



# THE IMPORTANCE OF SUPERFICIALLY POROUS PARTICLES IN MODERNIZING HPLC METHODS

EAS 2019

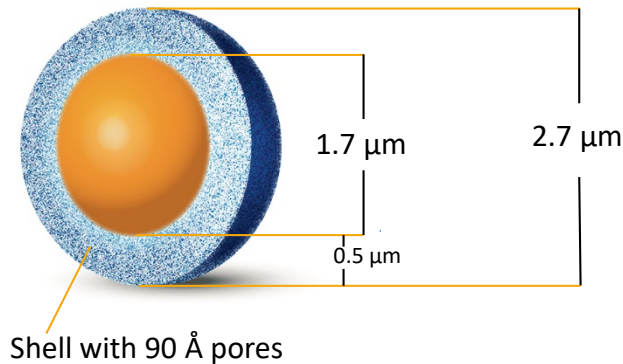
Presented by:

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Independent Consultant

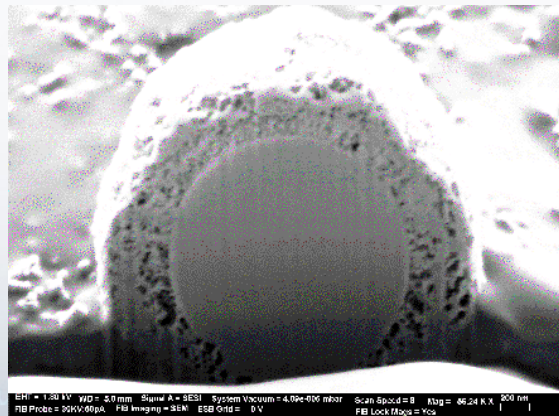


# The Unique Superficially Porous Particle (SPP)

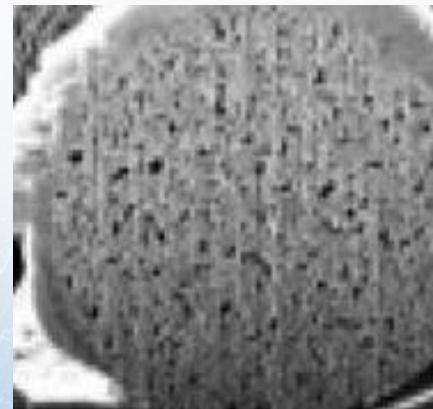


- Highest purity Type B silica
- Nonporous silica core
- Porous silica shell
- Shell thickness and pore size tightly controlled
- Particle size highly uniform

HALO® Particle (SPP)

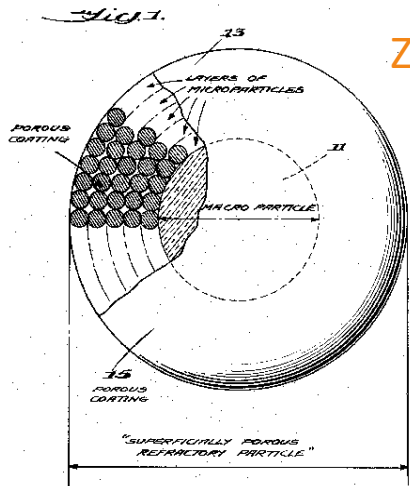


Fully Porous Particle (FPP)



# Milestones in Fused-Core® History

April 14, 1970 J. J. KIRKLAND 3,505,785  
 SUPERFICIALLY POROUS SUPPORTS FOR CHROMATOGRAPHY  
 Filed June 20, 1967 3 Sheets-Sheet 1



Zipax®

- **1960**- Golay first proposed superficially porous particles (SPPs) for GC.
- **1970**- Jack Kirkland (DuPont Company), inspired by a nucleotide separation of Horvath and coworkers using ~50 µm cores with thin layer of anion exchange resin, develops Zipax® particles with layers of silica sol on 30 µm glass beads.
- **1990**- Kirkland continues to advance fully porous particle technology by developing high purity Type B silica and creates a clear performance distinction between Type A and Type B.
- **2006**- Progressively smaller fully porous Type B silica particles develop rapidly in the HPLC column market. Kirkland meets the demand for higher speed and resolution and creates modern superficially porous particles that delivers higher performance at lower pressures.



