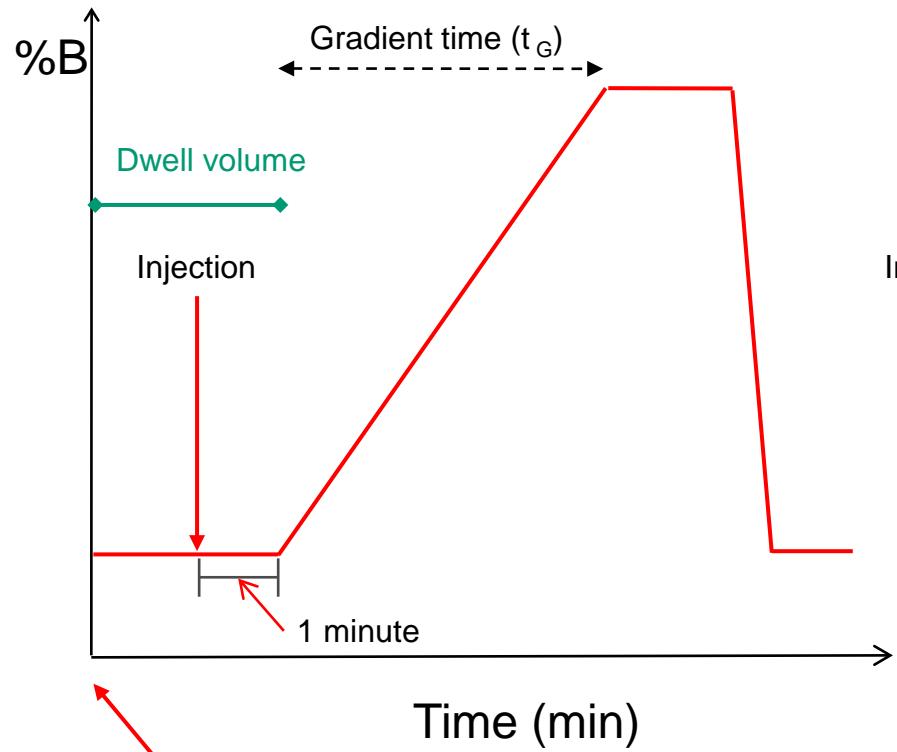
- ♦ For gradient method transfer,  $t_G$ , F and  $V_M$  all remain identical.
- ♦ Only need to correct for the system dwell volume
  - Given that  $V_M$  does not change for method transfer, the  $V_D/V_M$  equation simplifies to:

$$x = \left( \frac{V_{D1} - V_{D2}}{F} \right)$$

- Negative value: injection must be delayed x minutes after gradient starts.
- Positive value: a pre-gradient isocratic hold of x minutes should be added to gradient program.



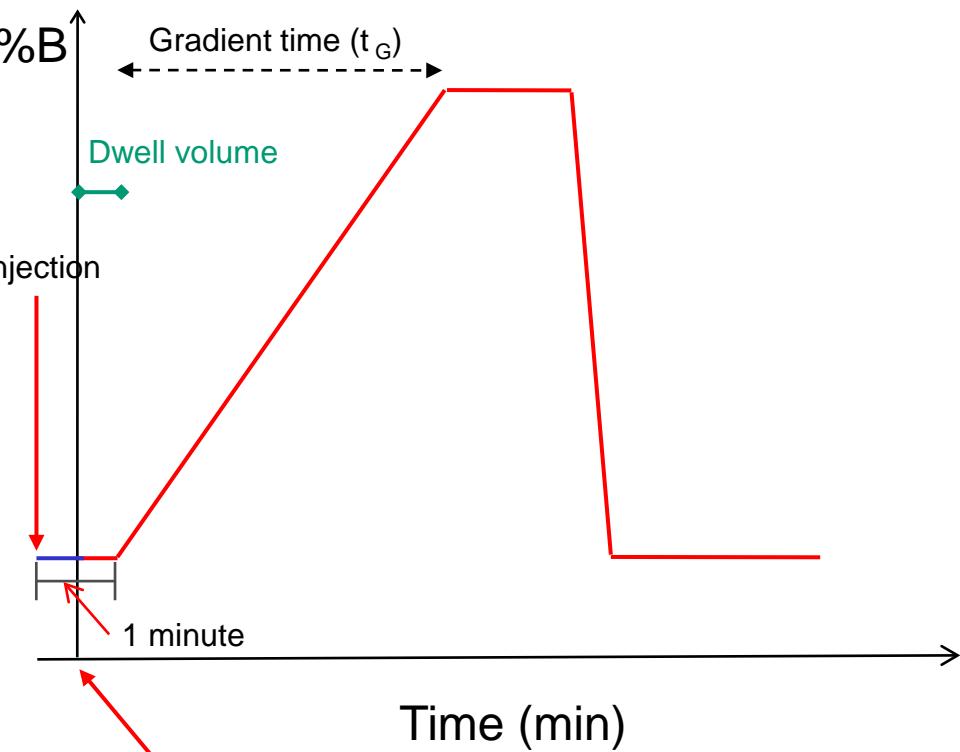
## Method Translations: Dwell Volumes / Injection Times



$$\Delta (V_D/V_M) = \text{negative}$$

e.g. small column on a high dwell volume system. The dwell time is artificially long.

