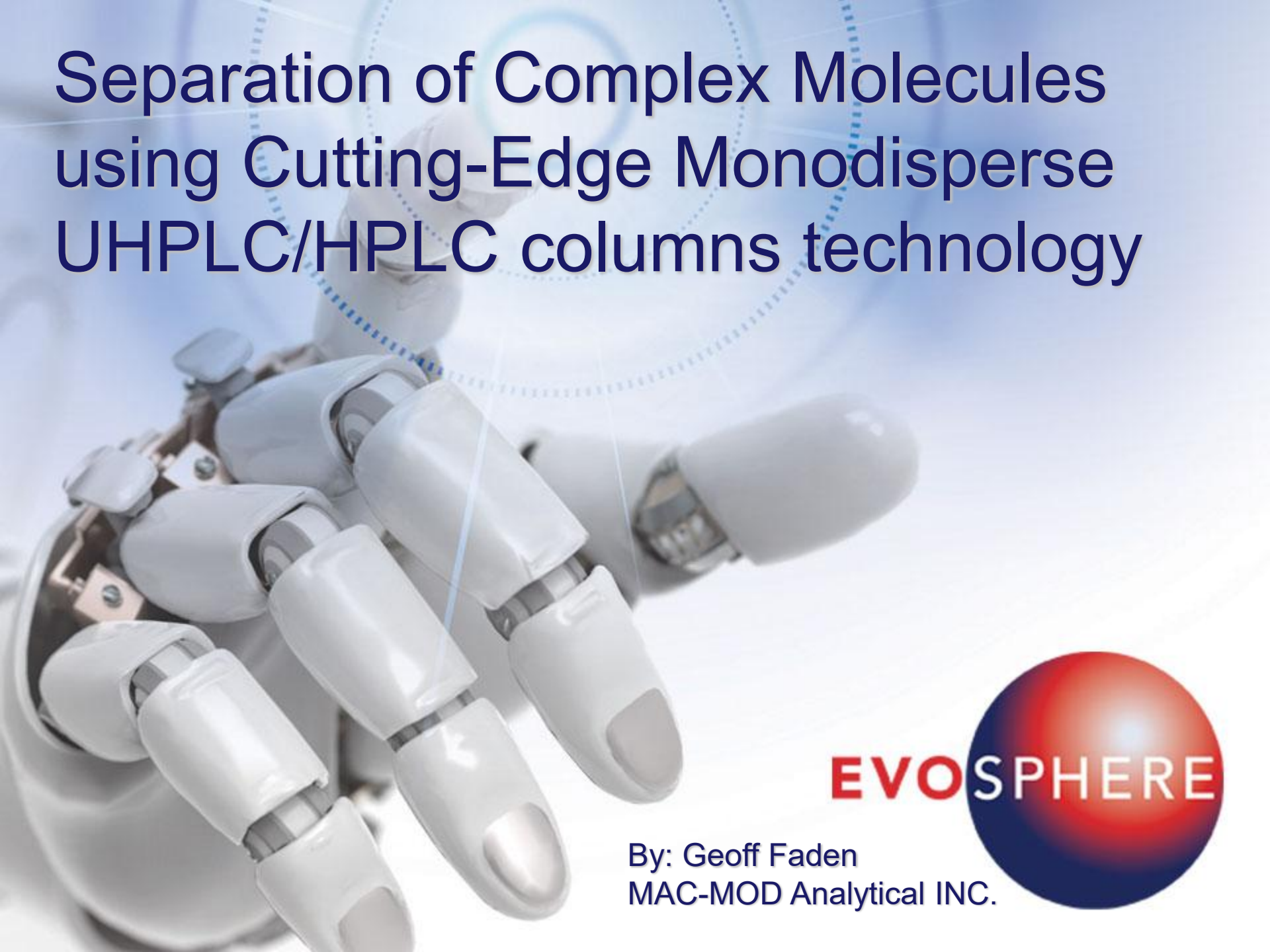



Separation of Complex Molecules using Cutting-Edge Monodisperse UHPLC/HPLC columns technology



By: Geoff Faden
MAC-MOD Analytical INC.

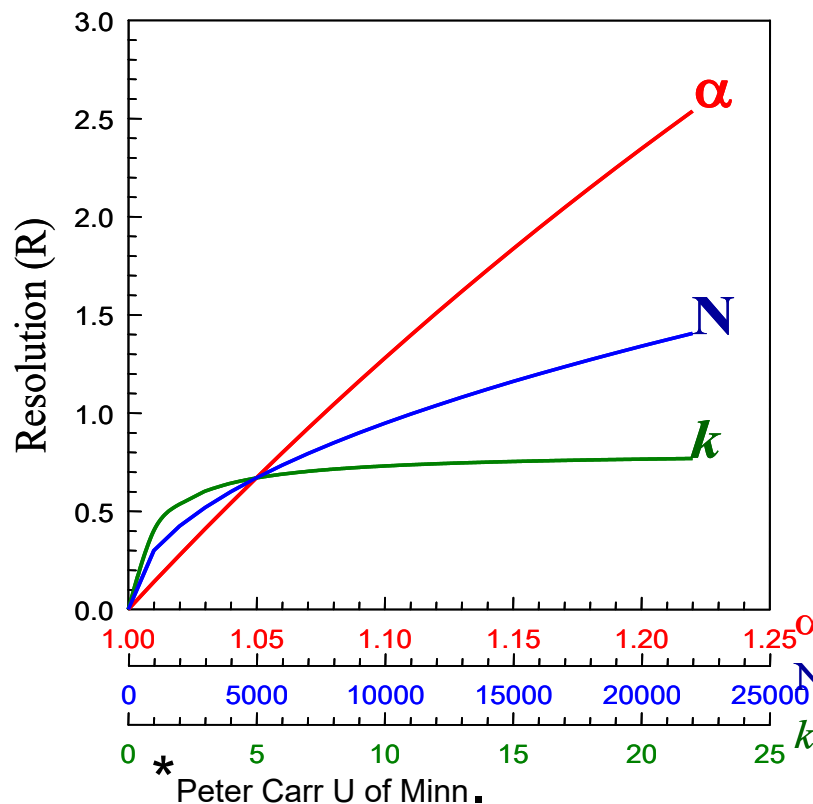
Evolution of UHPLC & HPLC Particles

- Morphology – Shape
 - Irregular, spherical, core-shell, etc.
- Size – Reduction
 - 100 μm  1.7 μm
- Silica Purity – Less Metals
- Size Distribution – Reduction in D90/D10
- Particle Hybridization –
 - Fully Silica, Coated Silica, Hybridized Silica

