

Modifying Separation Selectivity by Selection of Fused-Core® Particle Bonded Phases

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Abstract

Fused-Core® particles, with an overall diameter of 2.7 μm and a porous 0.5 μm thick shell, have demonstrated efficiencies comparable to those of sub-2- μm particles, but with significant lower column back pressure. Column efficiency (N) is an important parameter in the general resolution equation, yet resolution only increases by the square root of an increase in efficiency.

Selectivity (α , α), on the other hand, has a direct linear relationship with resolution, and, thus, has a much more powerful influence on resolution. While the majority of applications developed today utilize C18 bonded phase chemistry, there are frequent instances when an alternative bonded phase yields superior results in terms of selectivity and, ultimately, resolution.

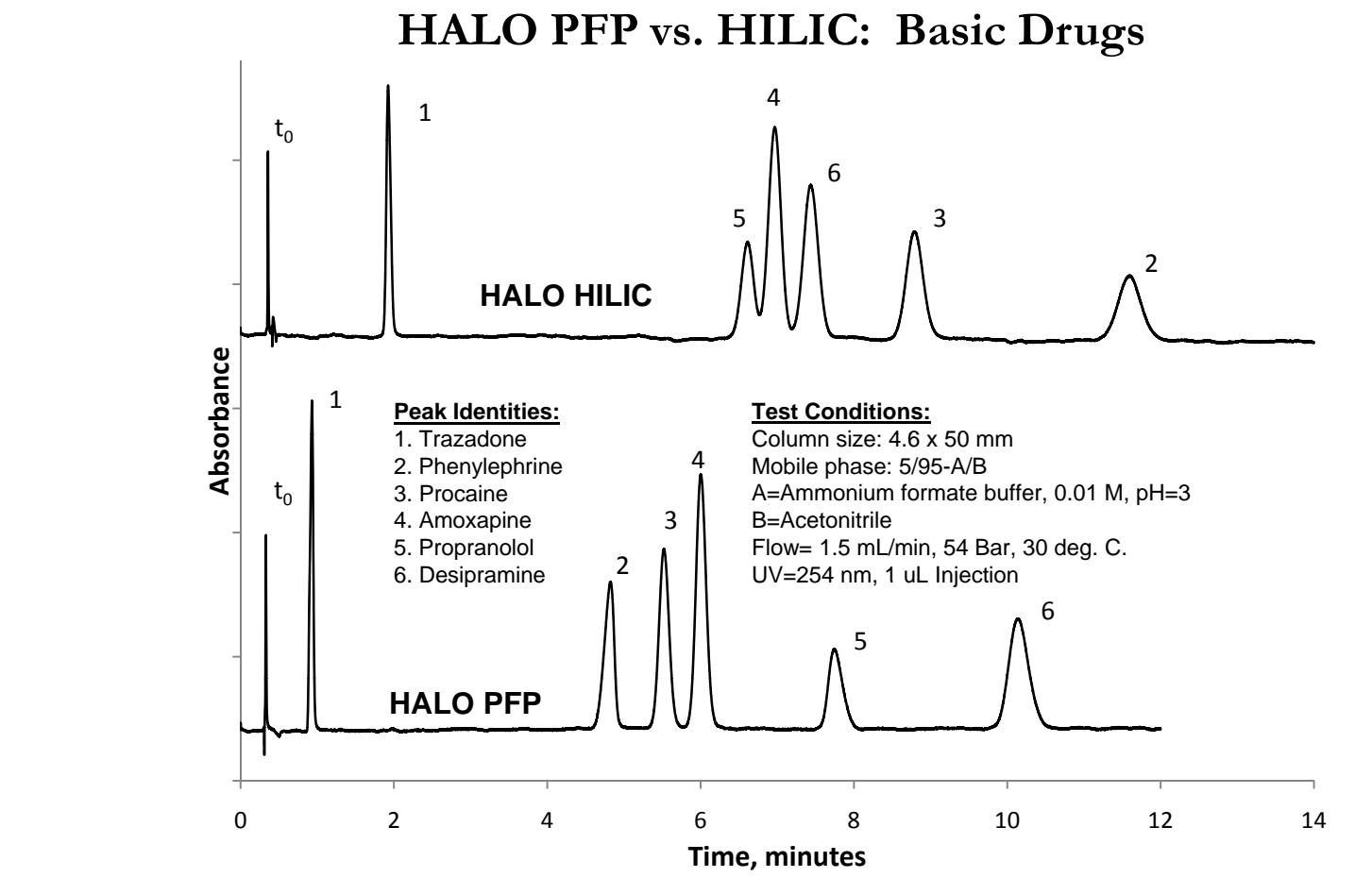
Illustrative chromatograms using stationary phases with different selectivities, while maintaining the high plate efficiencies generated by the Fused-core particles, will demonstrate the effectiveness of such an approach for obtaining optimized HPLC separations. The value of using multiple column selectivities as part of a fast method development strategy using DryLab® 2010 software with Peak Match® will also be highlighted.