**2006** Original HALO® 2.7 µm SPP  
 changed the perception of what is required for high efficiency separations

**2012** HALO® 5 µm SPP  
 robust replacement to conventional 5 µm particle columns with SPP benefits

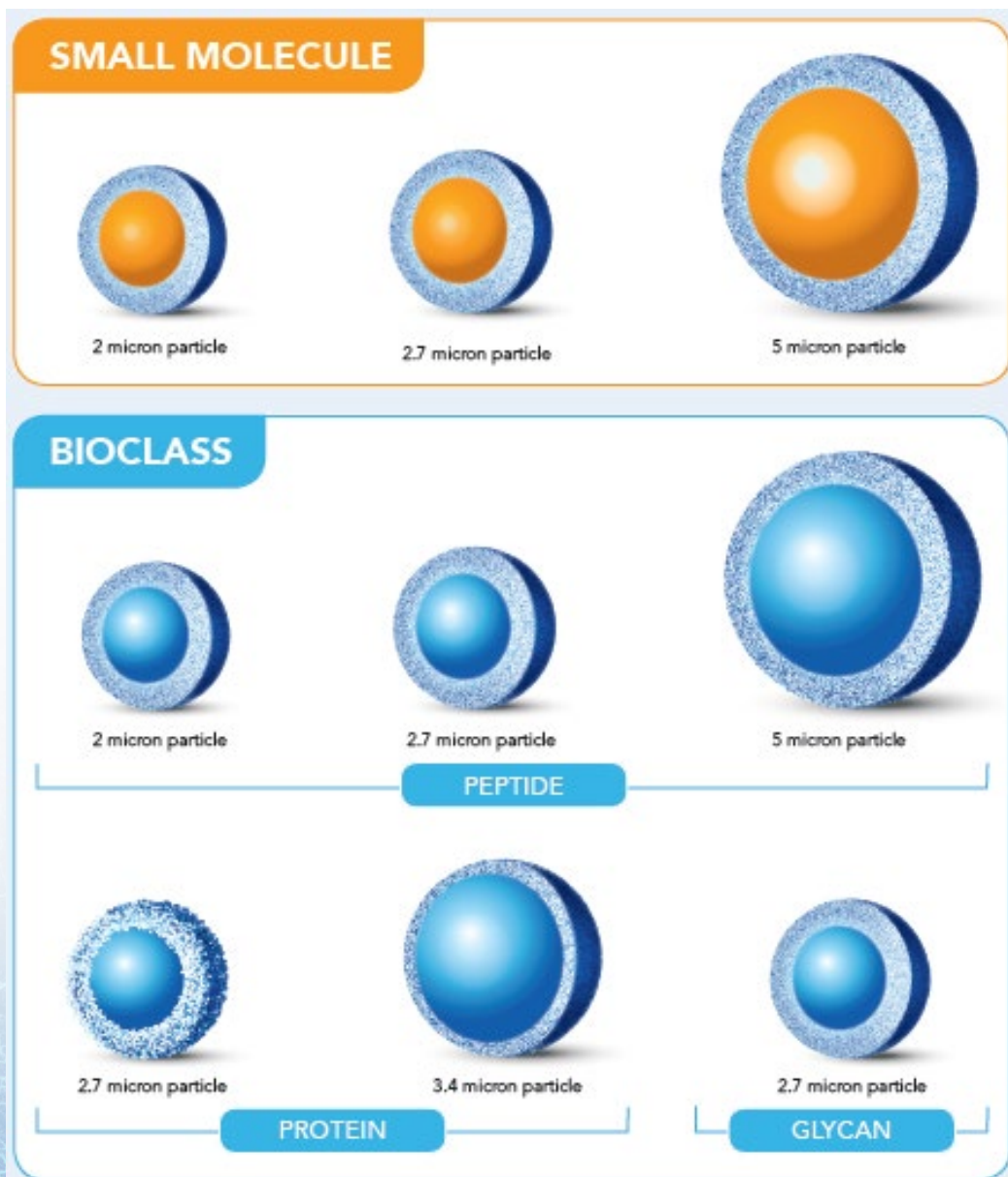
**2013** HALO® BioClass Line Introduced  
 Protein, Peptide and Glycan solutions to meet the challenges of biomolecule separations

**2014** HALO® 2 µm SPP  
 the go-to SPP for highest efficiency separations with UHPLC technology

**2017** HALO® 1000 Å Protein  
 First 1000 Å pore size providing the widest pore available in an SPP that delivered significant gains in resolution of large protein complexes



- Various stationary phases with particle and pore size morphologies
- 90 Å for small molecules
- 160 Å for intermediate size molecules
- 400 Å and 1000 Å for large molecules



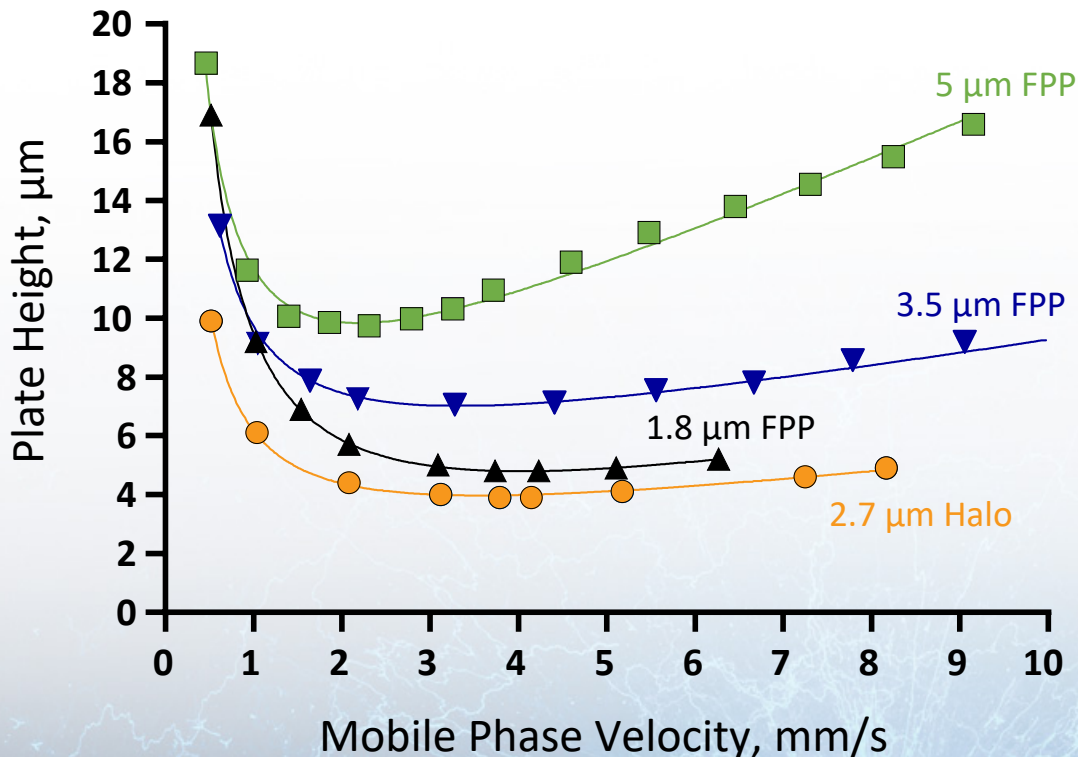
# How SPP Design Benefits HPLC Separation

## Effect of Particle Size and Type (small-pore)

Columns: 4.6 x 50 mm  
5  $\mu\text{m}$  FPP C18  
3.5  $\mu\text{m}$  FPP C18  
1.8  $\mu\text{m}$  FPP C18  
2.7  $\mu\text{m}$  HALO C18