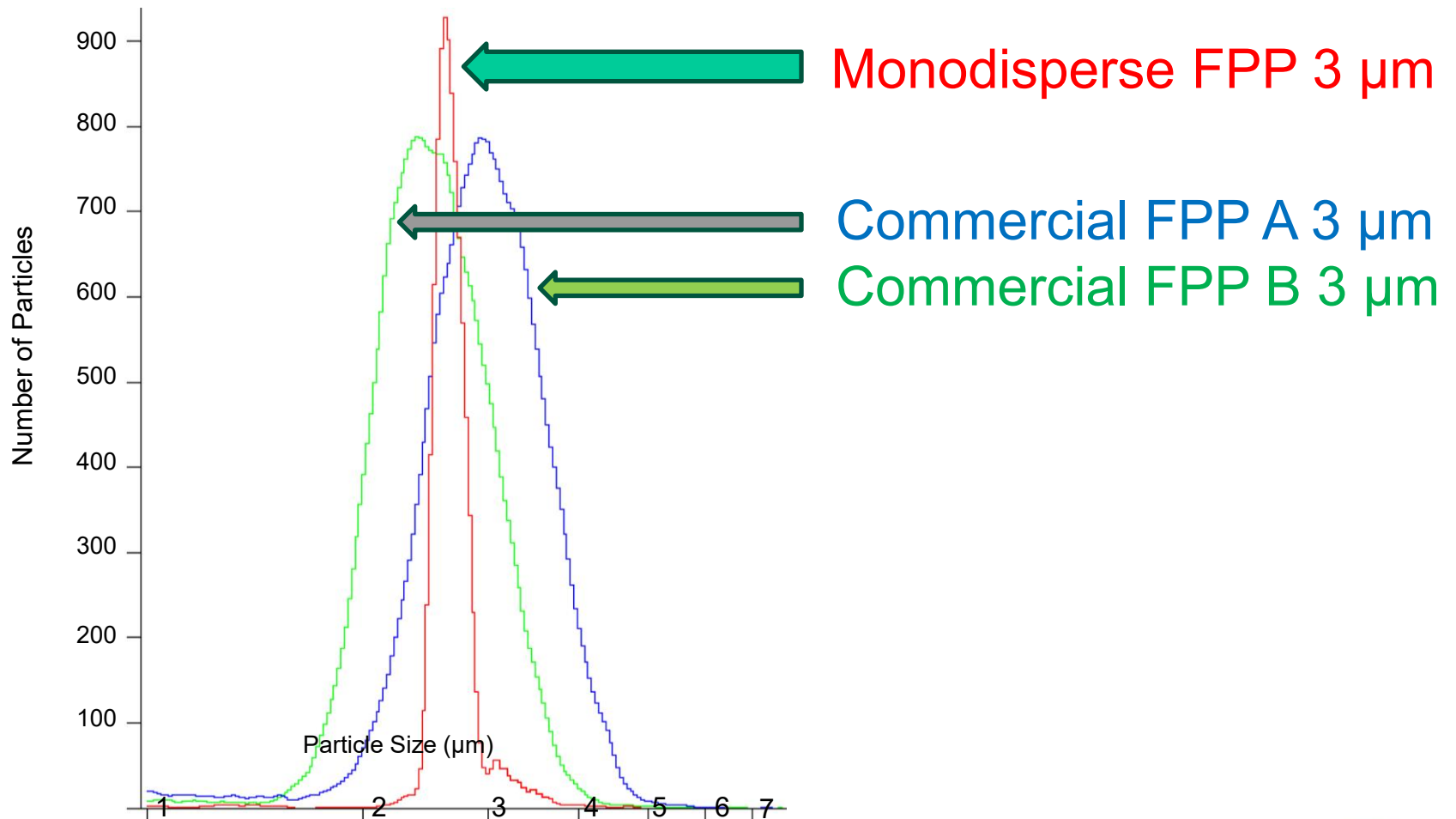
Resolution Equation

Efficiency	Retention	Selectivity
↓	↓	↓
$R = \frac{\sqrt{N}}{4}$	$\frac{k'}{k'+1}$	$\frac{\alpha-1}{\alpha}$

$$N = \frac{\text{Length of Column}}{\text{HETP}}$$



Monodisperse vs. Competitor Polydisperse Comparisons

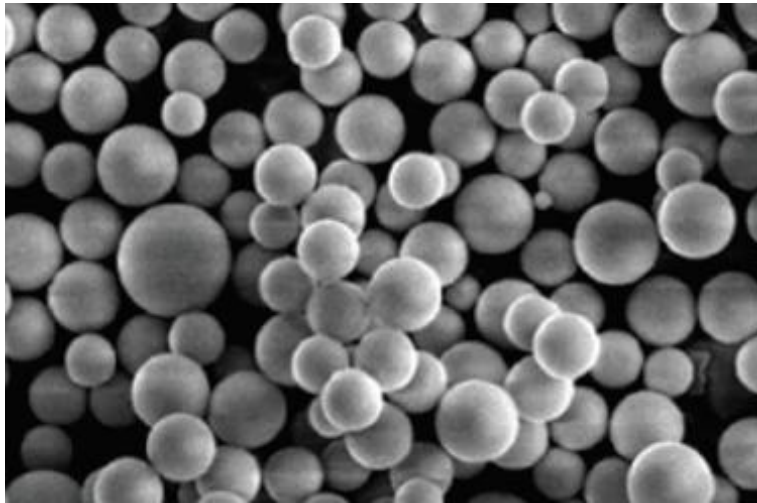


Defining Monodispersity

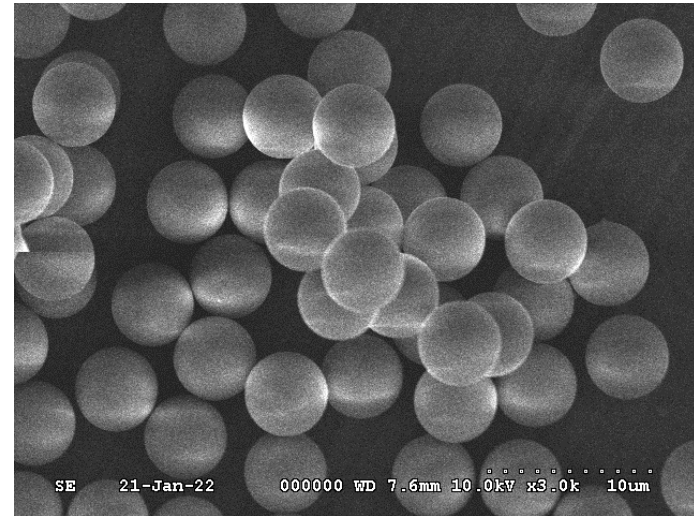
	Monodisperse silica	Commercial 3u silica - A	Commercial 3u Silica-B
Mean particle size (d50) *	2.66 μ m*	2.49 μ m	2.97 μ m
SEM particle diameter	3.0 μ m	2.8 μ m	3.3 μ m
D90/10	1.12	1.58	1.61
Pore volume	0.89	0.88	0.89

* Measured by Coulter Counter

SEM Comparison of Particles



Traditional
porous particles



Monodisperse
Fully Porous Particles

What is the Effect of Monodispersity

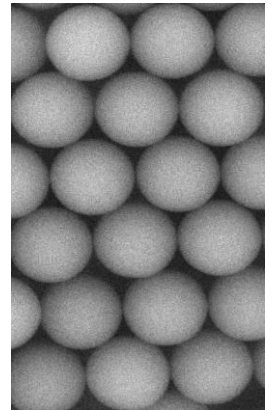
When particles have **very narrow size distribution**, they can arrange more regularly during column packing.

Effects include:

- More **uniform interstitial channels**
- Reduced **void heterogeneity**
- Less **local channeling**

This reduces the **variance of flow velocities** across the column.

However, **monodispersity alone does not directly shorten diffusion distances** inside a particle. Instead, it allows the bed to approach an **ordered packing structure** (often close to hexagonal or Face centred cubic (FCC) locally).



Effect of Improved Packed Bed Structure

The improved packing structure leads to:

- **Lower eddy dispersion (A-term)**
- **More uniform tortuosity**
- **Narrower flow path distribution**

In the **van Deemter framework**, this appears as a **smaller A-term**:

$$H = A + \frac{B}{u} + Cu$$

Monodisperse packings often achieve:

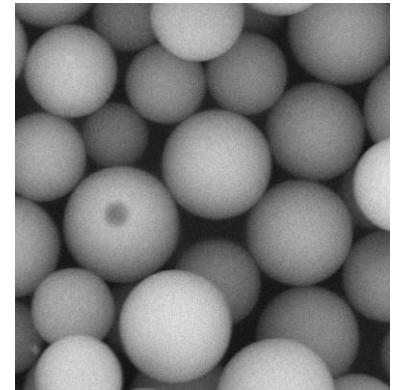
- **Reduced plate height $h = H/d_p$ of ~1.5–2**
- **Compared with ~2–3 for polydisperse packings**

This improvement arises from **more homogeneous interstitial flow paths**, not from a fundamentally different molecular diffusion mechanism.

Why Polydisperse Particles Give less Performance

If particle sizes vary:

- Smaller particles fill gaps between larger ones
- Packing becomes **structurally heterogeneous**
- Flow channels become **irregular**
- Some regions have **fast flow**, others **slow flow**



This increases **dispersion** by **eddy diffusion** because molecules experience different path lengths.