Most Effective Parameters to Change Selectivity

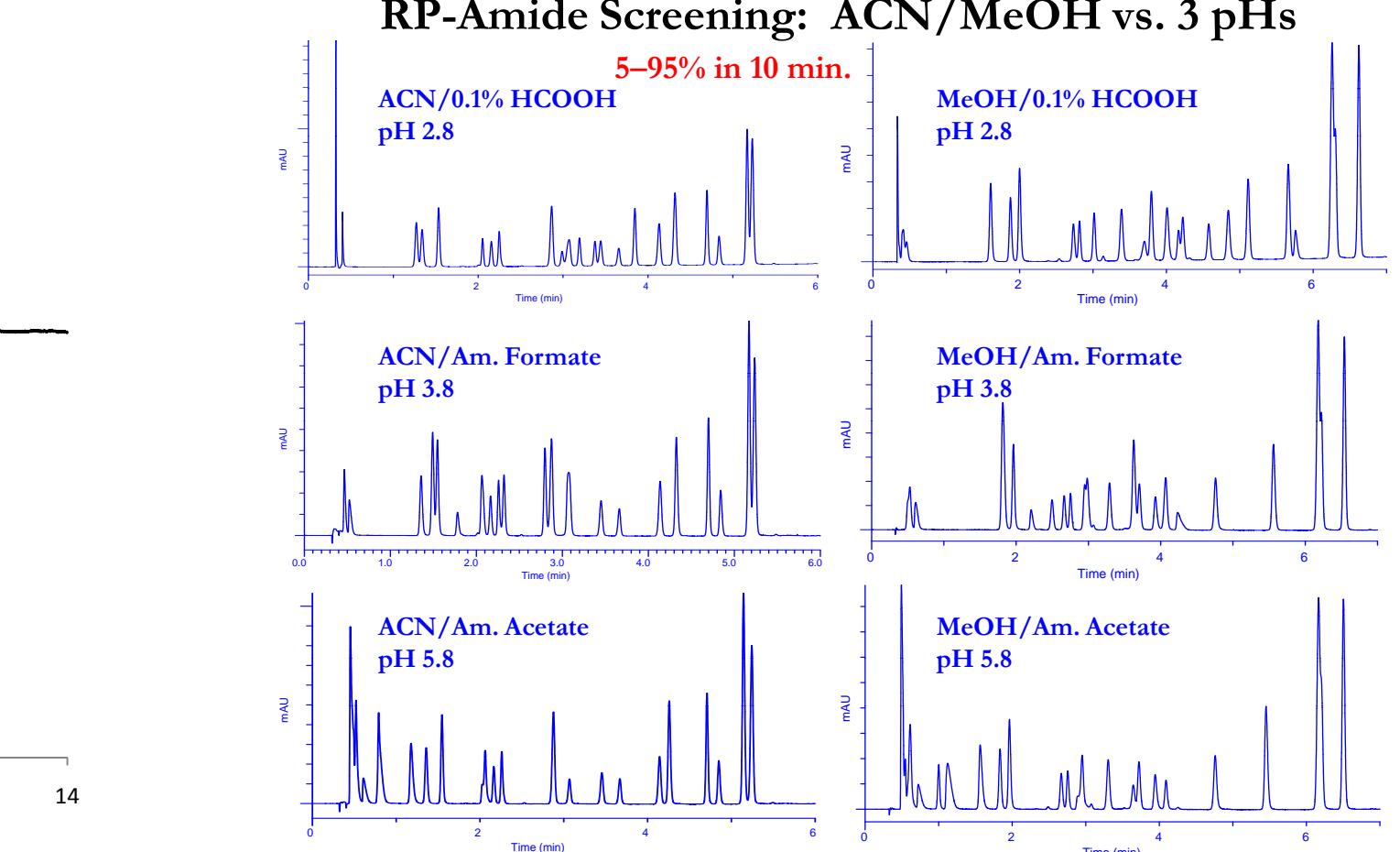
The analysis condition parameters that most affect selectivity, α are:

- Column type (C18, phenyl, amide, etc.) ++
- B-solvent (acetonitrile, methanol, etc.) ++
- Mobile phase pH ++
- Ion-pair concentration ++
- %B solvent/gradient steepness +
- Column temperature +
- Buffer concentration +

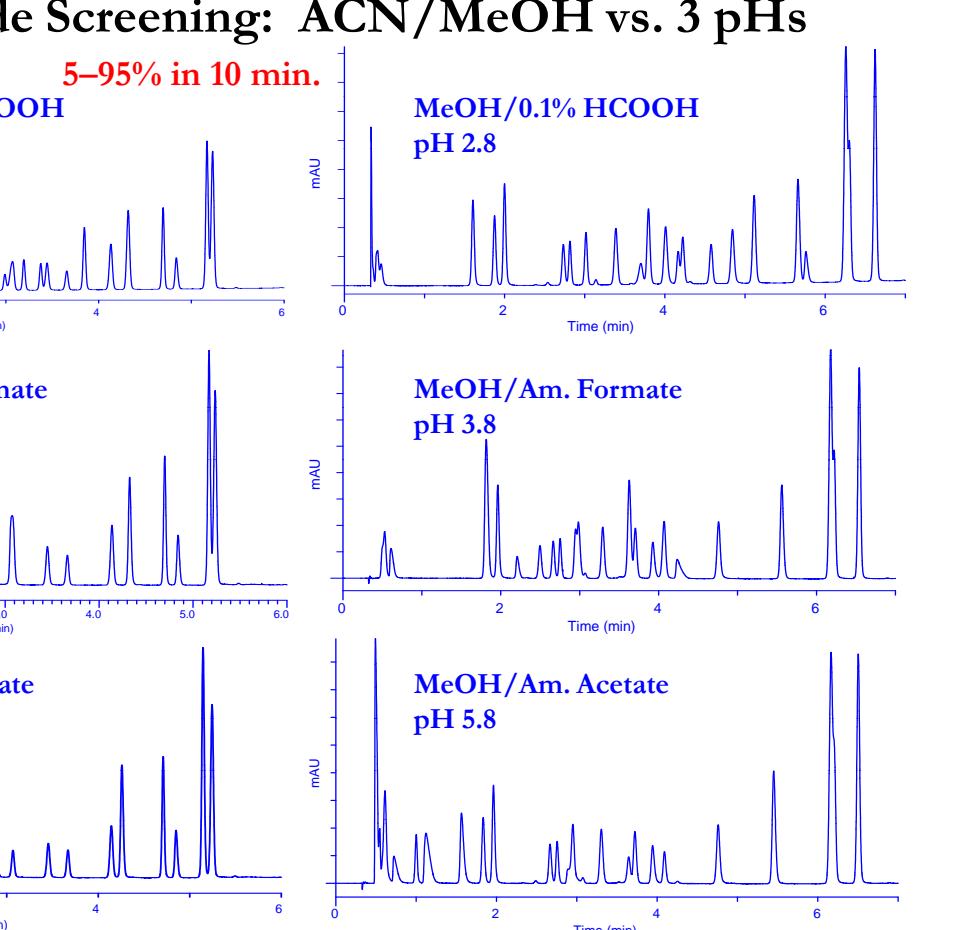
HALO C18 vs. RP-Amide: Polar Analytes



HALO PFP vs. HILIC: Basic Drugs



RP-Amide Screening: ACN/MeOH vs. 3 pHs



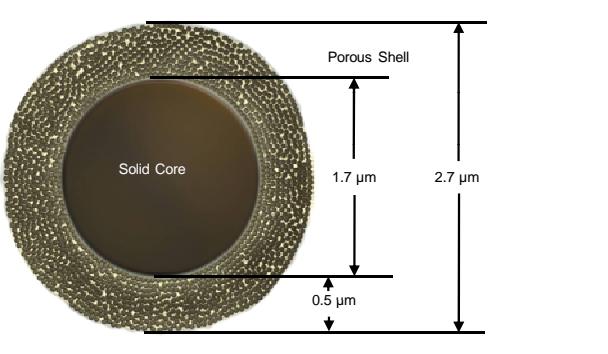
HALO Bonded Phase Characteristics

HALO Phase	Retention Mechanism	Retention Increased for	Best Application
C18, C8	Hydrophobic interactions	Lipophilic molecules, uncharged acids and bases, strong bases or acids in ion pairing mode	Analyses differing in hydrophobicity, homologues non-aqueous RPLC
RP-Amide	Hydrophobic, hydrogen bonding	Alcohols, acids, phenols	basic analyses, heterocycles, proton donors and acceptors, highly aqueous conditions
Phenyl-Hexyl	Hydrophobic, $\pi-\pi$	Electron-poor compounds, heterocycles, aromatics, highly aqueous conditions	ketones, nitriles, alkenes, etc)