Solute: naphthalene

Mobile phase: 60% ACN/40% water Temperature: 24 °C



## van Deemter Equation

$$H = A + \frac{B}{\mu} + C\mu$$

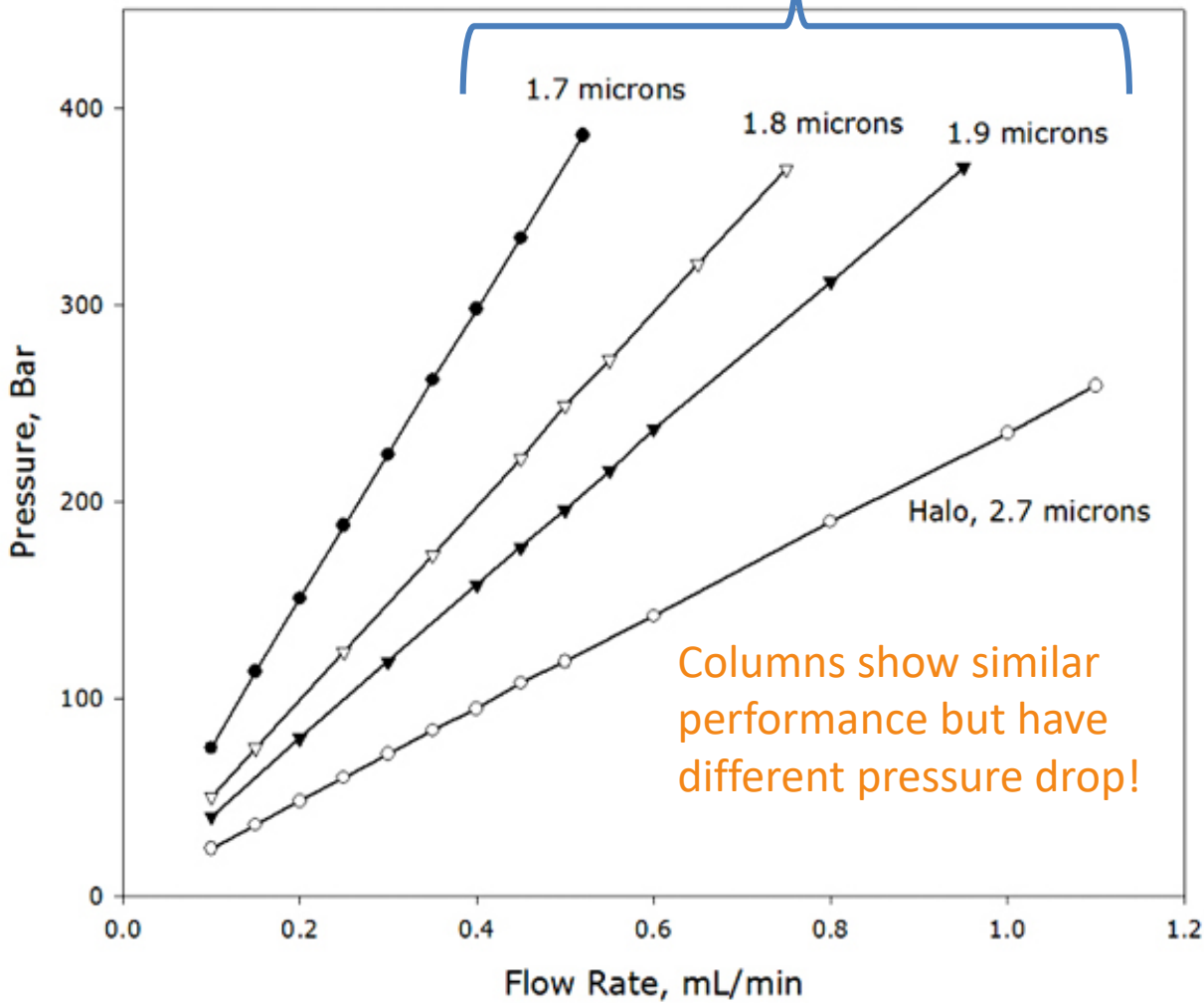
H = plate height  $H = L/N$

- A = eddy diffusion term  
(30 - 40% smaller vs FPP)
- B = longitudinal diffusion term  
(25 - 30% smaller vs FPP)
- C = resistance to mass transfer  
(smaller due to shorter flow path)

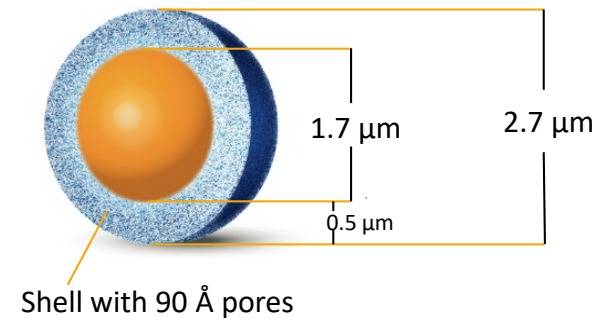
$\mu$  = mobile phase linear velocity ( $L/t_0$ )

# Low Backpressure of SPP vs FPP

Sub-2 $\mu\text{m}$  FPP



HALO<sup>®</sup> Fused-Core<sup>®</sup> Particle



Columns: 50 x 2.1 mm, C18  
Mobile Phase:  
70% ACN, 30% Water  
Temperature: 24 $^{\circ}$  C

# Summary of USP Modernization Efforts\*

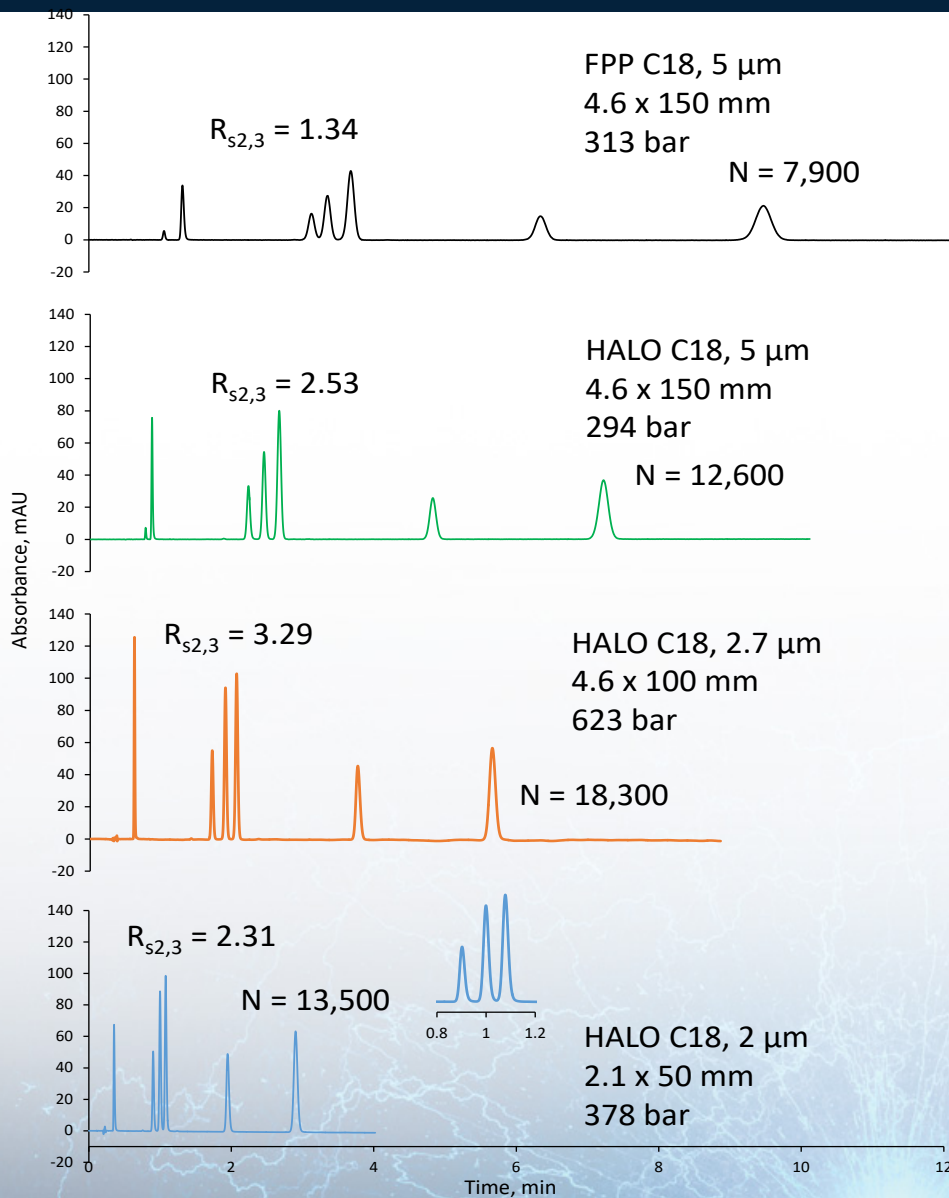
## (Following USP-NF Chapter 621 Guidelines)