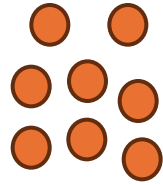
Key Findings of Monodispersity

The **reduced flow path dispersion** is:

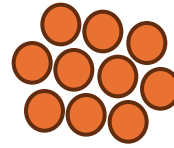
- **Not directly a property of monodispersity**

But.....it is a property of the improved bed geometry that monodispersity enables

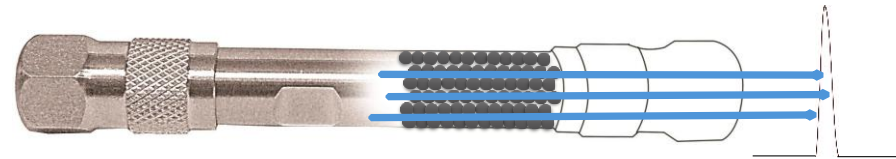
Monodisperse HPLC - Causal Effect



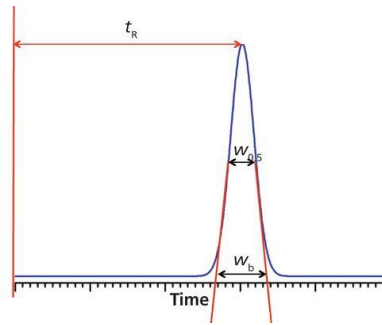
Monodisperse
Particles



More Ordered
Packing

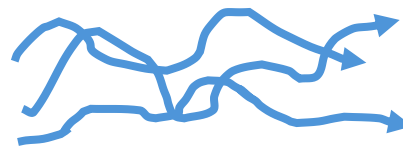


More uniform
flow channels



Lower plate height

Higher Efficiency



Reduced Eddy
Dispersion

Column Plate count Theory versus Reality

- **5 μm fully porous columns (150 mm)**

- ~10,000–15,000 plates

- ~70k–100k plates/m

- **3 μm columns (100 mm)**

- ~12,000–18,000 plates

- ~100k–140k plates/m

- **1.7 μm UHPLC columns (100 mm)**

- ~20,000–35,000 plates

- ~170k–250k plates/m

- **2.6 μm core-shell columns (100 mm)**

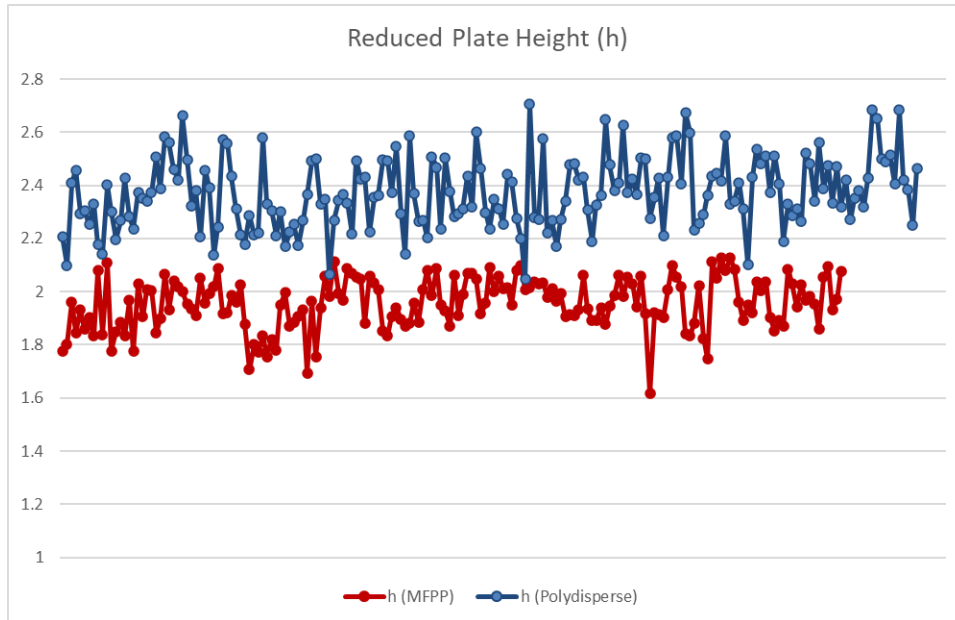
- ~18,000–25,000 plates

- ~160k–240k plates/m

What manufacturers typically report

Examples from common column types:

Monodisperse vs Polydisperse



	h	N/m
Max	2.817 =	71,042
Min	2.046 =	97,809
Ave	2.386 =	83,849

5um
Polydisperse

Max	2.120 =	94,680
Min	1.615 =	123,842
Ave	1.950 =	102,585

5um
Monodispers
e

The reduced flow path in columns packed with monodisperse particles is primarily a consequence of the improved packed bed structure, although monodispersity is the enabling factor that allows that packing structure to form.

In other words:

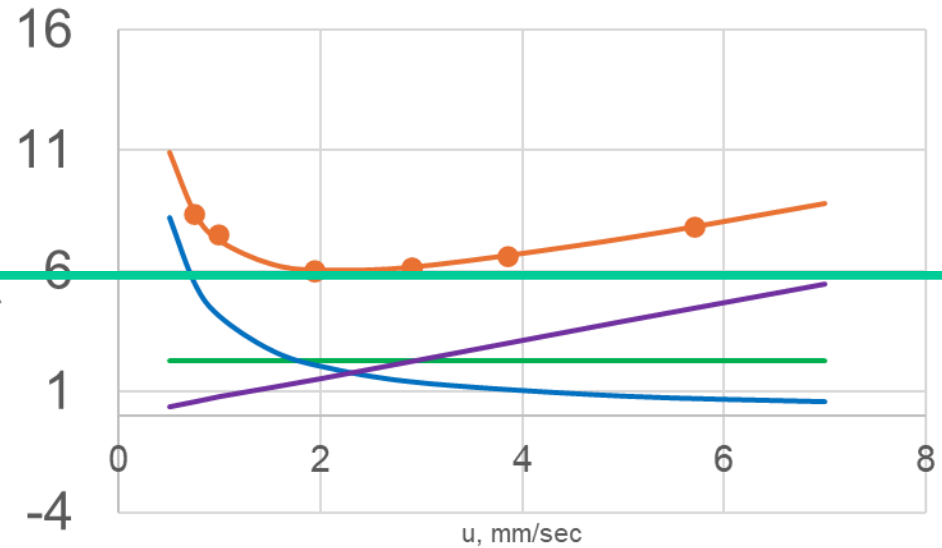
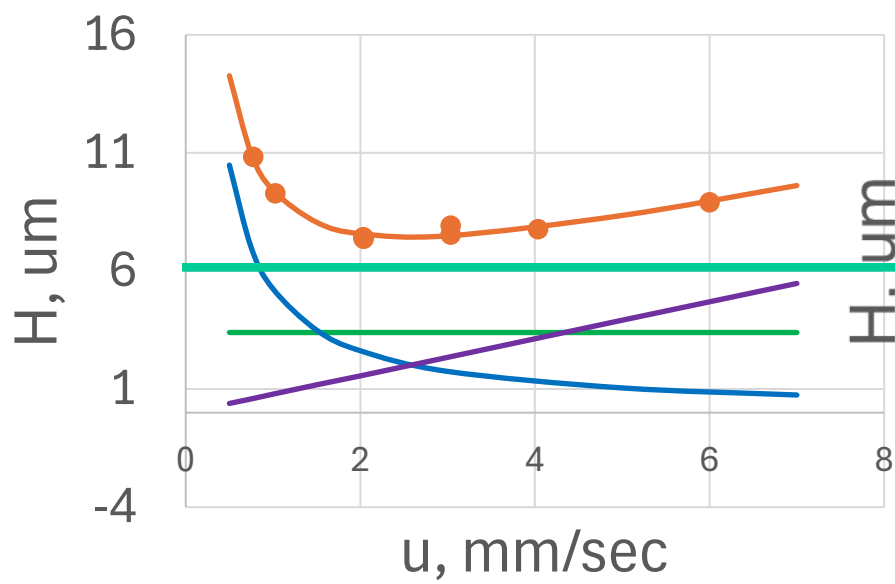
- Monodispersity → enables better packing
- Better packing → produces the shorter / more uniform flow paths

Experimental Van Deemter fits for Dihexylphthalate (DHP) of Polydisperse vs Monodisperse

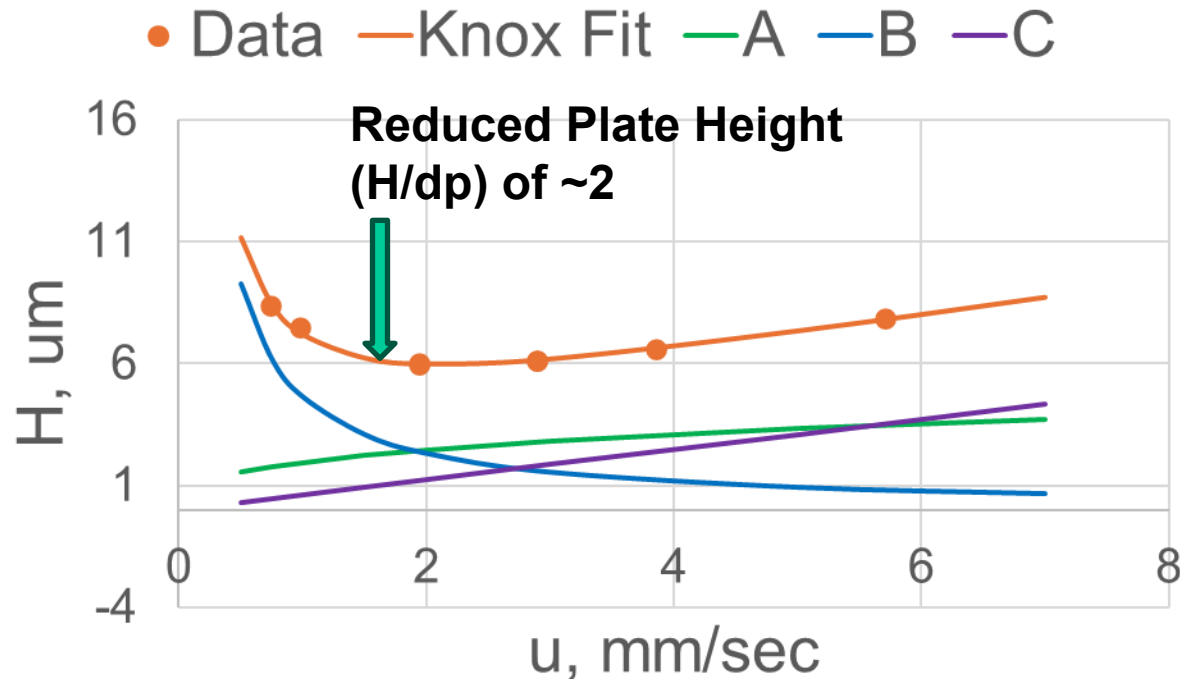
Polydisperse 3 μm C18 (DHP, $k=13.0$)