PFP	Hydrophobic, $\pi-\pi$, hydrogen bonding, dipole-dipole	Electron-rich compounds, steriods, taxanes, substituted aromatics, delocalization groups, proton donors, analays with different dipole moments	bases, stereoisomers, highly aqueous conditions, HILIC separations ≥ 80% ACN
Silica/HILIC	NPLC: analytic adsorption on silica, displacement by solvent more HILIC: partitioning of polar analytes between highly aqueous mobile phase and water layer near silica surface	NPLC: polar vs. nonpolar analytes, planar vs. nonplanar	HILIC: polar acids, bases, and neutrals

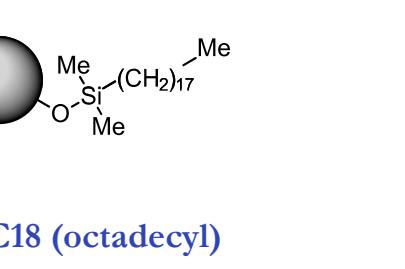
Fused-Core Particles

Particle Characteristics

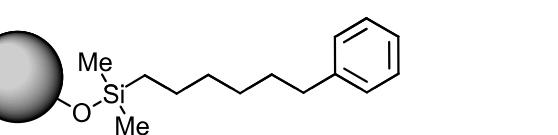
- Silica: High purity, Type B
- Pore Size: 90 Å and 160 Å
- Particle Size Distribution: 5 % RSD
- pH range: 2-9
- Efficiency: 230,000 plates/m



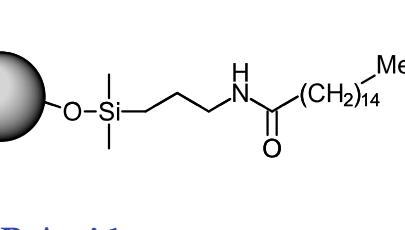
HALO Fused-Core Bonded Phases



C18 (octadecyl)



