# Enhanced Selectivity using Various Fused-Core® Bonded Phases to Enable Faster Method Development Stephanie A. Schuster<sup>1</sup>; Barry E. Boyes<sup>1</sup>; William L. Johnson<sup>1</sup>; and Thomas J. Waeghe<sup>2</sup>

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Instrument

Agilent 1100 quaternary

• 3.0 mL heat exchanger

(1 < V<sub>cell</sub> < 5 µL)

Analytical Columns

• 3.0 x 50 mm HALO

– RP-Amide

Phenyl-Hexyl

– C18

– PFP

• Shortest length 0.005" ID

tubing between modules

#### Abstract

HPLC columns of Fused-Core® superficially porous particles with an overall diameter of 2.7 µm and a 0.5 µm porous shell have been shown by many users to possess unusual efficiency and stability, allowing rapid separations with the needed ruggedness of operation of 5 µm particle columns. These 2.7 µm Fused-Core particles permit separation speeds competitive with sub-2-µm totally porous particles, but with 40–50% of the column back pressure. The general strategy for most RPLC method development has been to use a C18 bonded phase first, then look for alternatives if the C18 cannot provide the required resolution. Correct bonded phase selection at the start of the method development would result in faster method development since time constraints do not always permit the opportunity for quality by design experimentation. This presentation will include the Snyder/Dolan selectivity parameters of the various bonded phases that are available on Fused-Core particles. The "orthogonality" S values calculated using a set of acids, bases, neutrals, and zwitterions reveal several generalities that can be made. The PFP (pentafluorophenylpropyl) phase is more retentive for electron-rich compounds, and can show enhanced shape selectivity due to its electronegative aromatic ring, while Phenyl-Hexyl is more retentive for electron-poor compounds due to  $\pi$ - $\pi$  interactions with a flexible aromatic ring. In the case of the RP-Amide phase, polar compounds are better retained resulting from dipole interactions compared to the C18 bonded phase—especially those with –OH substituents. A new cyano bonded phase shows advantages for the separation of basic compounds. These generalizations, when combined with a defined strategy for column selection, facilitate rapid method development using the diverse bonded-phase selectivities of Fused-Core columns.

# **HALO Bonded Phase Characteristics**

HALO Phase	Retention Mechanism	Retention Increased for	Best Application	
C18, C8	Hydrophobic interactions	Lipophilic molecules, uncharged acids and bases, strong bases or acids in ion pairing mode	Analytes differing in hydrophobicity, homologues non-aqueous RPLC	
RP-Amide	Hydrophobic, hydrogen bonding	Alcohols, acids, phenols	basic analytes, heterocycles, proton donors and acceptors, highly aqueous conditions	
Phenyl-Hexyl	Hydrophobic, p–p	Electron-poor compounds, analytes with electron- withdrawing groups, (ketones, nitriles, alkenes, etc)	heterocycles, aromatics, highly aqueous conditions	
PFP	Hydrophobic, p–p, hydrogen bonding, dipole-dipole	Electron-rich compounds, analytes with p bonds, electron delocalization and electron- donating groups, proton donors, analytes with different dipole moments	Bases, stereoisomers, steroids, taxanes, substituted aromatics, highly aqueous conditions HILIC separations ≥ 80% ACN	
ES-CN	Hydrophobic, dipole-dipole	Polar molecules polarity of cyano phase requires 10-20% less organic for retention comparable to other phases, but produces different selectivity	Mixtures of polar and non-pola analytes, explosives, pesticide HILIC separations ≥ 80% ACN	

# **Fused-Core Particles**

#### **Particle Characteristics**

- Silica: High purity, Type B
- Pore Size: 90 Å and 160 Å
- Particle Size Distribution: 5% RSD
- pH range: 2–9
- Efficiency: 230,000 plates/m



#### **Features and Benefits**

- Ultrafast separations save time and improve productivity
- UHPLC performance without the need for UHPLC equipment
- Low pressures enable the coupling of columns for high efficiency/high resolution



### Some Suggested **Method Development Strategies**

#### **Isocratic Method**

- 1. Ensure instrument extracolumn volume is 20 µL or less
- 5 µL flow cell, 0.005" ID tubing
- Data Rate > 10 Hz, Response time: £ 0.2 sec 2. Choose 4.6 x 50 mm HALO C18 column.
- 3. Select pH of aqueous component and buffer.
- 4. Start with mobile phase having highest ratio of organic modifier to buffer permitted by
- 5. Set flow rate at 1.5 to 2.0 mL/min 6. Prepare sample in 50:50 organic
- modifier/buffer or aqueous.
- Inject 2 µL. 8. Ensure that all analytes elute at highest %
- organic
- Decrease % organic successively until last component elutes at  $\sim k = 10-15$ .
- 10. Adjust sample solvent composition to keep % organic £ mobile phase % organic.
- 11. Compare separation at same % organic with HALO RP-Amide, Phenyl-Hexyl and PFP phases
- 12. Select phase with best peak shape and band
- 13. Optimize % organic, column temp., pH.