Monodisperse 3 μm C18 (DHP, $k=6.81$)

● Data — vanDeemter Fit — A — B — C ● Data — vanDeemter fit — A — B — Series5

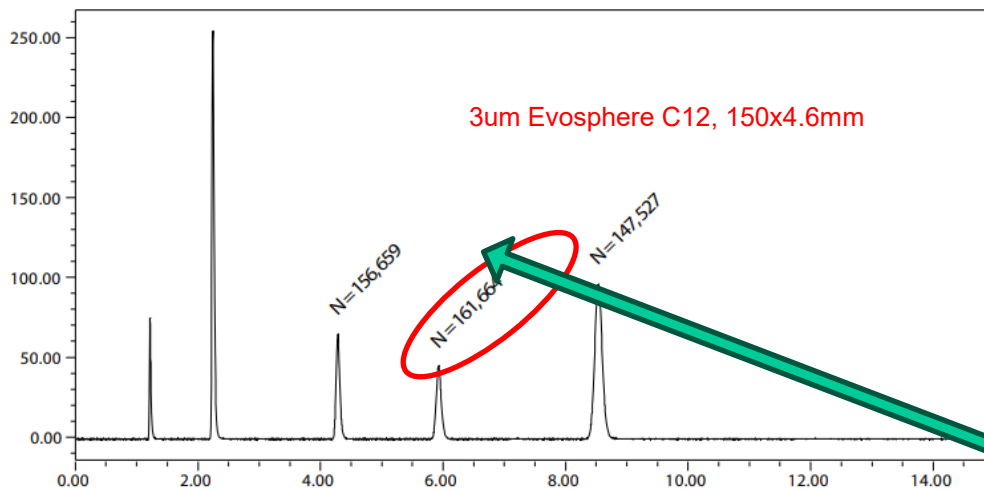


Monodisperse 3 μm (3.0 mm x 50 mm) Van Deemter Curve for Dihexylphthalate (DHP)



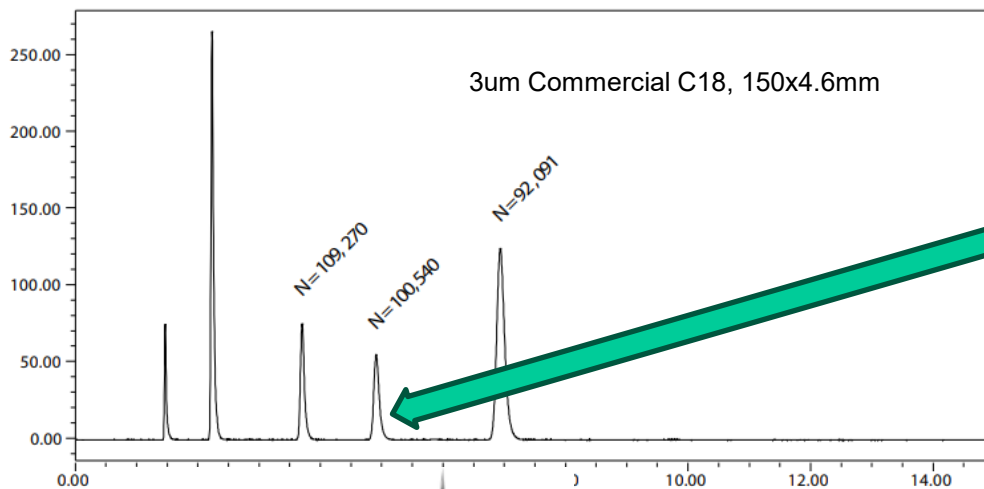
Data Generated by Merlin Bicking from ACCTA, Inc.

Unprecedented (N) Efficiency Gains



MONODISPERSE

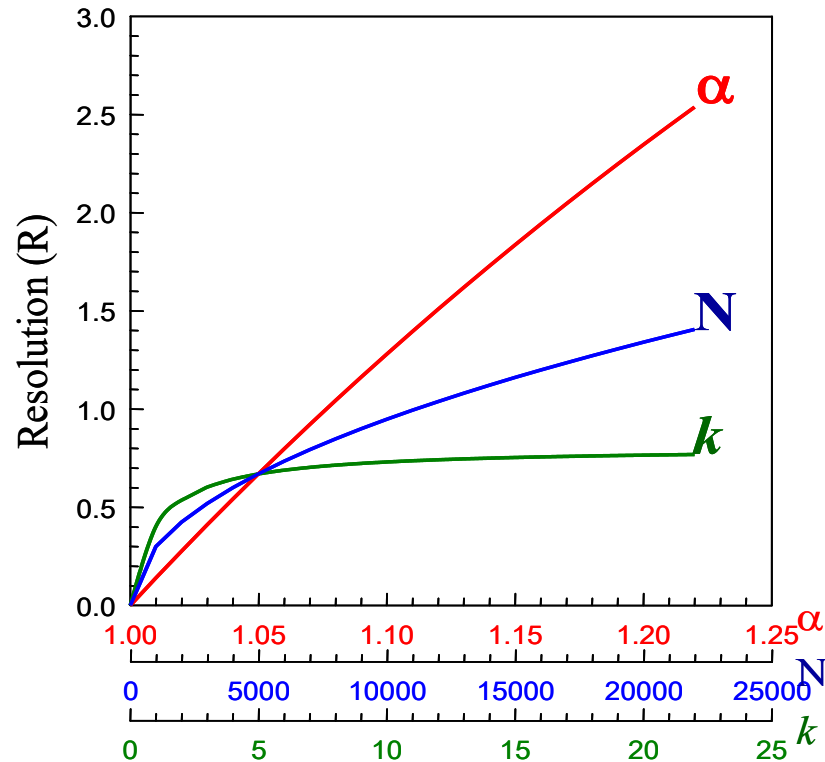
60% Higher N



POLYDISPERSE

Resolution Equation Pt. 2

Efficiency	Retention	Selectivity
↓	↓	↓
$R = \frac{\sqrt{N}}{4}$	$\frac{k'}{k'+1}$	$\frac{\alpha-1}{\alpha}$

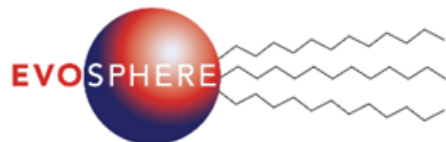


Selectivity is the Key to Resolution!