Phenyl-Hexyl



RP-Amide

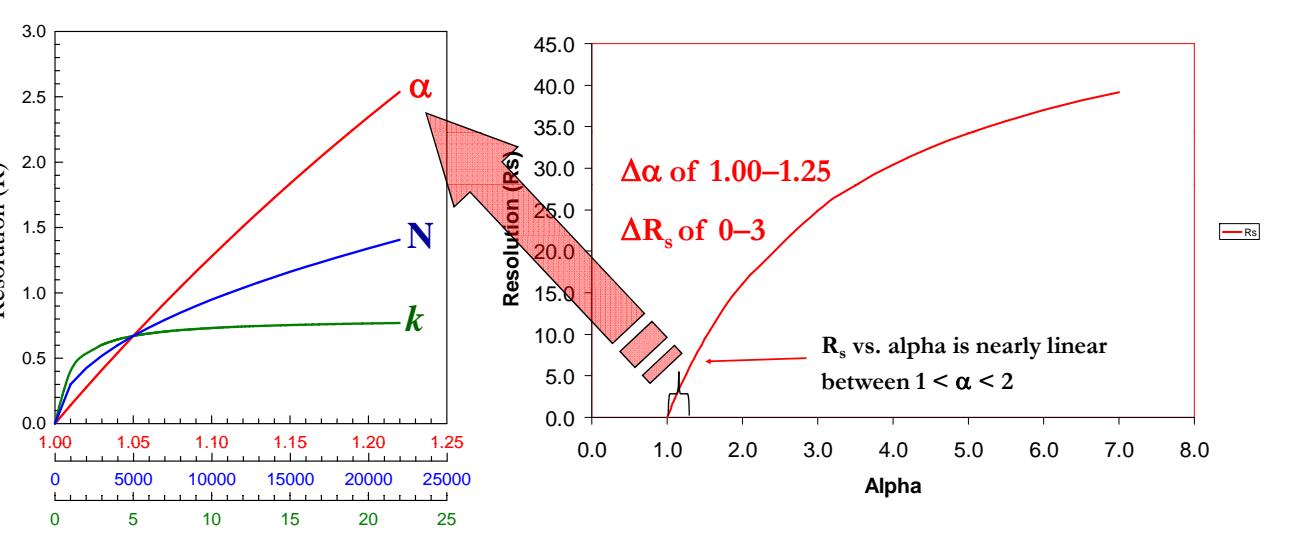


PFP (pentafluorophenylpropyl)

Resolution Equation Shows that Selectivity is Most Effective Parameter to Change

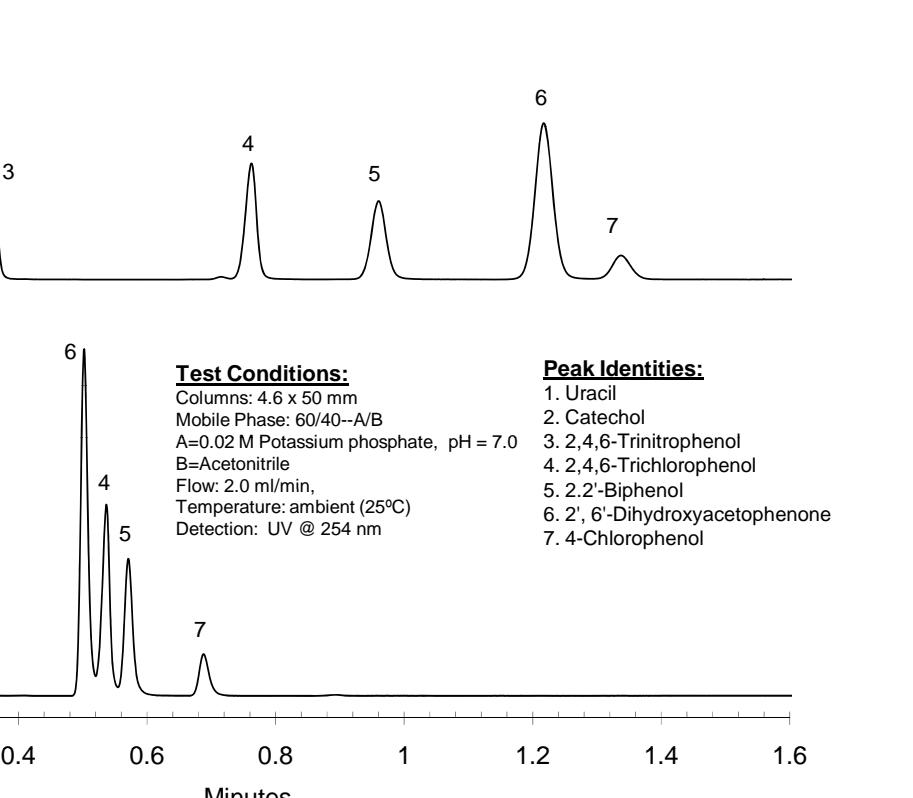
$$R_s = \left(\frac{1}{4} \right) \sqrt{N} \left[\frac{(\alpha - 1)}{\alpha} \right] \left[\frac{k_2}{(1 + k)} \right]$$

Resolution is directly proportional to selectivity (α), but is only proportional to the square root of efficiency, N .