Injection is delayed until after the start of gradient

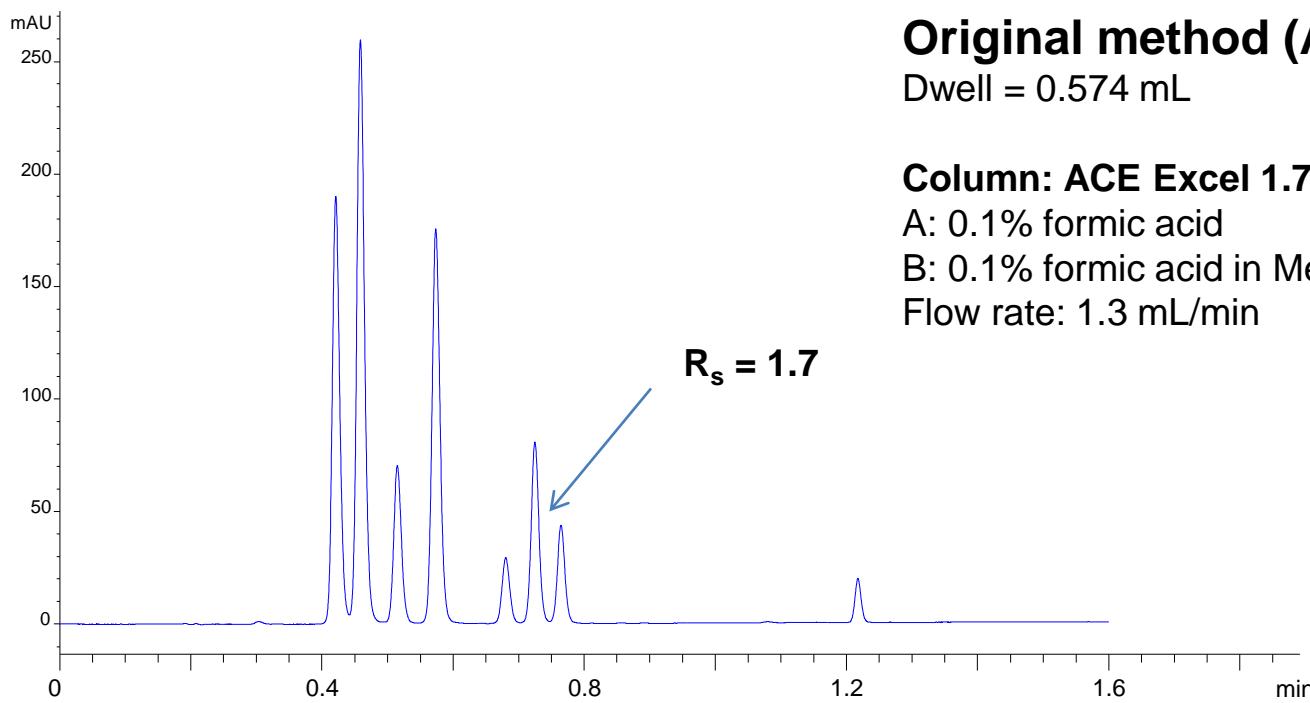


$$\Delta (V_D/V_M) = \text{positive}$$

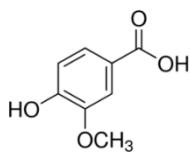
e.g. large column on a low dwell volume system. The dwell time is artificially short.