- Particle size and/or column length may be changed if ratio of column length (L) to particle diameter ( $d_p$ ) is the same or in range between -25% to +50% of the prescribed  $L/d_p$  ratio.  $L/d_p$  is proportional to column resolving power.
- Bonded phase may not be changed to another L Code.
- Temperature may be adjusted  $\pm 10$  °C.
- Flow rate may be adjusted  $\pm 50\%$ .
- Mobile phase may be adjusted but cannot exceed  $\pm 10\%$  (or introduce new chemical modifiers).

\* USP moves to encourage adoption of modern HPLC columns and particles in USP Monograph Methods. Changes currently allowed only for isocratic methods; efforts are underway by USP to establish guidelines for changing gradient methods.



# L/d<sub>p</sub> Ratios When SPP is Same or Smaller



$L/d_p = 150/.005 = 30,000$   
For -25 to +50%, L/d<sub>p</sub> can be  
22,500-45,000

$L/d_p = 150/0.005 = 30,000$   
L/d<sub>p</sub> criteria met  
37% higher plates



$L/d_p = 100/.0027 = 37,037$   
L/d<sub>p</sub> criteria met  
57% higher plates



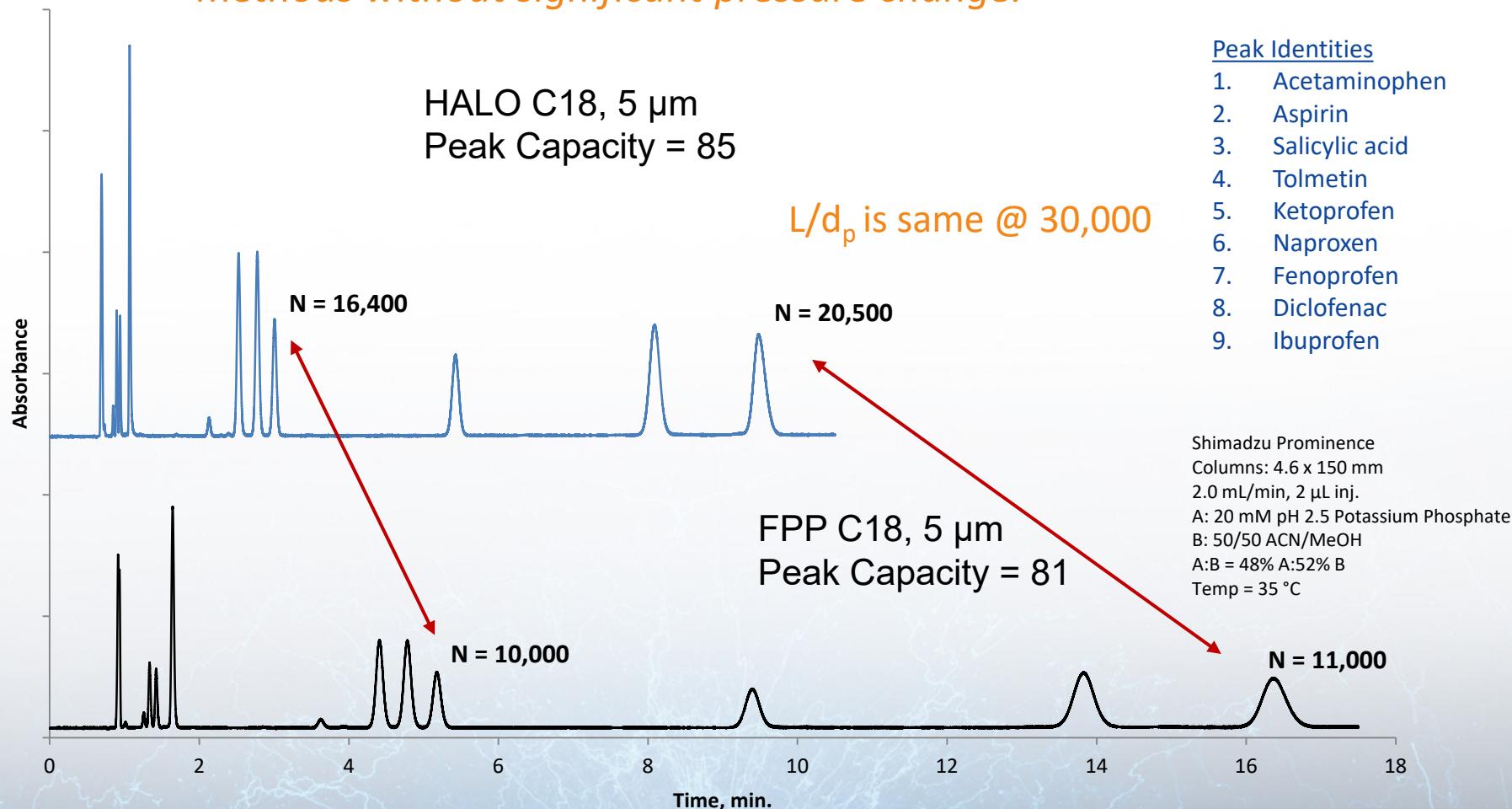
$L/d_p = 50/0.002 = 25,000$   
L/d<sub>p</sub> criteria met  
3x times faster  
41% higher plates