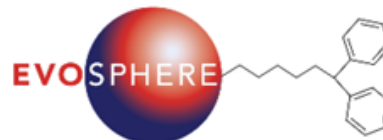
*P.W. Carr's Group U of M

Powerful Selectivity Choices

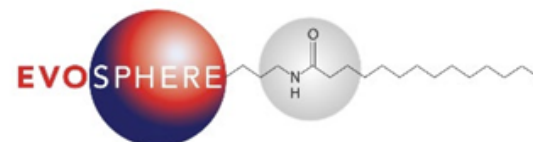
Evosphere C12



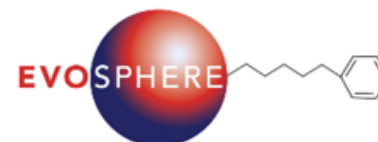
Evosphere Diphenyl



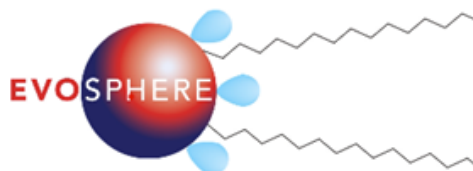
Evosphere RP18-Amide



Evosphere Phenyl-Hexyl



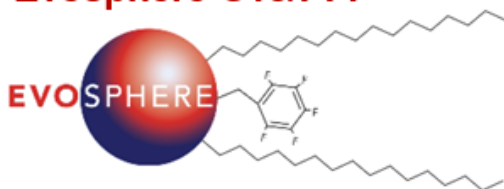
Evosphere AQUA



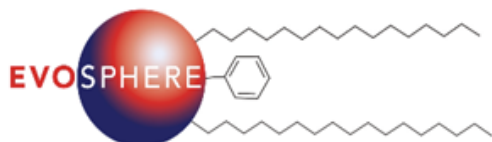
Evosphere PFP



Evosphere C18/PFP



Evosphere C18/AR



Efficiency	Retention	Selectivity
$R = \frac{\sqrt{N}}{4}$	$\frac{k}{k'+1}$	$\frac{\alpha-1}{\alpha}$

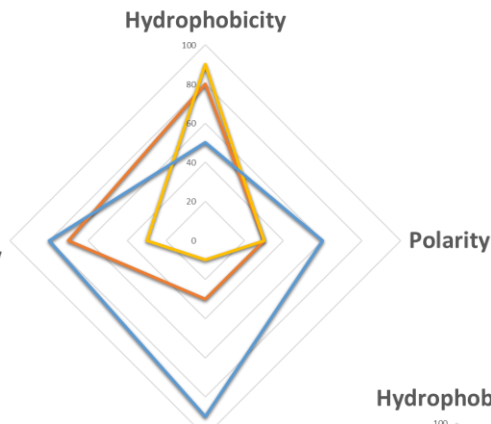
Orthogonal Selectivity

C12
Diphenyl
C18/PFP

Screen 1



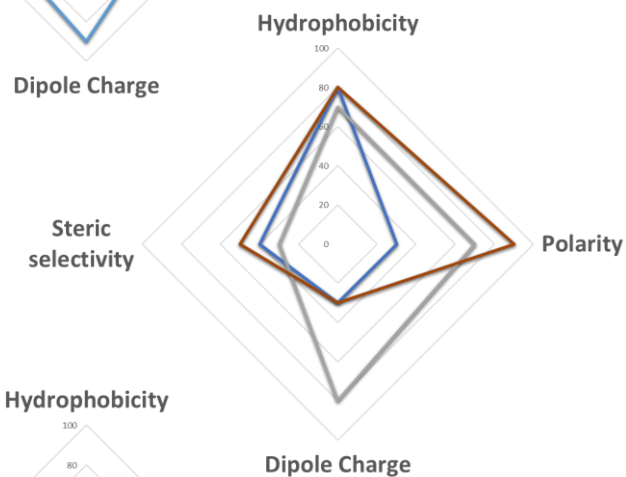
Steric
selectivity



Screen 2



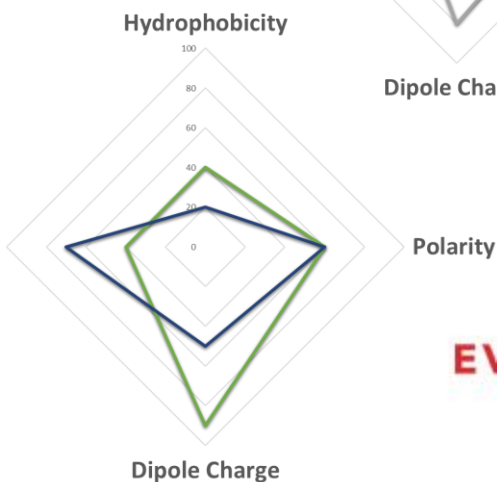
Steric
selectivity



Screen 3



Steric
selectivity



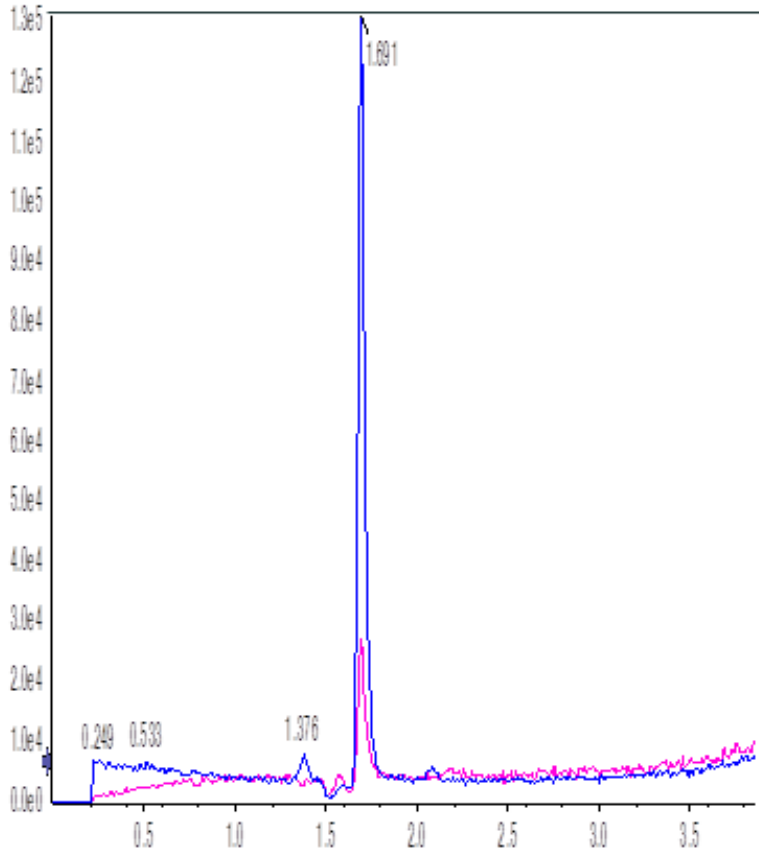
PFP
Phenyl-Hexyl



Challenges - a LC perspective

- Complex molecules require:
 - **Retention**
 - *Polar compounds are difficult to retain in RP*
 - *Non-polar*
 - **Sensitivity**
 - *Peak height (MS)*
 - *LOD*
 - **Resolution**
 - *MS (limited MP choices) vs LC (expanded buffer ranges)*
 - *Reproducibility*
 - *Ruggedness*

Sugar Analysis in Reversed Phase Polar Retention - Dextrose (D-Glucose) on Evosphere Diphenyl



Column: Evosphere Diphenyl
Column Format: 100 Å, 1.7 µm, 3.0 x 100 mm
Mobile Phase

- **A: 0.1% Formic Acid in Water**
- **B: 0.1% Formic Acid in 50% Methanol and 50% Acetonitrile**



Untargeted Metabolomics Screen - TIC

Column Phase – Evosphere C18/PFP

Dimensions - 3 μ m (2.1 mm x 100 mm)