Pre-gradient hold is added. Effectively extends dwell time

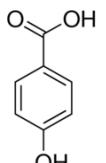
# UHPLC Method Transfer - Vanillins



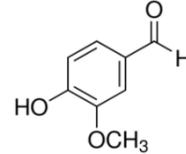
**t<sub>G</sub>** 1.32 min  
**Post time:** 2 min  
**Inj. Vol.** 1  $\mu$ L  
**P<sub>MAX</sub>** 502 bar



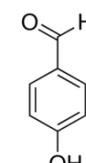
1. vanillic acid



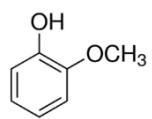
2. 4-hydroxybenzoic acid



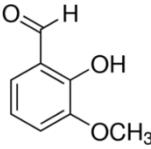
3. vanillin



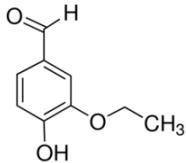
4. 4-hydroxybenzaldehyde



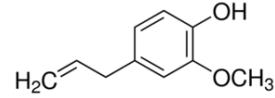
5. guaiacol



6. o-vanillin



7. ethyl vanillin



8. eugenol

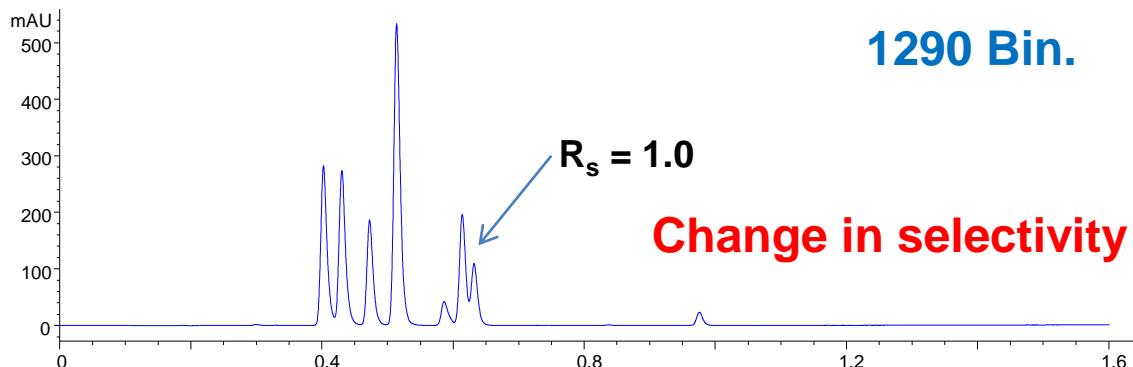
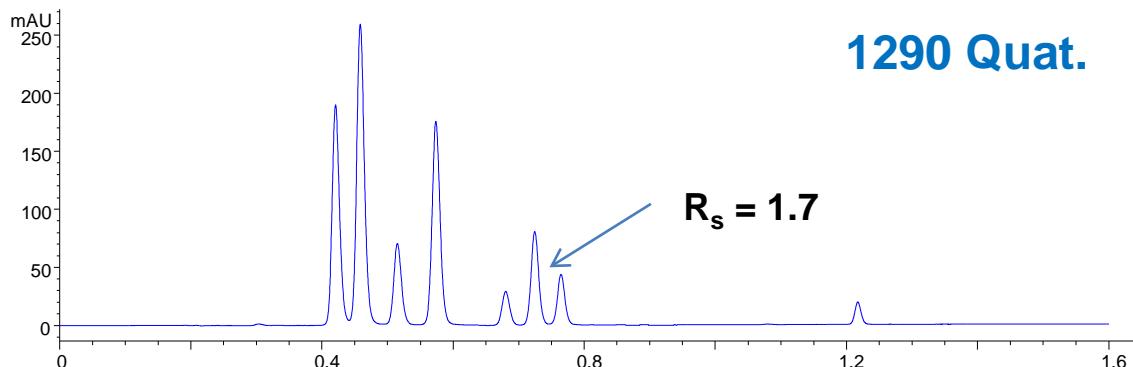
## UHPLC Method Transfer - Vanillins

### ◆ Aim

- To translate the method from Agilent 1290 Quaternary to Agilent 1290 binary
- Need to correct for system dwell volume only
- $V_{D1} - V_{D2} = \Delta = 0.574 - 0.202 = 0.372 \text{ mL}$
- Positive value, therefore a pre-gradient hold ( $x$ ) is required
- A negative value would require a delayed injection
- $x = \frac{|\Delta|}{F_2} = 0.29 \text{ min}$  pre-gradient hold