# When Should 5 $\mu\text{m}$ SPP Columns Be Used?

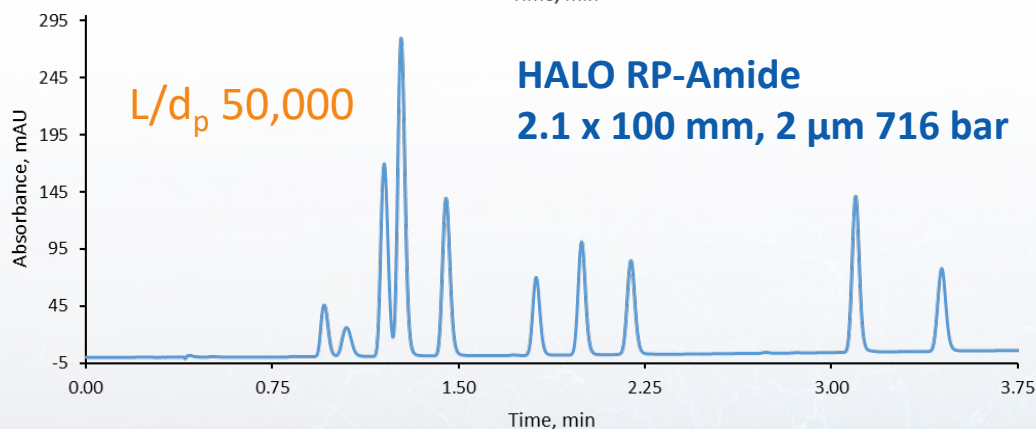
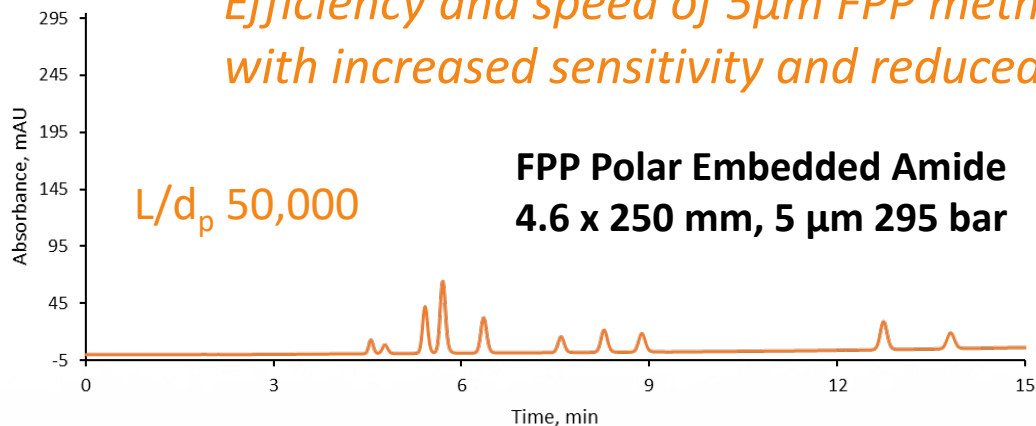
To increase efficiency, speed and response of 5 $\mu\text{m}$  FPP methods without significant pressure change.



60% more efficiency with shorter separation time

# What If UHPLC is Available?

*Efficiency and speed of 5 $\mu$ m FPP methods can be increased further with increased sensitivity and reduced solvent consumption.*