Instrument - Thermo Q-Exactive with Dionex UHPLC

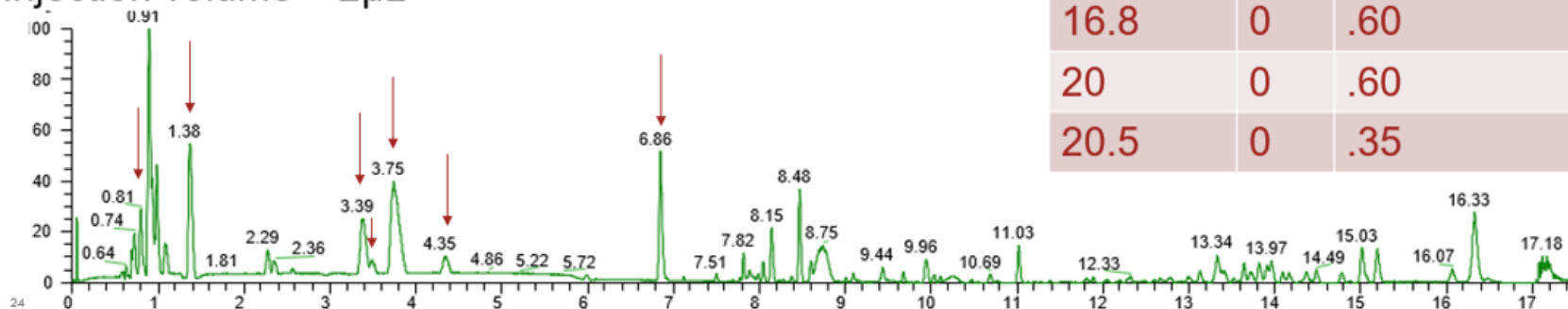
Sample – Plasma Extract

Mobile Phase A = 0.1% Formic Acid in H₂O

Mobile Phase B = Acetonitrile

Temperature = 25°C

Injection volume = 2 μ L

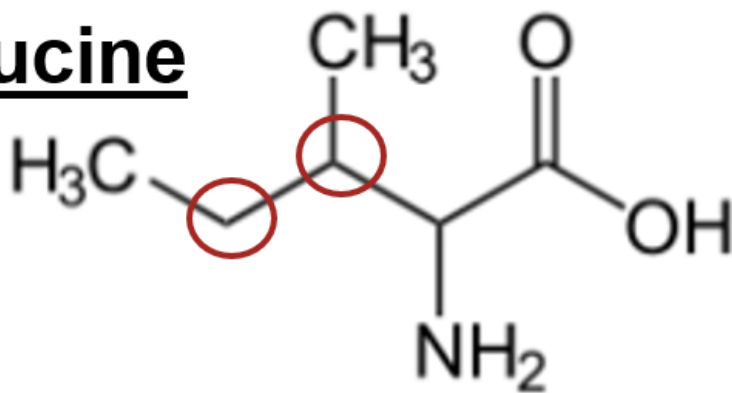


Time	% B	Flow Rate (mL/min)
3 min	0	.35
13 min	80	.35
16 min	80	.35
16.5	0	.35
16.8	0	.60
20	0	.60
20.5	0	.35

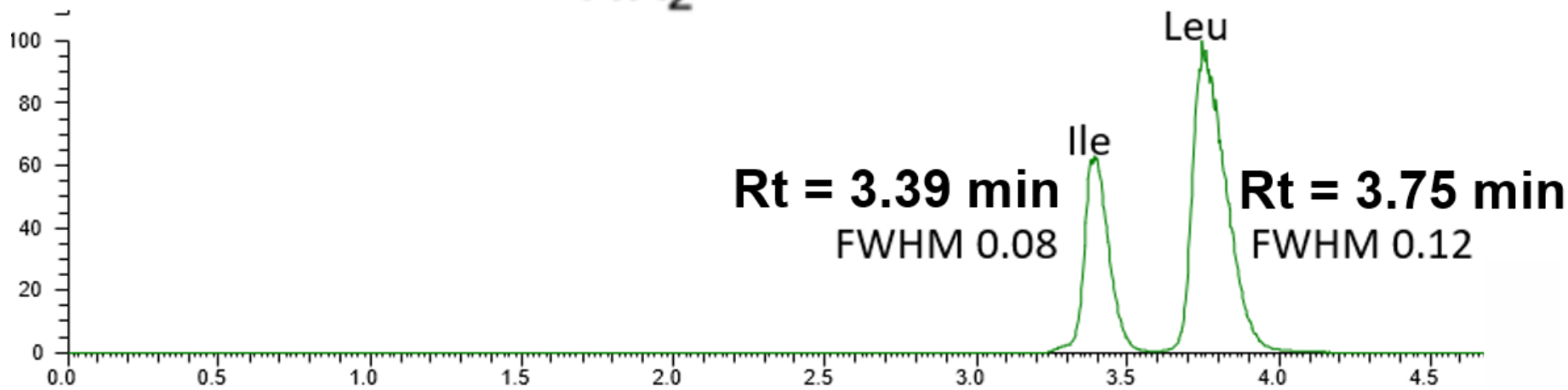
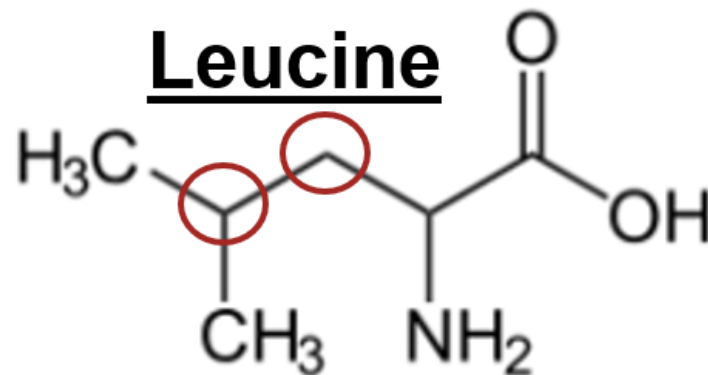
Data Generated from Tim Garret PhD

Methyl Position Switch

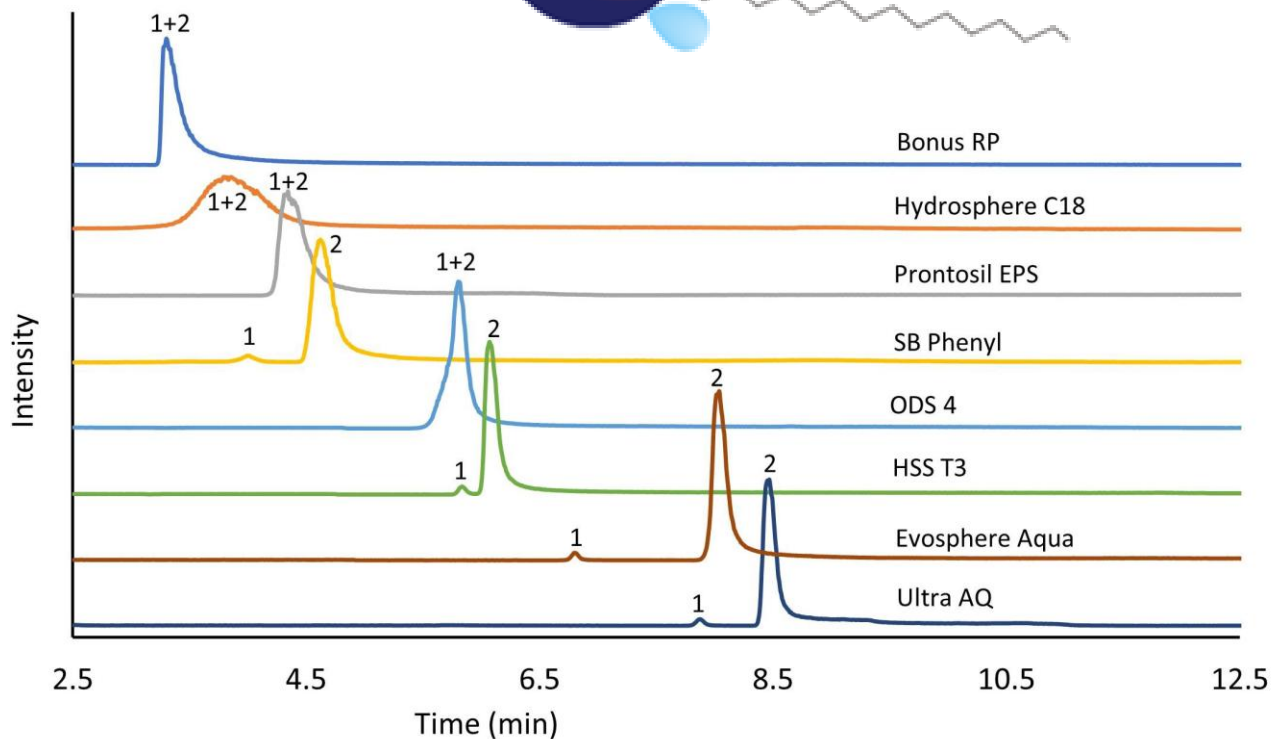
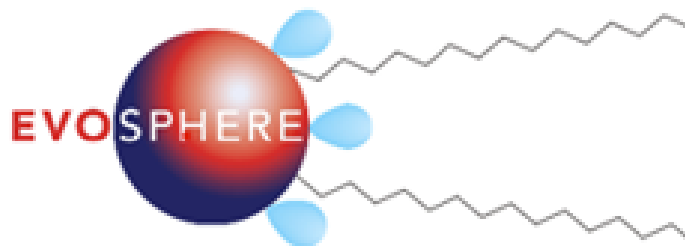
Isoleucine