## Instrument to Instrument Method Transfer



t	%B
0	25
1.32	75
1.49	75
1.6	25

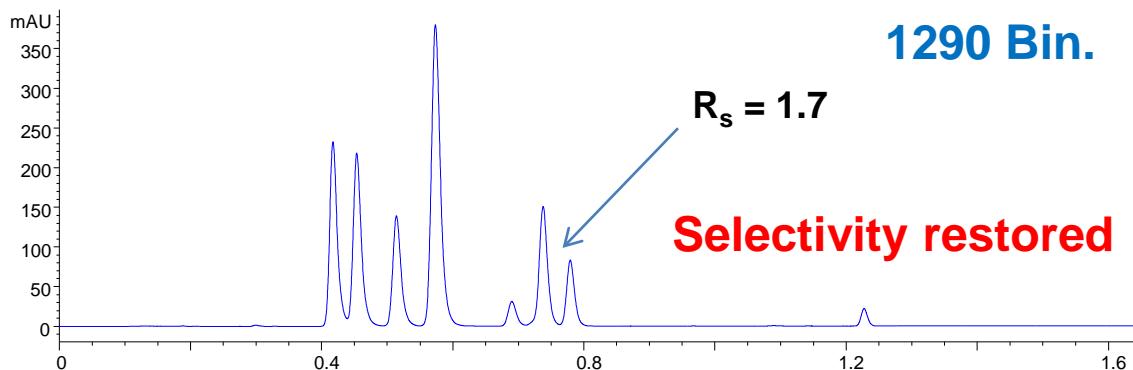
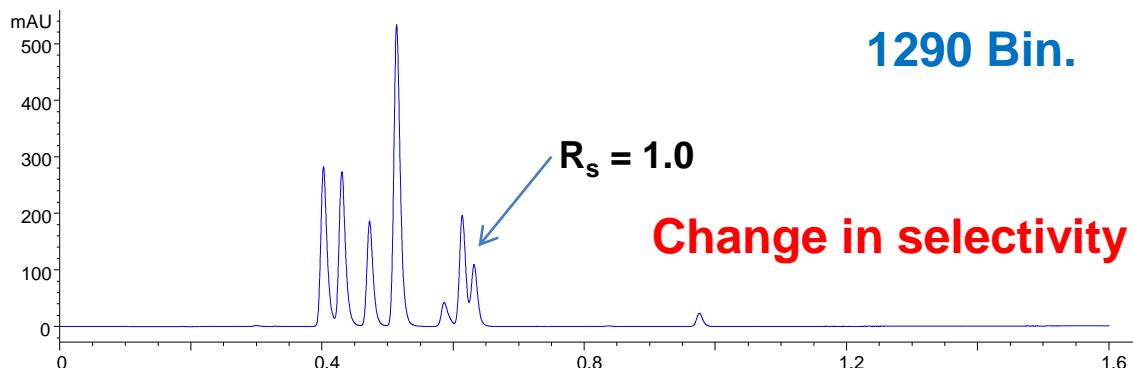
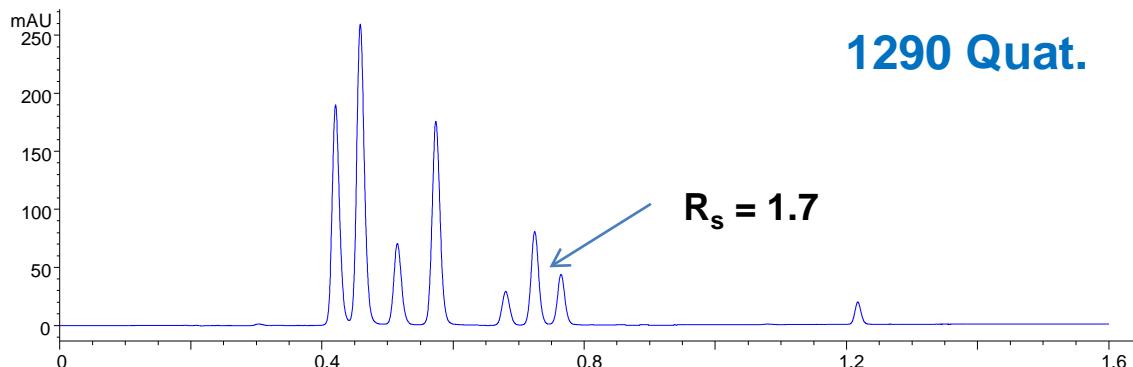


**Method transfer – no correction for system dwell**

t	%B
0	25
1.32	75
1.49	75
1.6	25



## Instrument to Instrument Method Transfer



t	%B
0	25
1.32	75
1.49	75
1.6	25



**Method transfer – no correction for system dwell**

t	%B
0	25
1.32	75
1.49	75
1.6	25



**Method transfer – Corrected for  $\Delta V_D$**

t	%B
0	25
0.29	25
1.61	75
1.78	75
1.89	25

# Method Transfer Tool Example - Gradient

V1.3

## Method Transfer

Method transfer involves moving a method from one LC instrument to another, whilst keeping the column format constant. Isocratic analysis is simple with no method changes required. Gradient methods however, should be adjusted to account for any change in system dwell volume in order to ensure accurate method transfer. This tool automatically determines any correction required to ease method transfer. See the Dwell Volume tab for details of how to determine system dwell volume.