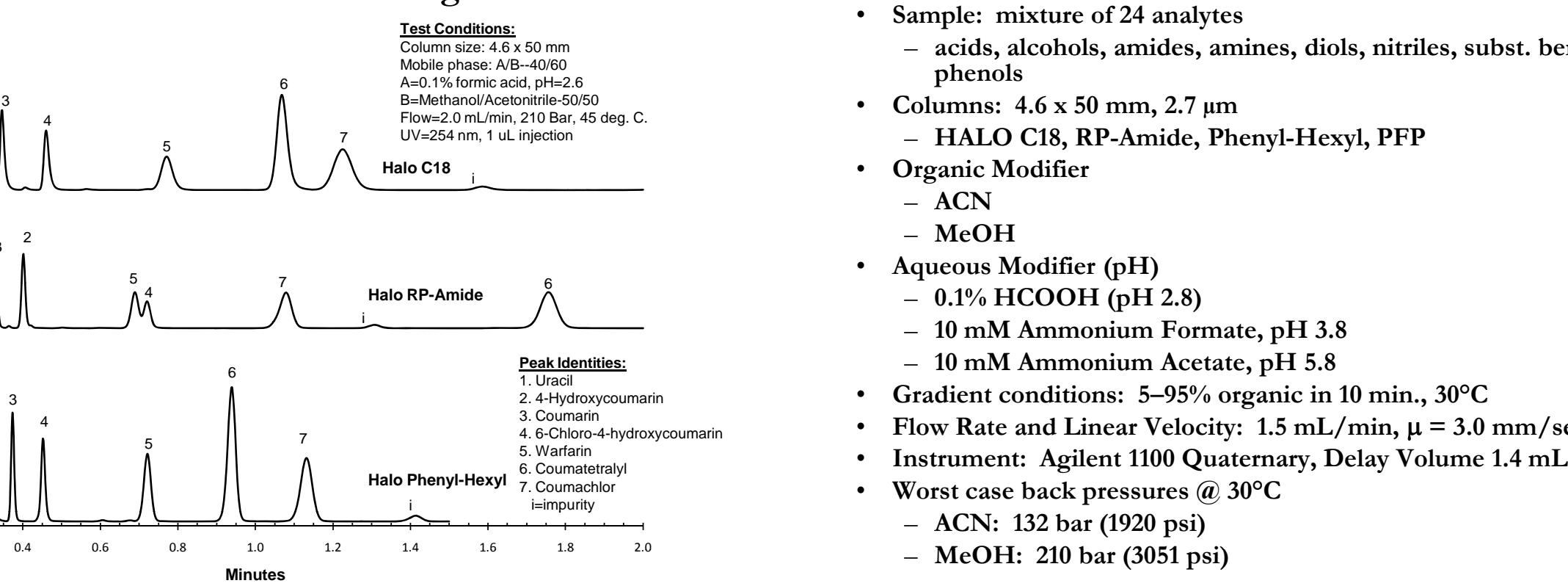


Source: Jun Mao, PhD Thesis with Professor Peter Carr, U. of Minnesota, 2001

HALO C18 vs. RP-Amide for Phenolics

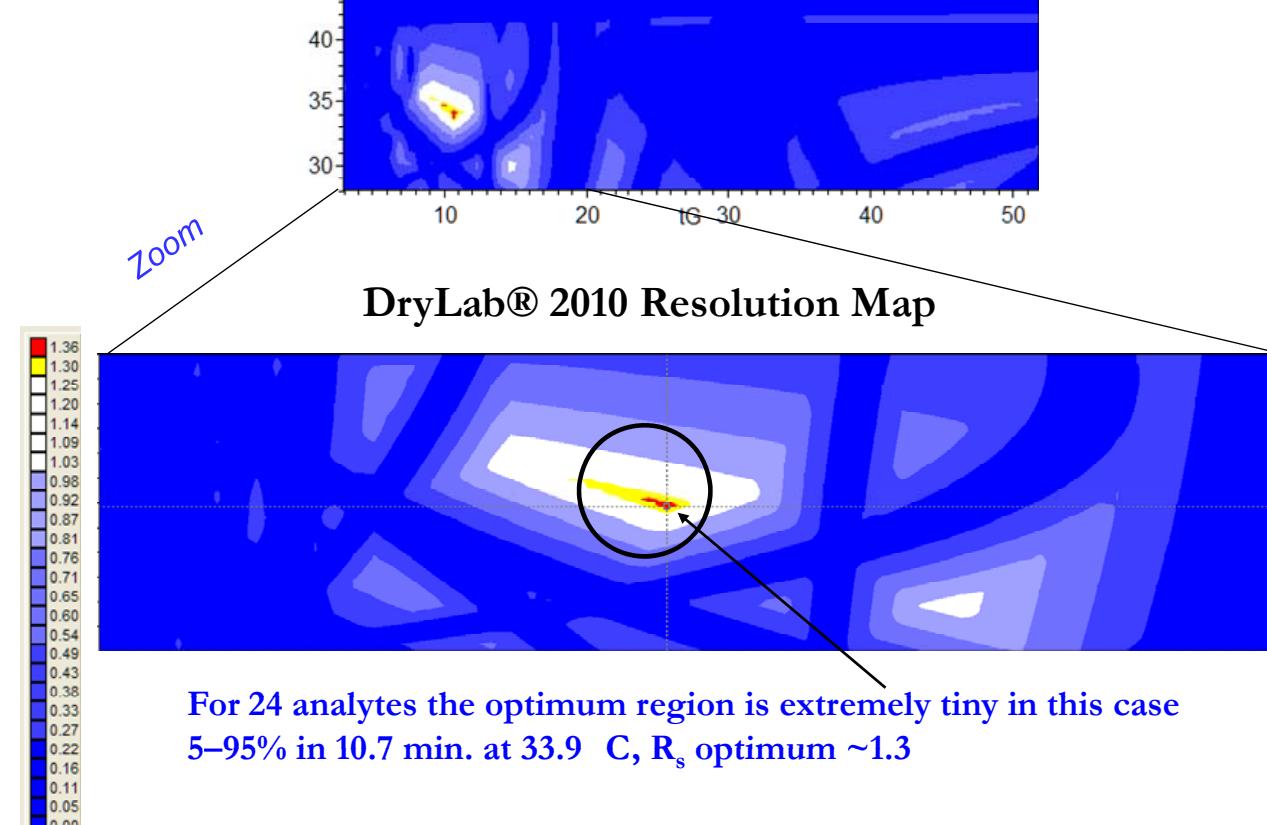


Fast Separation of Anticoagulants on HALO Fused-Core Packings



Column Phase Screening: Conditions

- Sample: mixture of 24 analytes
 - acids, alcohols, amides, amines, diols, nitriles, subst. benzenes, phenols
- Columns: 4.6 x 50 mm, 2.7 μm
 - HALO C18, RP-Amide, Phenyl-Hexyl, PFP
- Organic Modifier
 - ACN
 - MeOH
- Aqueous Modifier (pH)
 - 0.1% HCOOH (pH 2.8)
 - 10 mM Ammonium Formate, pH 3.8
 - 10 mM Ammonium Acetate, pH 5.8
- Gradient conditions: 5-95% organic in 10 min., 30°C
- Flow Rate and Linear Velocity: 1.5 mL/min, $\mu = 3.0 \text{ mm/sec}$
- Instrument: Agilent 1100 Quaternary, Delay Volume 1.4 mL
- Worst case back pressures @ 30°C
 - ACN: 132 bar (1920 psi)
 - MeOH: 210 bar (3051 psi)



Summary

- Changing selectivity, α , is most effective way to improve resolution and adjust elution order.
- Column phases having different retention mechanisms are one of the top 3 parameters for changing selectivity of an HPLC separation.
- The HALO Fused-Core column family has a set of orthogonal phases to enable fast, effective method development.
- Combining HALO phase selectivities with different organic modifiers and blends over a broad pH range is an excellent approach for developing superior, robust and rugged RPLC methods.