Leucine

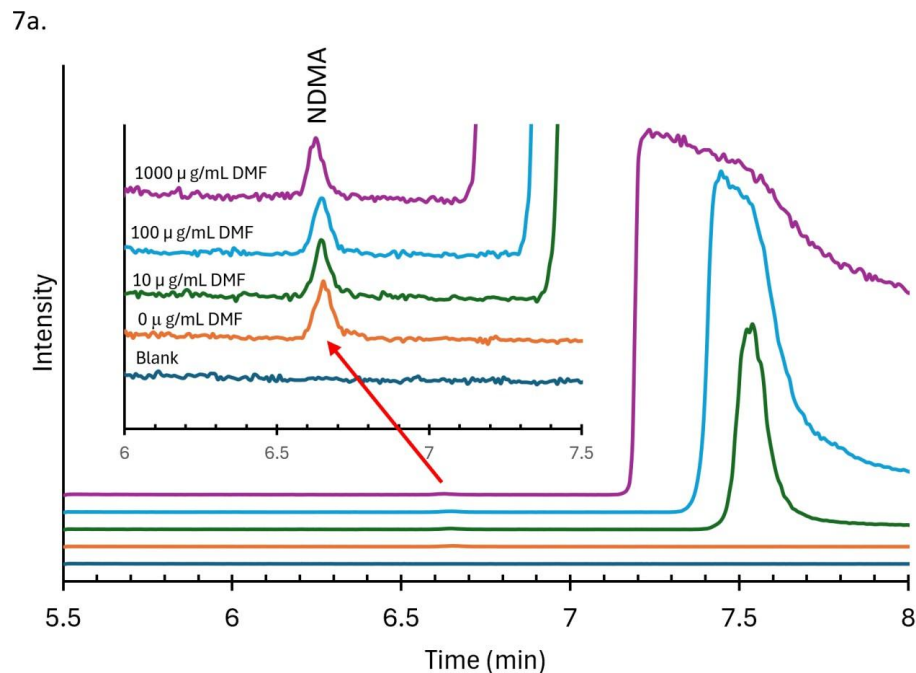


Initial Merck Screening wettable phase chemistries for Nitrosamine studies



The link to the online article - [Analysis of N-Nitrosodimethylamine \(Ndma\) in Pharmaceutical Products in the Presence of High Concentration N,N-Dimethylformamide \(Dmf\) by Jinjian Zheng, Mikhaila D. Ritz, Ana I. Martinez, Timothy Yaroshuk, Xihui Liang, Xiaoyi Gong, Mark D. Mowery :: SSRN](#)

Why Evosphere Aqua was chosen:



- Significant resolution between DMF and NDMA
- Multiple lots were validated multiple sites
- Separation confirmation was done via single quad MS, but validated on LC-UV
- Enough Rs to overload $1 \times 10^6:1$ DMF to NDMA

Non-Ion-Pair Method for Oligos on Evosphere C18/AR INERT



Column: Evosphere C18/AR INERT 1.7 μm (2.1 mm x 100 mm)