**Column Information**

Column Length (L)	50 mm
Column i.d. (d <sub>c</sub> )	3.0 mm
Particle Diameter (d <sub>p</sub> )	1.7 $\mu$ m
L/d <sub>p</sub>	29412
Column Porosity	0.63 <small>What's This?</small>
Column Volume (V <sub>M</sub> )	0.223 mL

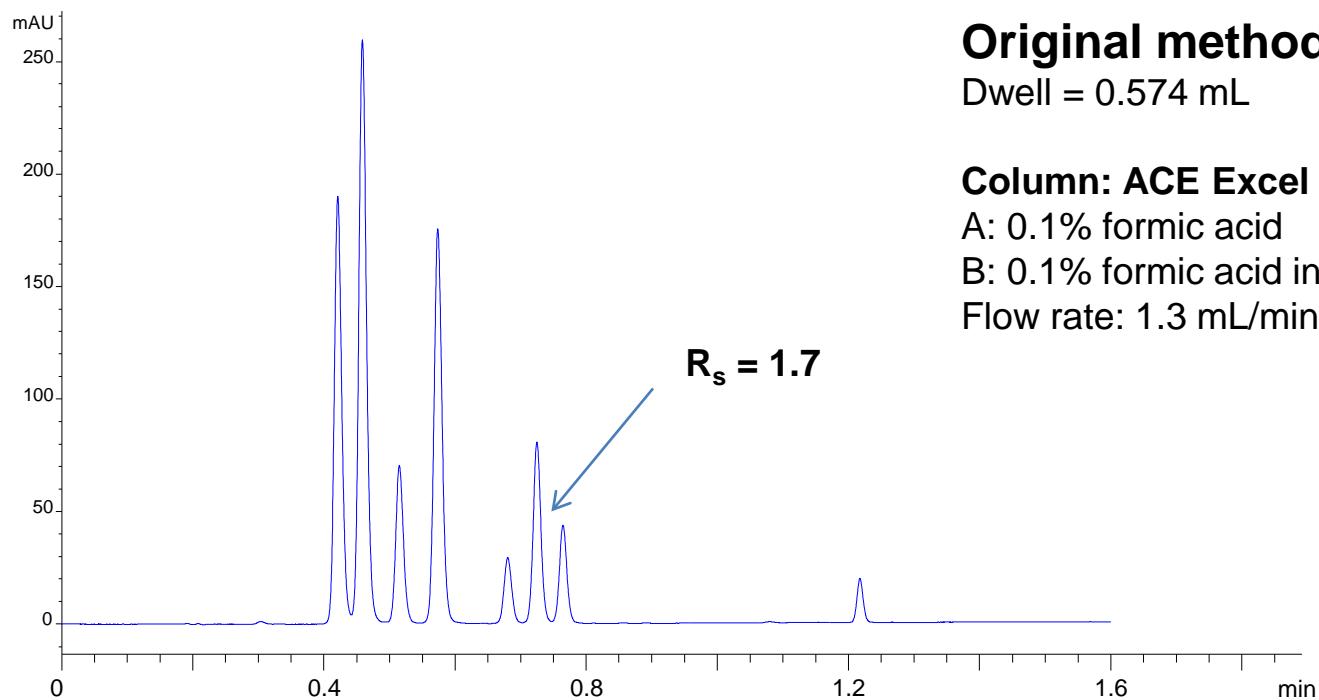
**Method**

Current		Transferred																											
Flow Rate	1.30 mL/min	LC Name	UHPLC 3																										
Dwell Volume (V <sub>D</sub> )	0.574 mL	Dwell Volume (V <sub>D</sub> )	0.202 mL																										
Gradient	<table border="1"><thead><tr><th>Time</th><th>%B</th></tr></thead><tbody><tr><td>0.00</td><td>25.0</td></tr><tr><td>1.32</td><td>75.0</td></tr><tr><td>1.49</td><td>75.0</td></tr><tr><td>1.60</td><td>25.0</td></tr><tr><td>3.60</td><td>25.0</td></tr></tbody></table>	Time	%B	0.00	25.0	1.32	75.0	1.49	75.0	1.60	25.0	3.60	25.0	Gradient	<table border="1"><thead><tr><th>Time</th><th>%B</th></tr></thead><tbody><tr><td>0.00</td><td>25.0</td></tr><tr><td>0.29</td><td>25.0</td></tr><tr><td>1.61</td><td>75.0</td></tr><tr><td>1.78</td><td>75.0</td></tr><tr><td>1.89</td><td>25.0</td></tr><tr><td>3.89</td><td>25.0</td></tr></tbody></table>	Time	%B	0.00	25.0	0.29	25.0	1.61	75.0	1.78	75.0	1.89	25.0	3.89	25.0
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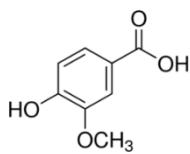
*New Gradient with isocratic hold*