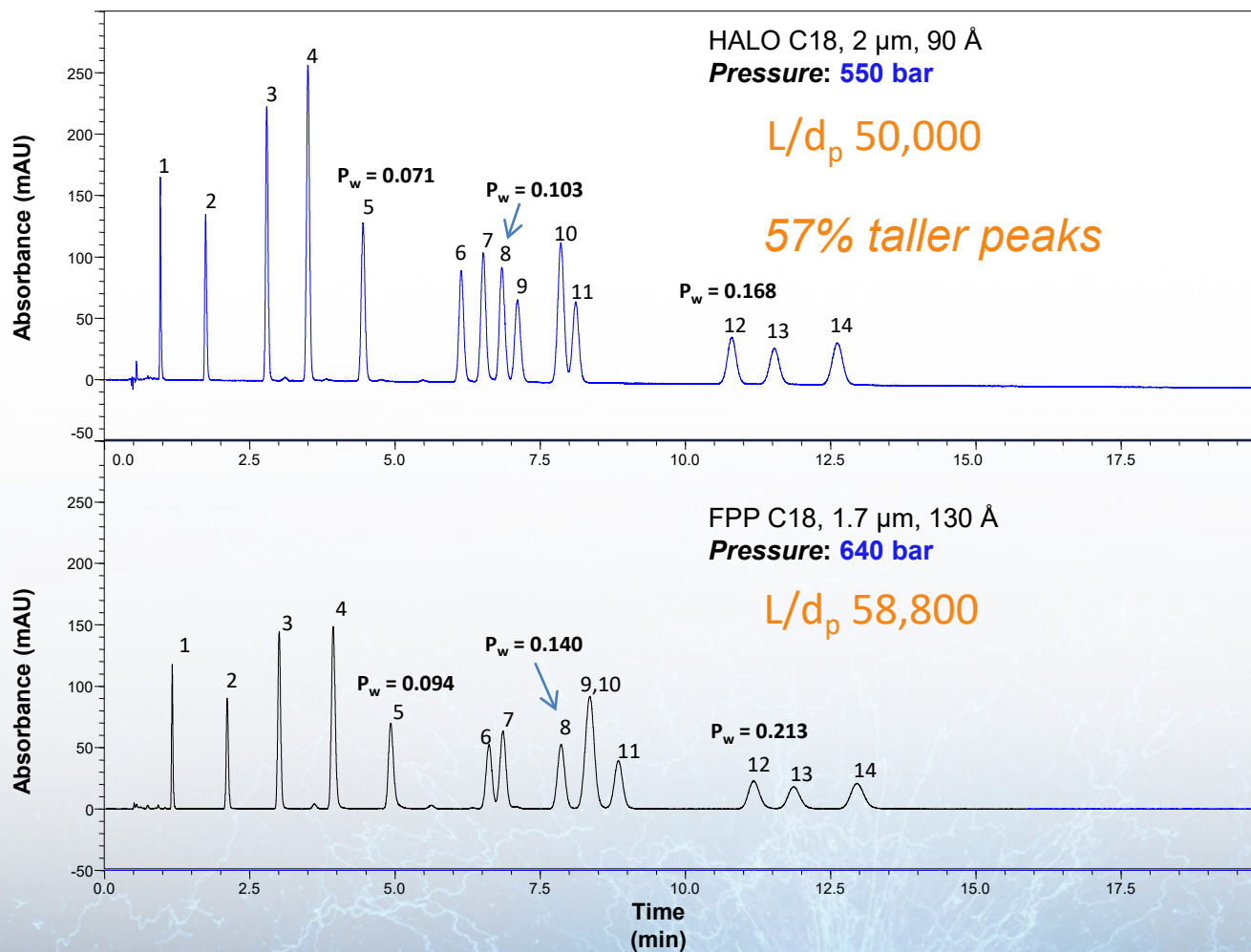


- Significant improvements in speed by moving from 5 $\mu$ m FPP to 2.7 $\mu$ m or even 2 $\mu$ m SPP.
- Note equivalent selectivity for polar embedded phase on SPP.
- **4x faster**
- **12x less solvent**
- **14.5x gain in sensitivity**

*“This particle lets you do “UHPLC-like” separations on a standard system or do ultrafast HPLC on a UHPLC system” -Customer Comment*

# When Should 2 $\mu\text{m}$ SPP Columns Be Used?

To decrease pressure and peak width of 1.7  $\mu\text{m}$  FPP methods



**Instrument:** Shimadzu Nexera  
**Column:** 2.1 x 100 mm  
**Mobile Phase:** Water/MeOH (72/28)  
**Flow rate:** 0.4 mL/min  
**Temperature:** 42 °C  
**Detection:** 254 nm  
**Injection volume:** 1.0  $\mu\text{L}$

#### Peak Identities

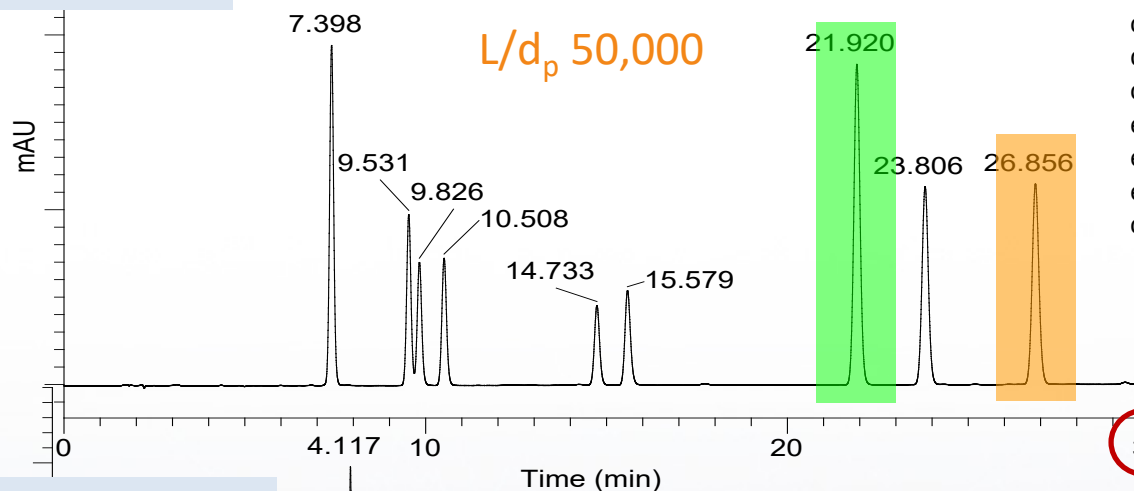
1. HMX
2. RDX
3. 1,3,5-Trinitrobenzene
4. 1,3-Dinitrobenzene
5. Nitrobenzene
6. Tetryl
7. 2,4,6-Trinitrotoluene
8. 2-Amino-4,6-Dinitrotoluene
9. 4-Amino-2,6-dinitrotoluene
10. 2,4-Dinitrotoluene
11. 2,6-Dinitrotoluene
12. 2-Nitrotoluene
13. 4-Nitrotoluene
14. 3-Nitrotoluene

Higher efficiency and sensitivity at lower pressure



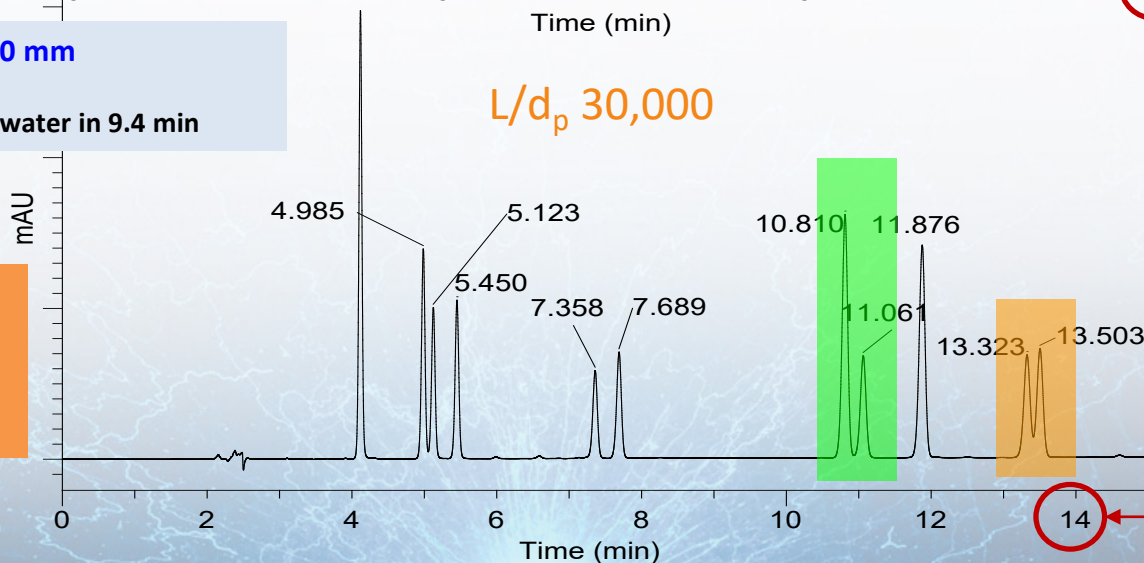
# Case Study: Gradient Steroids Separation from 5 $\mu\text{m}$ FPP to 5 $\mu\text{m}$ SPP