#### **Gradient Method**

- 1. Ensure instrument extracolumn volume is 10–20 µL
- 2–5 µL flow cell, 0.005" ID tubing • Data Rate > 10 Hz, Response time: £ 0.1 sec
- 2. Choose 4.6 x 50 mm\* HALO C18 column.
- 3. Select pH of aqueous component and buffer. 4. Set flow rate at 1.5 to 2.0 mL/min
- 5. Prepare sample in minimum ratio of organic modifier/aqueous possible.
- 6. Inject 2 µL. 7. Run gradient from 5–100% organic in 10 min. 8. Compare 3 x 50 mm HALO RP-Amide.
- Phenyl-Hexyl, and PFP phases using same gradient profile.
- 9. Inject 2 µL. 10. Select best phase for further optimization
- based on band spacing, peak shape, run time. 11. Optimize starting and ending % organic,
- gradient time, column temp., pl 12. If more resolution is needed, increase column
- length as required.

\*For those with UHPLC systems and low dispersion volumes, 3.0 x 50 or 2.1 x 50 mm columns are recommended for solvent reduction and faster equilibration.

# **HALO Fused-Core Bonded Phases**



(pentafluorophenylpropyl)

# **Selectivity and Orthogonal Separations**

 Selectivity is the most influential parameter in the resolution equation





- Orthogonal
- Marked changes in relative retention so that peaks which are unresolved in one (chromatogram) are likely to be separated in the second chromatogram
- Orthogonal separations are conducted on columns with significantly different selectivities



extra stable-cyanopropyl)

# Experimental

#### **Column Screening Conditions**

Gradients: 5-95%B in 5 min.

- ACN
- 10 mM ammonium formate. pH 3.0
- Semi-micro flow cell, bypassed Temperature: 40 C
  - Flow rate: 0.85 mL/min
  - Detection: UV @ 254 nm
  - Injection volume: 2 µL

#### Analytes

Set represents 23 compounds, specifically pharmaceuticals, including acids, bases, zwitterions, and neutrals.





# HALO C18 vs. Phenyl-Hexyl: Organic Acids



## HALO C18 vs. ES-CN: Mixed EPA Explosive Standards 8330 A and B



### "Orthogonality" S Values for HALO Bonded Phases

$$S = 100 \text{ '} \sqrt{1 - R^2}$$

where  $R^2$  is the correlation coefficient of a graph of  $t_r$  (phase 1) vs.  $t_r$  (phase 2) or log k' (phase 1) vs. log k' (phase 2)

	AMT S values	Snyder/Dolan S values
C18 vs. Phenyl-Hexyl	9	18
C18 vs. RP-Amide	14	26
RP-Amide vs. Phenyl-Hexyl	19	35
C18 vs. PFP	63	68

The S values are dependent upon the selected mobile phase and compound set used for screening.

**AMT Buffer:** 10 mM Ammonium Formate pH 3

AMT Compound Set: 2-fluorobenzoic acid, 3-cyanobenzoic acid, 3-indoleacetic acid, 3-nitrobenzoic acid, 4-aminobenzoic acid, benzoic acid, beta-estradiol, biochanin A, chloramphenicol, cortisone, fenoprofen, ibuprofen, ketoprofer mefenamic acid, naringin, norfloxacin, nortriptyline hydrochloride, prednisolone, prednisone, procainamide hydrochloride, prunetin, ranitidine, sulfamerazine

Snyder/Dolan Buffer: 30 mM Potassium Phosphate pH 2.8

Snyder/Dolan Compound Set: 5,5-diphenylhydantoin, 5-phenylpentanol, acetophenone, amitriptyline, anisole, benzonitrile, berberine, cis-chalcone, ethylbenzene, mefenamic acid, N,N-diethylacetamide, N,N-dimethylacetamide, n-butylbenzoic acid, nortriptyline, p-nitrophenol, thiourea, toluene, trans-chalcone

Column	н	S*	A	В	C <sub>2.8</sub>	C <sub>7.0</sub>
HALO C18	1.107	0.048	0.006	-0.050	0.056	0.040
HALO Phenyl-Hexyl	0.789	-0.094	-0.233	-0.006	0.101	0.456
HALO RP-Amide	0.859	0.080	-0.384	0.190	-0.417	0.312
HALO PFP	0.702	-0.117	-0.073	-0.062	1.168	0.972

F<sub>s</sub> = column selectivity comparison function, based on differences in H, S\*, A, B and C for two columns, where the larger the F<sub>s</sub> value, the greater the difference in selectivity.

Column	F <sub>s</sub>		
HALO C18	0.00		
HALO Phenyl-Hexyl	17.8		
HALO RP-Amide	53.9		
HALO PFP	92.7		

# **Snyder/Dolan Values for HALO Phases**

# **Benzodiazepines on HALO Fused-Core Bonded Phases**

# HALO C18 vs. PFP: Steroids



### **Benzoic Acids on HALO Fused-Core Bonded Phases**



# Summary

- Presented differences in selectivity among various HALO phases.
- Neue "S" values for orthogonality
- Snyder-Dolan Hydrophobic Subtraction Model values
- Demonstrated examples of selectivity differences among HALO phases for variety of analytes.
- Offered example method development strategies for isocratic and gradient methods.
- Recommended more elaborate method development schemes using several different pHs, temperatures, etc. for difficult samples, including use of computer simulation such as DryLab<sup>®</sup> 2010.
- Summarized HALO stationary characteristics and offered suggestions for best application of each phase.

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