## Transferring Between Instruments Part II

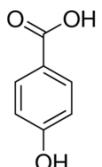
# UHPLC Method Transfer - Vanillins



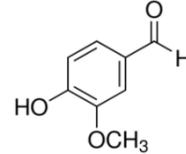
$t_G$  1.32 min  
 Post time: 2 min  
 Inj. Vol. 1  $\mu\text{L}$   
 $P_{MAX}$  502 bar



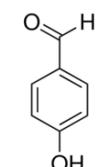
1. vanillic acid



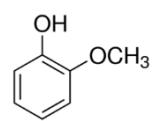
2. 4-hydroxybenzoic acid



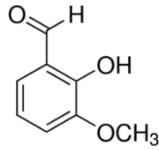
3. vanillin



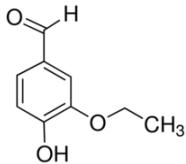
4. 4-hydroxybenzaldehyde



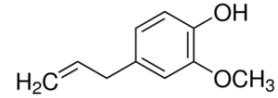
5. guaiacol



6. o-vanillin



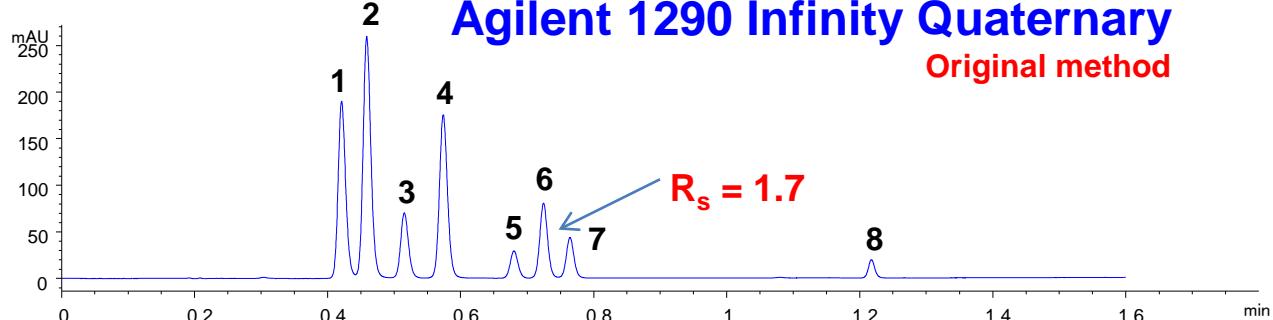
7. ethyl vanillin



8. eugenol

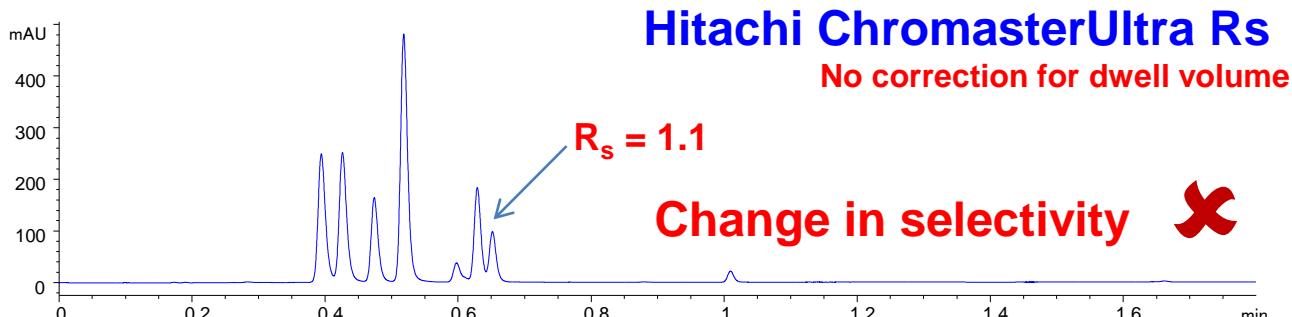


## Instrument to Instrument Method Transfer



t	%B
0	25
1.32	75
1.49	75
1.6	25

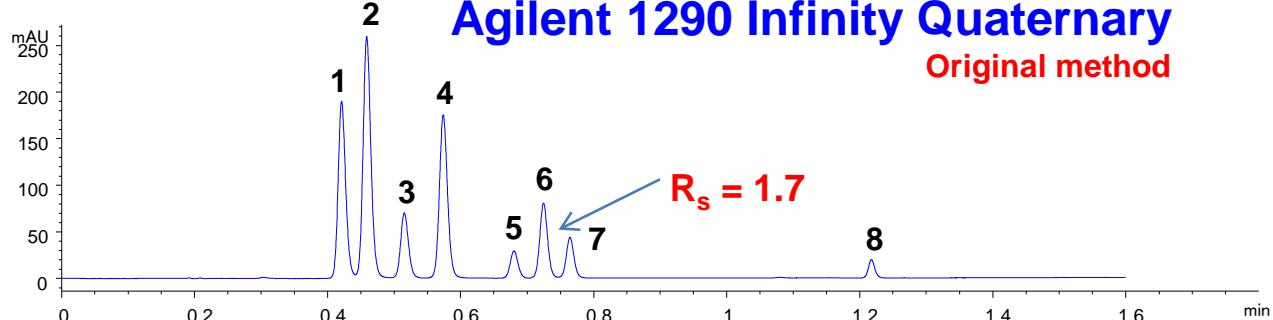
no correction for system dwell



t	%B
0	25
1.32	75
1.49	75
1.6	25

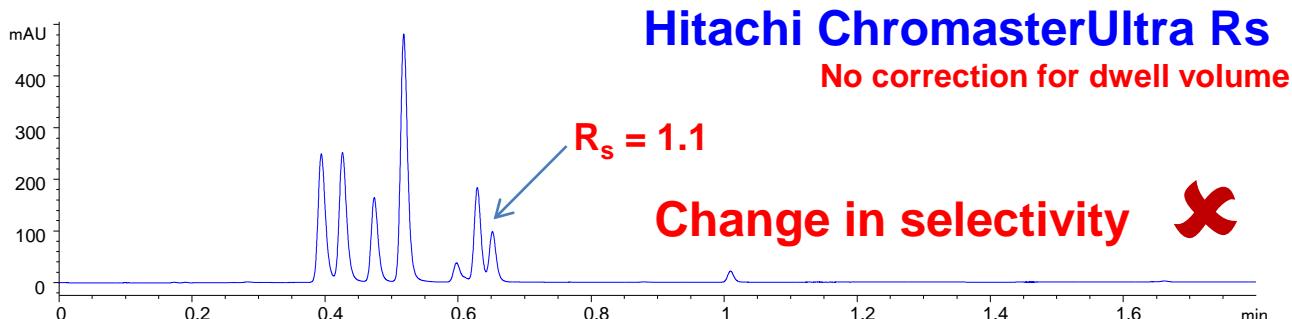