FPP C18, 5  $\mu\text{m}$ , 4.6 x 250 mm  
10  $\mu\text{L}$ , 1.5 mL/min, 20  $^{\circ}\text{C}$   
Gradient : 25–46%  $\text{CH}_3\text{CN}$ /water in 26.67 min



Analyte Elution order on HALO 5: (1) estriol, (2) prednisolone, (3) hydrocortisone, (4) cortisone, (5) dexamethasone, (6) corticosterone, (7) 17- $\beta$ -estradiol, (8) 17- $\alpha$ -estradiol, (9) estrone, (10) epi-testosterone, (11) cortisone acetate

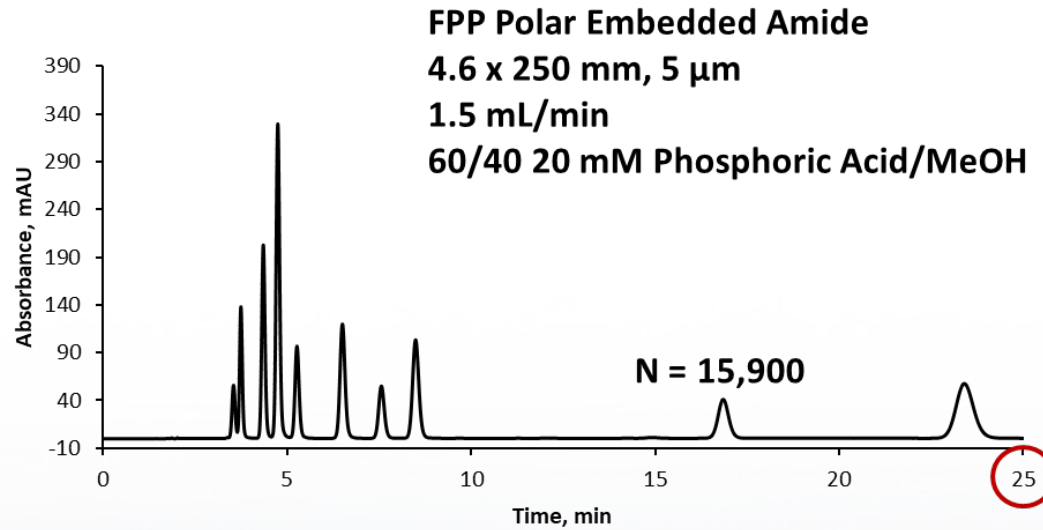
HALO 90  $\text{\AA}$  C18, 5  $\mu\text{m}$ , 3 x 150 mm  
2.3  $\mu\text{L}$ , 1.0 mL/min, 20  $^{\circ}\text{C}$   
Gradient from 25–46%  $\text{CH}_3\text{CN}$ /water in 9.4 min



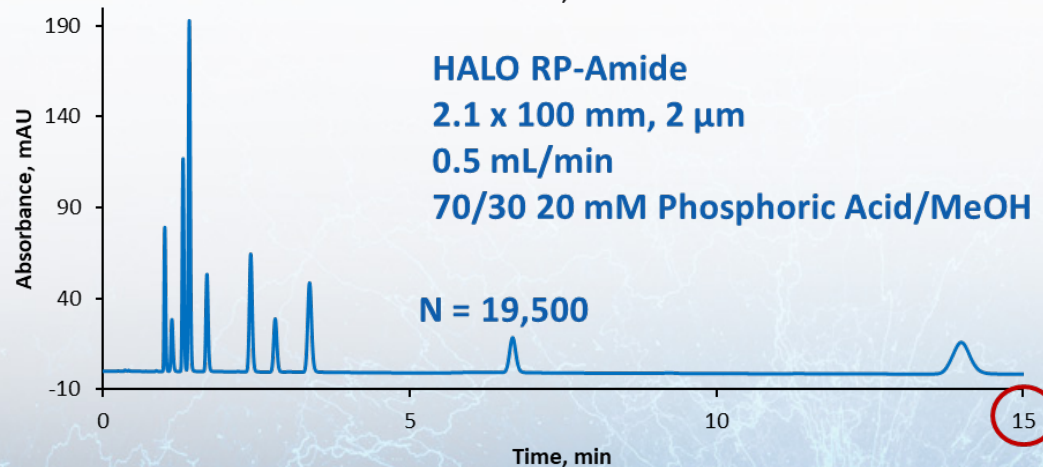
## SPP method

- 2x faster!
- 3x less solvent!