Mobile Phase - (Unmodified Oligonucleotides) - 10-30% v/v MeOH in 10 minutes

Mobile Phase - (Modified Oligonucleotides) - 15-50% v/v MeOH in 10 minutes

Temperature - 30°C or 60°C

Flow Rate - 0.3 mL/min

Injection Volume - 0.5 μL

Instrument - Thermo Scientific™ Vanquish™ Horizon UHPLC system with DAD

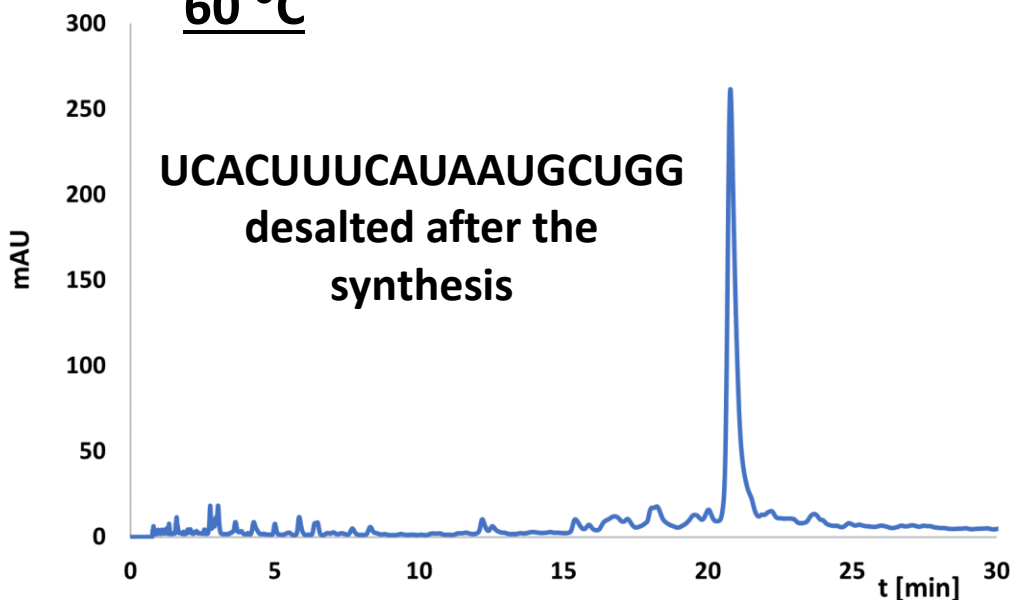
Autosampler Temperature - 4°C

UV Wavelength – 260 nm

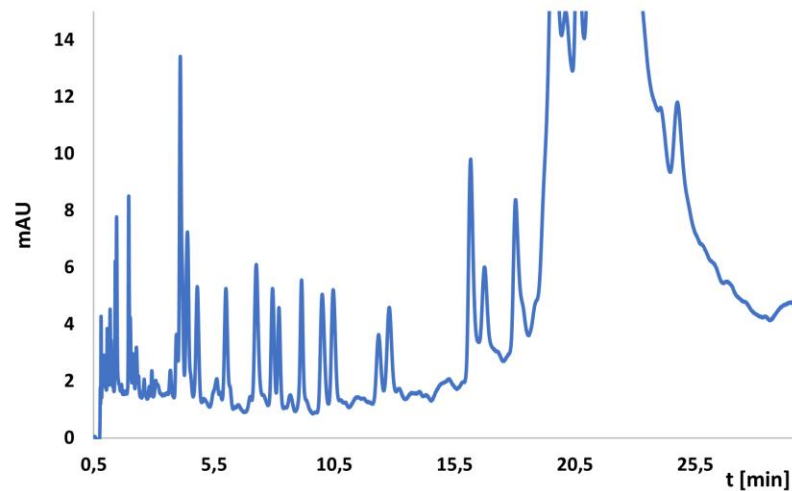
Unpurified Oligonucleotide Separation

25mM Amm. Ac. pH 6
2-8% MeOH 30min
60 °C

UCACUUUCAUAAUGCUGG
desalted after the
synthesis

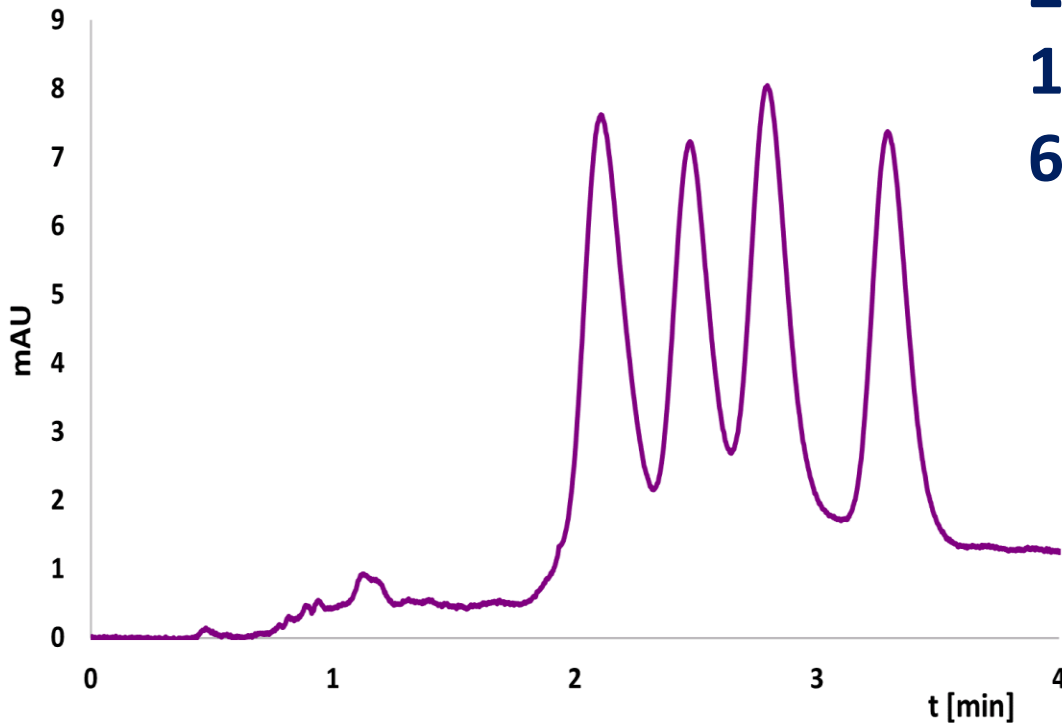


Magnified Chromatogram



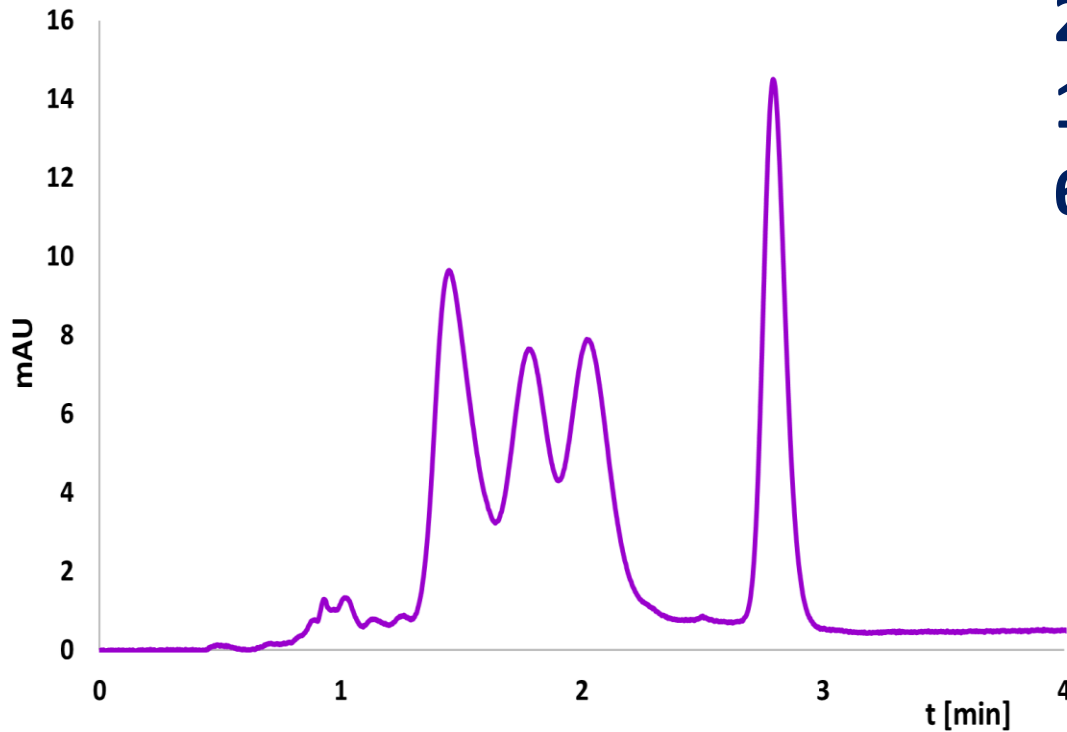
Separation of Unmodified Sequence Isomers

25mM AA pH=6
12-20 % MeOH in 10min
60 °C



- 1 - ATCGATCGATCGATCGATCG
- 2 - ATCGATCGATCGATCGATCT
- 3 - ATCGATCGATCGATCGATCC
- 4 - ATCGATCGATCGATCGATCA

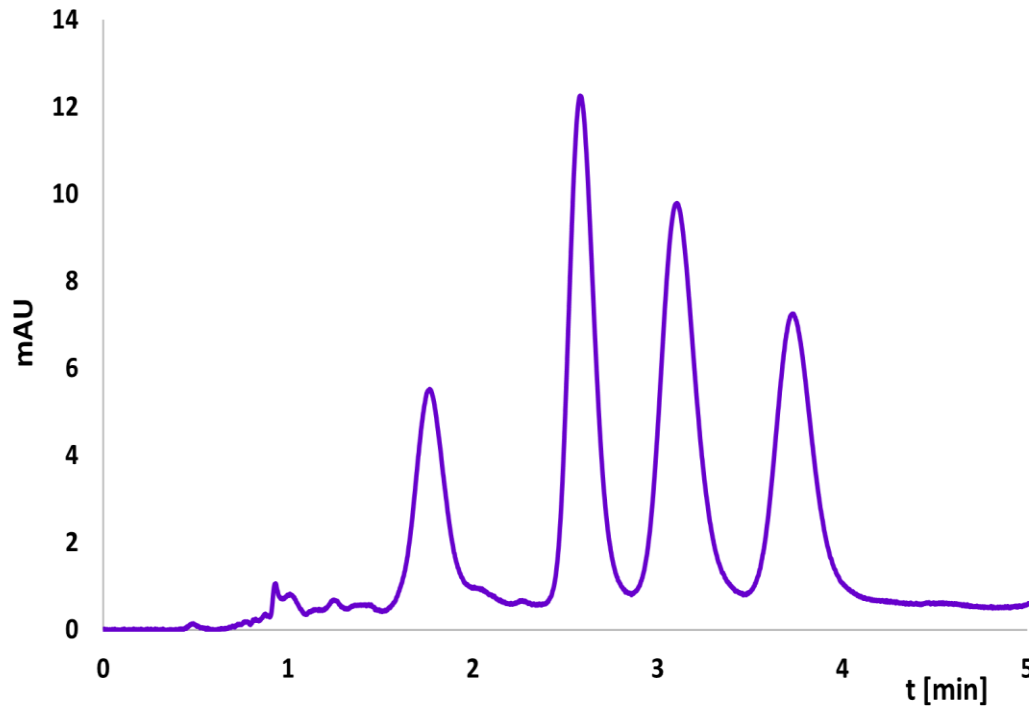
Separation of Sequence Isomers



25mM AA pH=6
12-20 % MeOH in 10min
60 °C

- 1 - ATCGATCGAGCGATCGATCG
- 2 - ATCGATCGAACGATCGATCG
- 3 - ATCGATCGATCGATCGATCG
- 4 - ATCGATCGACCGATCGATCG

Separation of Sequence Isomers



25mM AA pH=6

12-20 % MeOH in 10min

60 °C

1 - ATCGATCGAACGATCGATCG

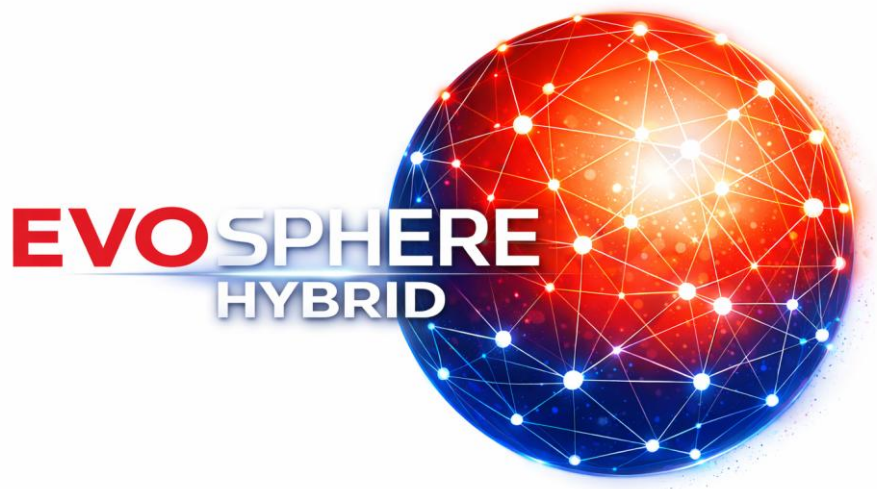
2 - ATCGATCGATAGATCGATCG

3 - ATCGATCGATCGATCGAACG

4 - ATCGATCGATCGATCGATCA

Launching at HPLC 2026

Evosphere Hybrid Monodisperse C18!



- pH Range – 1-12
- Temperature Range – 40°C - 90°C
- Available in:
 - All analytical sizes (1.7, 3.0, and 5 μm)
 - Preparative Sizes (7.8, 10.0, 21.2 mm, 30.0 mm, and 50 mm ID)
 - Inert Coated Hardware

The Evolution of HPLC Columns



Analytical and Preparative Sizes readily Available!

Conclusion

- Monodisperse Particle Columns lead to lower Reduced Plate Heights in the same standard UHPLC and HPLC formats
- Leveraging higher N and surprising polar retention can lead to useful tools for the chromatographers toolbox

Acknowledgements

- All Industry and Academic Collaborators
- All MM collaborators including Edward Faden and Alex Nasseh
- All Fortis Technologies Collaborators including Mark Woodruff, and Ken Butchart