## Instrument to Instrument Method Transfer



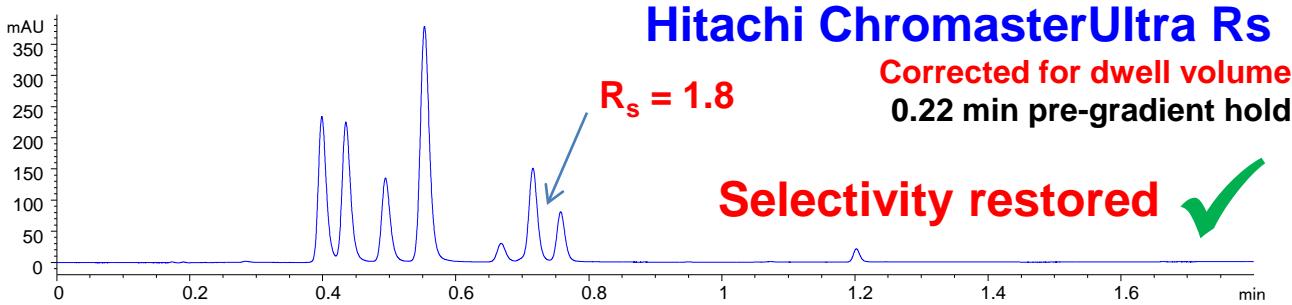
t	%B
0	25
1.32	75
1.49	75
1.6	25

no correction for system dwell



t	%B
0	25
1.32	75
1.49	75
1.6	25

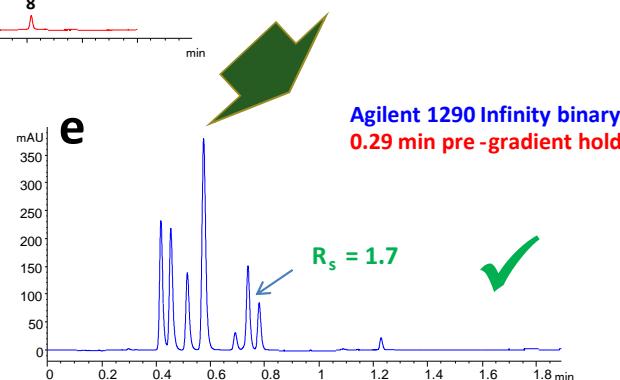
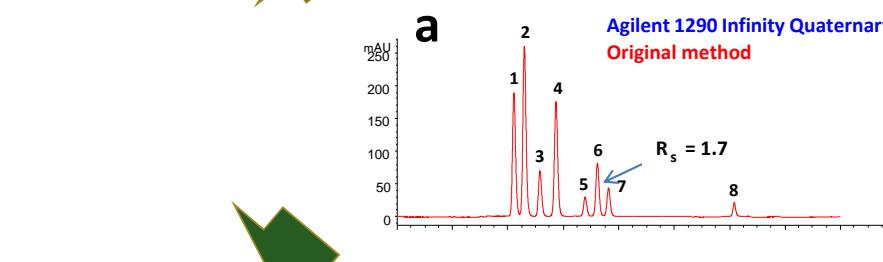
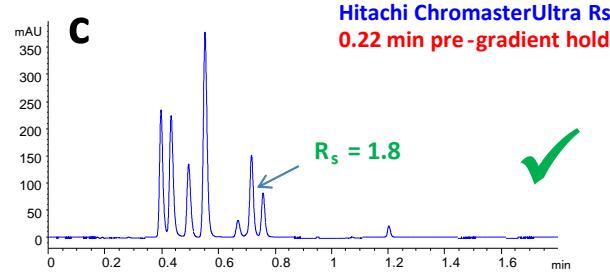
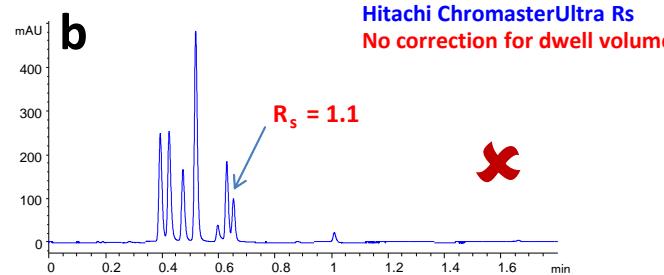
Method corrected for  $\Delta V_D$



t	%B
0	25
0.22	25
1.54	75
1.71	75
1.82	25



## Transferring Methods - Success or Not?



## Summary

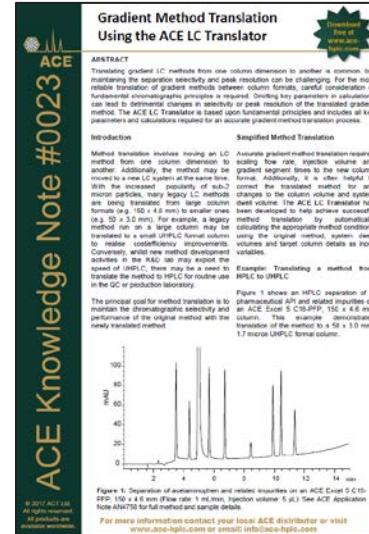
- ◆ Relatively accurate method translations are achievable
- ◆ Isocratic translations
  - ◆ Use L/dp ratio translations
  - ◆ Translate to columns that have scalable bonded phases
  - ◆ Be aware of system dispersion effects on smaller columns
- ◆ Gradient translations
  - ◆ More complex but easy calculations described
  - ◆ Consider  $V_D/V_m$  impact upon translation accuracy ← few online calculators currently offer this
- ◆ High peak capacities are possible for complex samples using column coupling
- ◆ Instruments transfers are simple for isocratic methods but gradient methods must consider dwell volumes

# Useful Resources

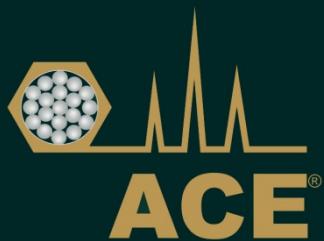
- ◆ ACE Translation Tool:
    - ◆ (+help file)
    - ◆ (+ AKN#0023)



- ◆ ACE Knowledge Notes (AKNs):
    - ◆ **AKN0001** - How to Determine System Dwell Volume
    - ◆ **AKN0006** – UHPLC Column Connections
    - ◆ **AKN0011** – Practical UHPLC
    - ◆ **AKN0012** - Understanding the Relationship between Particle Size, Performance and Pressure
    - ◆ **AKN0017** - How to Determine Extra Column Dispersion and Extra Column Volume
    - ◆ **AKN0023** - Gradient Method Translation Using the ACE LC Translator



Download Resources at [mac-mod.com](http://mac-mod.com)  
or contact [info@mac-mod.com](mailto:info@mac-mod.com)



Thank You For Your Attention

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