# Case Study: Isocratic Phenolic Acids Separation from 5 $\mu\text{m}$ FPP to 2 $\mu\text{m}$ SPP



$$L/d_p = 50,000$$

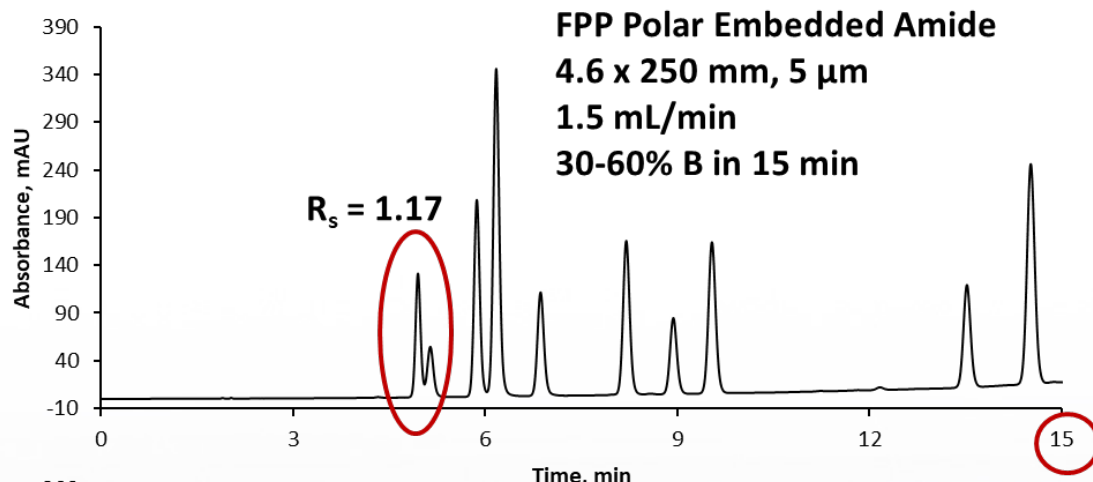


$$L/d_p = 50,000$$

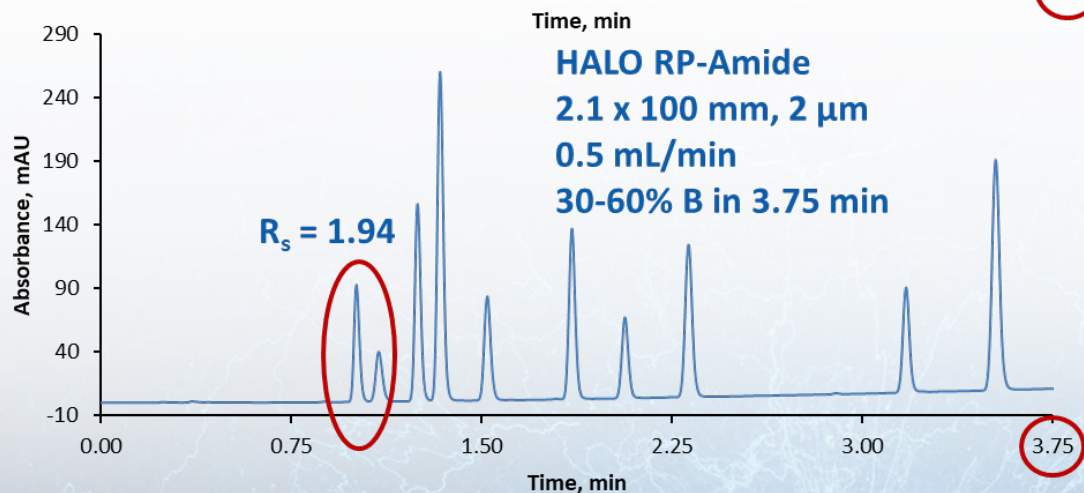
SPP method

- 1.7x faster
- 5x less solvent

# Case Study: Gradient Phenolic Acids Separation from 5 $\mu\text{m}$ FPP to 2 $\mu\text{m}$ SPP



$L/d_p = 50,000$  for  
both columns



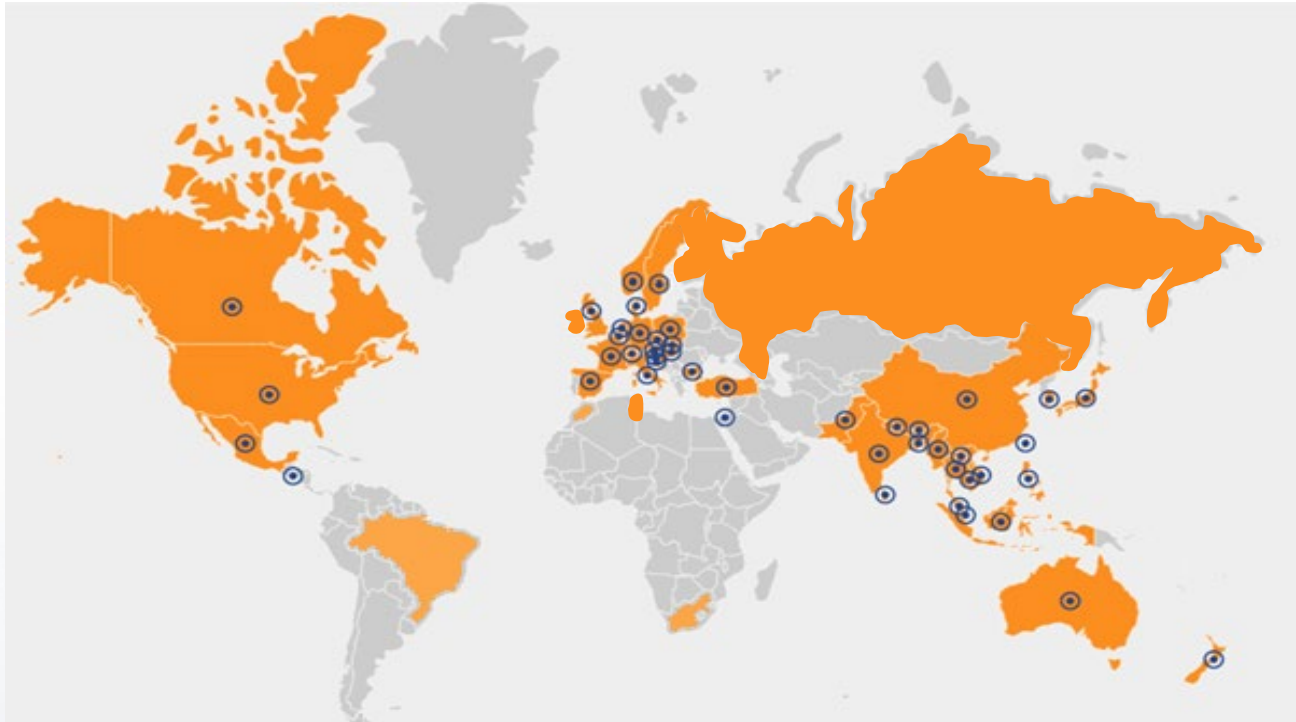
SPP method

- 4x faster
- 12x less solvent



# Summary

- Fused-Core<sup>®</sup> columns are designed for rugged, high-efficiency and high-speed separations.
  - UHPLC instruments that have been optimized for low dispersion are required to take full advantage of Fused-Core<sup>®</sup> for fast separations on short, small ID columns.
- Following new USP <621> guidelines for method modernization, many existing FPP methods can be quickly improved for speed and sensitivity using HALO<sup>®</sup> Fused-Core<sup>®</sup> column technology. Guidelines apply only to updating USP monographs.
- Examples and case studies were shown for FPP to SPP method transfer.



***HALO® is supplied through distributors in most major countries around the world and proudly holds a 99% on time (within 24 hours) shipping record!***

